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SÍNTESIS ESTEREOSELECTIVA DE cis-DECAHIDROQUINOLINAS: INTERMEDIOS AVANZADOS PARA EL ACCESO A LAS LEPADINAS
6. PUBLICACIONES

## 6.1

Model studies in the lepadin series: synthesis of enantiopure decahydroquinolines by aminocyclization of 2-(3aminoalkyl)cyclohexenones

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# Model studies in the lepadin series: synthesis of enantiopure decahydroquinolines by aminocyclization of 2-(3-aminoalkyl)cyclohexenones 

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#### Abstract

Syntheses of enantiopure 3-acetoxy-2-methyldecahydroquinolines are accomplished by coupling cyclohexenyllithium $\mathbf{3}$ with $\alpha$-amino epoxides and an aminocyclization of 2-(3-aminoalkyl)cyclohexenones (i.e., $\mathbf{5}$ and $\mathbf{9})$ as the key steps. The procedure allows the incorporation of alkyl substituents at $\mathrm{C}(5)$ to give enantiopure 2,3,5-trisubstitued decahydroquinolines. © 2005 Elsevier Ltd. All rights reserved.


## 1. Introduction

Lepadin alkaloids are structurally characterized by the presence of a 2,3,5-trisubstituted cis-fused decahydroquinoline ring. The substitution pattern, which has a methyl group at $\mathrm{C}(2)$, a hydroxyl group, free or acylated, at $\mathrm{C}(3)$, and an eight carbon side chain at $\mathrm{C}(5)$, shows a variety of stereochemical arrangements, as shown in Figure 1. Eight lepadins ( $\mathrm{A}-\mathrm{H}$ ) have been isolated from marine sources since 1991, ${ }^{1-4}$ of which lepadins A-C have been found to possess significant in vitro cytotoxicity against several human cancer cell lines, whereas lepadins D-F have shown low cytotoxicity but significant and selective antiplasmodial and antitrypanosomal activity.


Figure 1.
Total enantioselective syntheses of lepadins $\mathrm{A},{ }^{5} \mathrm{~B},{ }^{5-7} \mathrm{C},{ }^{5}$ D-E, ${ }^{7}$ and $H,{ }^{7}$ as well as a formal route to rac-lepadin $\mathrm{B}^{8}$ have been reported. The strategies described for the construction of 5 -substituted 3-hydroxy-2-

[^0]methyldecahydroquinolines in these synthetic approaches involve the elaboration of a polyfunctionalized piperidine followed by carbocyclic ring closure through aldol processes ${ }^{5,6}$ or the construction of the piperidine ring from cyclohexanone derivatives either by an intramolecular enamine alkylation ${ }^{7}$ or using a xanthate-mediated radical cyclization ${ }^{8}$ (Scheme 1).

In this work, we report our studies on a new synthetic entry to the azabicyclic core of lepadins, either those that show a cis or trans relationship between the respective methyl and hydroxyl substituents at $\mathrm{C}(2)$ and $\mathrm{C}(3)$ of the decahydroquinoline ring (see Fig. 1). In our approach, we envisaged enantiopure cyclohexenones of type $\mathbf{I}\left(\mathrm{R}^{\prime}=\mathrm{H}\right)$ as potential intermediates as they would bring about ring closure by forming the $\mathrm{N}-\mathrm{C}(8 \mathrm{a})$ bond. Here, we present the synthesis of these building blocks and the results obtained by their aminocyclization, either when $\mathrm{R}^{\prime}=\mathrm{H}$ or $\mathrm{R}^{\prime}=$ alkyl.

## 2. Results and discussion

### 2.1. Synthetic aspects

The required starting materials are 2-bromocyclohex-2enone ethylene acetal (1) and the $(S)$ and $(R)$ isomers of $[(S)$ -$1^{\prime}$-(dibenzylamino)ethyl]oxirane ( $\mathbf{2 a}$ and $\mathbf{2 b}$ ). The cyclohexenone derivative $\mathbf{1}$, reported by Smith, ${ }^{9}$ is a precursor of the $\alpha$-ketovinyl anion equivalent $\mathbf{3}$, often used in the formation of $\mathrm{C}-\mathrm{C}$ bonds, for example, in reactions with alkyl halides, ${ }^{9,10}$ ketones, ${ }^{11}$ ethyl chloroformate, ${ }^{9}$ and DMF. ${ }^{12}$ Moreover, this vinyllithium derivative has been


Scheme 1. Synthetic approaches to lepadin alkaloids.
transmetallated with copper, ${ }^{13}$ tin, ${ }^{14}$ and palladium ${ }^{15}$ reagents and then used in coupling processes. Finally, the lithium compound 3 reacts with $\mathrm{TMSCl}^{16}$ and sulfinates to give vinylsilane and vinylsulfoxide ${ }^{17}$ derivatives, respectively. To our knowledge, this versatile lithium derivative has not been used in reactions with epoxides, such as described in the present work. On the other hand, epoxides $\mathbf{2}^{18}$ have been described by Reetz, ${ }^{19}$ Barluenga and Concellón ${ }^{20}$ and Beaulieu, ${ }^{21}$ but there are no examples of their reactions with organolithium derivatives. ${ }^{22}$

The vinyllithium 3 formed on treatment of bromoacetal $\mathbf{1}$ with $n$ - BuLi in THF reacted with epoxide $2 \mathbf{a}^{20}$ in presence of $\mathrm{BF}_{3} \cdot \mathrm{Et}_{2} \mathrm{O}$ (Ganem's conditions) ${ }^{23,24}$ to give enantiopure alcohol 4a (Scheme 2). After protection of the hydroxyl group as an acetate and subsequent deprotection of the acetal, the resulting cyclohexenone 5a was submitted to a hydrogenation reaction, which involves a reduction of the double bond, a double debenzylation of the tertiary amine and an intramolecular reductive amination, to give the decahydroquinoline ring. In this process, in which two new


Scheme 2.
stereogenic centers are formed, the bicyclic compounds $\mathbf{6 a}$ and 7a were isolated in a $2: 1$ ratio. We then carried out the same sequence of reactions but starting from epoxide $\mathbf{2} \mathbf{b}^{20}$ (Scheme 3). In this series, the aminocyclization step starting from cyclohexenone 5b gave a nearly equimolecular mixture of decahydroquinolines $\mathbf{6 b}$ and $\mathbf{7 b}$. Thus, 5-dealkyllepadin derivatives with the same absolute configuration as lepadins D, E, and H (i.e., compound 6a), and lepadins A, B, and C (i.e., compound 6b) were achieved.



Scheme 3.
At this point, we explored the usefulness of cyclohexenones 5 as precursors of 5 -alkylsubstituted decahydroquinolines (Scheme 4). Treatment of 5a with $n$-BuLi gave a tertiary alcohol as an epimeric mixture, which was reacetylated upon the hydroxyl of the side chain, and the resulting $\mathbf{8 a}$ was oxidized $^{25}$ to give the rearranged enone $9 \mathbf{9}$. The multi-step tranformation of 9a under a hydrogen atmosphere (hydrogenation, debenzylation, and reductive aminocyclization) gave a mixture of trisubstituted decahydroquinolines 10a


Scheme 4.
and 11a in a nearly equimolecular ratio ( $71 \%$ overall yield), in which three new stereogenic centers were formed. Working with the epimeric epoxide $\mathbf{5 b}$, and following the same reaction sequence, decahydroquinolines $\mathbf{1 0 b}$ and 11b were formed in a $1: 4$ ratio ( $65 \%$ overall yield).

In all the cyclization processes $(\mathbf{5} \rightarrow \mathbf{6}+\mathbf{7}$ and $\mathbf{9} \rightarrow \mathbf{1 0}+\mathbf{1 1})$, both in series a ( $3 R$ configuration) and series $\mathbf{b}$ ( $3 S$ configuration), the isolated decahydroquinolines show an $R$ configuration at $\mathrm{C}(8 \mathrm{a})$ (see Fig. 2). The configuration at $C(4 a)$ is controlled by the configuration of $C(3)$ as well as by the presence or absence of a substituent at $\mathrm{C}(5)$. From the $\beta$-unsubstituted cyclohexenones (i.e., compounds 5), the aminocyclization takes place with some diastereoselection if the acetoxy substituent can adopt a pseudo-equatorial disposition in the transition state leading to the reduced product, as occurs in 6a, whereas in the epimeric series no stereocontrol was observed in the formation of the $C(4 a)$ stereocenter. Since it has not been established if the course of the reaction follows a pathway through an enimine intermediate or if there is a reduction of the double bond prior to the cyclization step, a clear understanding of the stereochemical course is not possible at this stage. More intriguing is the pathway of the aminocyclization leading to 2,3,5-trisubstituted decahydroquinolines 10 and 11. The configuration at $\mathrm{C}(4 \mathrm{a})$ and $\mathrm{C}(5)$ in all cases showed a trans
relationship between the hydrogen atoms of these stereocenters suggesting that the double bond underwent a trans hydrogenation, as has been reported in some tetrasubstituted alkenes, ${ }^{26}$ or, after a cis hydrogenation and formation of the subsequent imine, an epimerization took place at $\mathrm{C}(4 \mathrm{a})$ through an enamine intermediate. Again, as occured in the 5 -unsubstituted series, the ratio of trans decahydroquinolines (i.e., 11b) to the cis epimers was higher in compounds with a $3 S$ rather than $3 R$ configuration.

### 2.2. NMR studies of decahydroquinolines 6, 7, 10, and 11 (series a and b)

The cis ( $\mathbf{6}$ and 10) and trans decahydroquinolines (7 and 11) are clearly differentiated by two NMR features: (i) the ${ }^{1} \mathrm{H}$ NMR chemical shift of $\mathrm{H}-8 \mathrm{a}$, which appears more deshieled ( $\delta 2.95$ ) in the cis-than in the trans-derivatives ( $\delta 2.20$ ); (ii) the ${ }^{13} \mathrm{C}$ chemical shift of $\mathrm{C}(7)$ is more deshielded $(\sim 4-$ 5 ppm ) in the trans than in the cis derivatives. ${ }^{27}$ In all cases, the preferred conformation of the cis decahydroquinolines has the $\mathrm{H}-8 \mathrm{a}$ axial with respect to the N -containing ring ( N endo conformer).

The absolute configuration of $\mathbf{6 a}$ was deduced considering that: (a) the coupling constants for $\mathrm{H}-2(\mathrm{dq}, J=10,6.5 \mathrm{~Hz})$ and H-3 (td, $J=10.5,4.8 \mathrm{~Hz}$ ) determined their axial location and hence, fixed the methyl at $\mathrm{C}(2)$ and the acetoxy at $\mathrm{C}(3)$ to an equatorial disposition; (b) the multiplicity of $\mathrm{H}-8 \mathrm{a}$ ( br s) implied an equatorial relationship with respect to the cyclohexane ring, which discarded not only a trans junction of the decaline ring but also, taking into account the preferred conformation, implied an $R$ configuration for $\mathrm{C}(8 \mathrm{a})$. The ${ }^{13} \mathrm{C}$ chemical shifts also agree with this elucidation since the value of $\delta 20.3$ for $\mathrm{C}(7)$ is diagnostic of a cis decahydroquinoline in a N -endo conformation. For trans compound $7 \mathbf{7 a}$, the axial proton $\mathrm{H}-8 \mathrm{a}$ is strongly coupled to two adjacent axial protons and one equatorial proton. Hence, its resonance signal appears as a deceptively simple triplet $(J=10.4 \mathrm{~Hz})$ of doublets $(J=3.2 \mathrm{~Hz})$ centered at $\delta 2.19$. The NMR data for compounds $\mathbf{6 b}$ and 7b follow the same pattern of signals as that of their corresponding epimers at $\mathrm{C}(3)$, the major differences being in the chemical shift for $\mathrm{H}-3$, which is now more deshielded since it is located in an equatorial arrangement, and in C-3 and C-4a, which resonate at a lower field, due to the axially

$6 a$


7a


6b


7b


10a


11a


11b

Figure 2. Preferred conformation of decahydroquinolines 6, 7, 10, and 11.
Table 1. ${ }^{13} \mathrm{C}$ NMR data for decahydroquinolines 6, 7, 10, and 11

|  | C-2 | C-3 | C-4 | C-4a | C-5 | C-6 | C-7 | C-8 | C-8a | C-9 | $\mathrm{C}-1^{\prime}$ | C-2' | C-3' | C-4' | OAc |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 6 a | 56.5 | 73.0 | 36.4 | 36.8 | 26.5 | 26.1 | 20.3 | 31.9 | 54.3 | 18.9 | - | - | - | - | 170.6/21.3 |
| 6b | 55.0 | 70.3 | 35.0 | 33.5 | 27.5 | 26.8 | 20.8 | 32.6 | 54.6 | 18.2 | - | - | - | - | 170.8/21.4 |
| 7a | 55.7 | 75.7 | 37.5 | 41.6 | 31.8 | 25.6 | 25.3 | 32.6 | 60.9 | 18.5 | - | - | - | - | 170.4/21.2 |
| 7b | 54.0 | 71.6 | 36.6 | 36.9 | 31.9 | 26.0 | 25.5 | 33.2 | 61.4 | 18.3 | - | - | - | - | 171.0/21.3 |
| 10a | 56.5 | 72.7 | 32.5 | 41.8 | 33.0 | 29.7 | 21.3 | 32.7 | 55.5 | 18.8 | 31.9 | 28.1 | 23.1 | 14.1 | 170.5/21.1 |
| 11a | 55.4 | 76.4 | 34.4 | 46.3 | 40.8 | 31.3 | 24.7 | 33.1 | 60.8 | 18.7 | 32.1 | 28.5 | 23.1 | 14.1 | 170.5/21.3 |
| 11b | 53.7 | 71.6 | 33.5 | 40.9 | 40.7 | 31.9 | 24.9 | 33.4 | 61.1 | 18.2 | 31.7 | 28.4 | 23.1 | 14.1 | 171.0/21.3 |

All spectra were recorded at 100 MHz in $\mathrm{CDCl}_{3}$ and the assignments were aided by HSQC experiments.
located acetoxy group (Table 1). For trisubstituted cis decahydroquinoline 10a, the butyl substituent at $\mathrm{C}(5)$ controls the preferred conformation of the bicyclic ring, which agrees with the conformation showed for lepadins where the substituent at $C(5)$ is always equatorially located. The stereochemistry at $C(5)$ for the butyl substituted products ( $\mathbf{1 0}$ and 11) was determined considering that the equatorially located butyl side chain exerts a steric crowding on $\mathrm{H}-4 e q$, due to their 1,3-synperiplanar relationship, which is reflected in the ${ }^{13} \mathrm{C}$ and ${ }^{1} \mathrm{H}$ NMR spectra by an upfield chemical shift ( $\sim 3 \mathrm{ppm}$ ) for $\mathrm{C}(4)$ and a downfield chemical shift ( $\delta 2.25 \pm 0.05$ ) for H-4eq as compared to the NMR data for compounds 6 and 7.

In summary, a new synthetic entry to enantiopure polysubstituted decahydroquinolines has been reported. Although the observed stereoselectivity does not allow lepadin-type stereochemistries to be achieved, further studies in aminocyclization processes, starting from cyclohexenones of type 5, are in progress with the aim of achieving the required stereochemistry of lepadin derivatives. Interestingly, the reported methodology could be applied to the synthesis of another type of natural decahydroquinolines, such as trans-195A, ${ }^{28}$ 5-epi-trans$243 \mathrm{~A},{ }^{29}$ and related alkaloids isolated from dendrobatid frogs, ${ }^{27 \mathrm{c}}$ which show the same pattern of relative configuration as compounds 11a and 11b in their four stereocenters.

## 3. Experimental

### 3.1. General

All reactions were carried out under an argon atmosphere with dry, freshly distilled solvents under anhydrous conditions. Analytical TLC was performed on $\mathrm{SiO}_{2}$ (silica gel $60 \mathrm{~F}_{254}$, Merck) or $\mathrm{Al}_{2} \mathrm{O}_{3}$ (ALOX N/UV 254 , Polygram), and the spots were located with iodoplatinate reagent (compounds 4, 5, 8, and 9) or $1 \%$ aqueous $\mathrm{KMnO}_{4}$ (compounds 6, 7, 10, and 11). Chromatography refers to flash chromatography and was carried out on $\mathrm{SiO}_{2}$ (silica gel 60, SDS, 230-240 mesh ASTM) or $\mathrm{Al}_{2} \mathrm{O}_{3}$ (aluminium oxide 90 , Merck). Drying of organic extracts during workup of reactions was performed over anhydrous $\mathrm{Na}_{2} \mathrm{SO}_{4}$. Optical rotations were recorded with a Perkin-Elmer 241 polarimeter. ${ }^{1} \mathrm{H}$ and ${ }^{13} \mathrm{C}$ NMR spectra were recorded with a Varian Gemini 200 or 300 , or a Varian Mercury 400 instrument. Chemical shifts are reported in ppm downfield ( $\delta$ ) from $\mathrm{Me}_{4} \mathrm{Si}$. All new compounds were determined to be $>95 \%$ pure by ${ }^{1} \mathrm{H}$ NMR spectroscopy.
3.1.1. 2-[(2R,3S)-3-Dibenzylamino-2-hydroxybutyl] cyclohex-2-enone ethylene acetal (4a). A solution of 6-bromo-1,4-dioxaspiro[4.5]dec-6-ene (1, $\quad 1.04 \mathrm{~g}$, 4.75 mmol ) in THF ( 3 mL ) was added to a solution of $n-\mathrm{BuLi}(1.6 \mathrm{M}$ in hexanes, 3.2 mL , 5.11 mmol ) in THF $(7 \mathrm{~mL})$ at $-78{ }^{\circ} \mathrm{C}$. The reaction mixture was stirred for 90 min , treated with a solution of $(2 S)$-[1' $(S)$-(dibenzylamino) ethyl]oxirane ( $\mathbf{2 a}, 489 \mathrm{mg}, 1.83 \mathrm{mmol}$ ) in THF $(6 \mathrm{~mL})$ and $\mathrm{BF}_{3} \cdot \mathrm{Et}_{2} \mathrm{O}(0.64 \mathrm{~mL}, 5.11 \mathrm{mmol})$, and continuously stirred at $-78{ }^{\circ} \mathrm{C}$ for 2 h prior to being quenched with saturated $\mathrm{NaHCO}_{3}$ solution $(10 \mathrm{~mL})$ and warmed to rt . The
product was extracted with $\mathrm{Et}_{2} \mathrm{O}(3 \times 20 \mathrm{~mL})$, the combined organic layers were dried, concentrated, and the residue was chromatographed $\left(\mathrm{SiO}_{2}\right.$, elution with $9: 1$ hexane/EtOAc) to give $560 \mathrm{mg}(75 \%)$ of 4 a as a colorless oil: $R_{\mathrm{f}}=0.31\left(\mathrm{SiO}_{2}\right.$, 8:2 hexane/EtOAc); $[\alpha]_{\mathrm{D}}^{20}+10.0\left(c 1.2\right.$ in $\left.\mathrm{CHCl}_{3}\right) ;{ }^{1} \mathrm{H}$ NMR ( $300 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $1.15(\mathrm{~d}, J=6.3 \mathrm{~Hz}, 3 \mathrm{H}), 1.60-1.80(\mathrm{~m}$, $5 \mathrm{H}), 1.95-2.05(\mathrm{br}, 2 \mathrm{H}), 2.52-2.61(\mathrm{~m}, 1 \mathrm{H}), 2.85(\mathrm{dm}, J=$ $14 \mathrm{~Hz}, 1 \mathrm{H}$ ), 3.25 (br, 1H), 3.45 (d, $J=13.8 \mathrm{~Hz}, 2 \mathrm{H}$ ), 3.66$3.73(\mathrm{~m}, 1 \mathrm{H}), 3.77(\mathrm{~d}, J=13.8 \mathrm{~Hz}, 2 \mathrm{H}), 3.84-3.90(\mathrm{~m}, 2 \mathrm{H})$, 3.91-3.97 (m, 2H), 5.76 (t, $J=3 \mathrm{~Hz}, 1 \mathrm{H}$ ), 7.18-7.40 (m, $10 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR $\left(50 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 8.4\left(\mathrm{CH}_{3}\right), 20.6\left(\mathrm{CH}_{2}\right)$, $25.4\left(\mathrm{CH}_{2}\right), 33.2\left(\mathrm{CH}_{2}\right), 36.2\left(\mathrm{CH}_{2}\right), 54.3\left(\mathrm{CH}_{2}\right), 57.7(\mathrm{CH})$, $64.6\left(\mathrm{CH}_{2}\right), 64.7\left(\mathrm{CH}_{2}\right), 74.1(\mathrm{CH}), 107.5(\mathrm{C}), 126.7(\mathrm{CH})$, $128.1(\mathrm{CH}), 128.7(\mathrm{CH}), 133.6(\mathrm{CH}), 140.3(\mathrm{C})$. HRFABMS calcd for $\mathrm{C}_{26} \mathrm{H}_{34} \mathrm{NO}_{3}\left(\mathrm{M}^{+}+1\right) 408.2539$, found 408.2516 .
3.1.2. 2-[(2S,3S)-3-Dibenzylamino-2-hydroxybutyl] cyclohex-2-enone ethylene acetal (4b). Operating as above, starting from 881 mg ( 4.02 mmol ) of $\mathbf{1}$ and using $(2 R)$ - $\left[1^{\prime}(S)\right.$-(dibenzylamino)ethyl]oxirane ( $\mathbf{2 b}, 414 \mathrm{mg}$, $1.55 \mathrm{mmol})$, and after chromatography ( $\mathrm{SiO}_{2}, 9: 1$ hexane/ EtOAc) 4b ( $518 \mathrm{mg}, 82 \%$ ) was isolated as an oil: $R_{\mathrm{f}}=0.28$ ( $\mathrm{SiO}_{2}, 9: 1$ hexane/EtOAc); ${ }^{1} \mathrm{H}$ NMR ( $200 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $1.05(\mathrm{~d}, J=6.6 \mathrm{~Hz}, 3 \mathrm{H}), 1.62-1.72(\mathrm{~m}, 5 \mathrm{H}), 1.95-2.05(\mathrm{br}$, 2H), 2.17-2.25 (m, 1H), 2.52-2.62 (m, 1H), 3.33 (d, J= $13.6 \mathrm{~Hz}, 2 \mathrm{H}), 3.64-3.76(\mathrm{~m}, 1 \mathrm{H}), 3.88(\mathrm{~d}, J=13.6 \mathrm{~Hz}, 2 \mathrm{H})$, $3.90-3.98(\mathrm{~m}, 4 \mathrm{H}), 5.93(\mathrm{t}, J=2 \mathrm{~Hz}, 1 \mathrm{H}), 7.16-7.40(\mathrm{~m}$, $10 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $\left.50 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 8.6\left(\mathrm{CH}_{3}\right), 20.6\left(\mathrm{CH}_{2}\right)$, $25.3\left(\mathrm{CH}_{2}\right)$, $33.6\left(\mathrm{CH}_{2}\right), 34.3\left(\mathrm{CH}_{2}\right), 53.6\left(\mathrm{CH}_{2}\right), 58.2(\mathrm{CH})$, $64.7\left(\mathrm{CH}_{2}\right), 64.8\left(\mathrm{CH}_{2}\right), 70.8(\mathrm{CH}), 107.8(\mathrm{C}), 127.0(\mathrm{CH})$, $128.3(\mathrm{CH}), 128.9(\mathrm{CH}), 131.5(\mathrm{CH}), 139.2(\mathrm{C})$.
3.1.3. 2-[(2R,3S)-2-Acetoxy-3-(dibenzylamino)butyl] cyclohex-2-enone (5a). To a solution of $\mathbf{4 a}(101 \mathrm{mg}$, $0.25 \mathrm{mmol})$ in pyridine ( 0.8 mL ) and $\mathrm{Ac}_{2} \mathrm{O}(0.24 \mathrm{~mL}$, 2.5 mmol ) was added DMAP ( $5 \mathrm{mg}, 0.04 \mathrm{mmol}$ ). The reaction mixture was stirred overnight at rt . A saturated $\mathrm{NaHCO}_{3}$ solution ( 15 mL ) was added and the mixture was extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}(3 \times 10 \mathrm{~mL})$. The dried organic extracts were concentrated to give the corresponding acetate, which was used directly in the following acetal hydrolysis step. The above crude acetal was dissolved in $1: 1$ $\mathrm{H}_{2} \mathrm{O} / \mathrm{THF}(4 \mathrm{~mL})$ and stirred at rt for 1 h . The reaction mixture was basified with saturated aqueous $\mathrm{NaHCO}_{3}$ ( 15 mL ) and extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}(3 \times 10 \mathrm{~mL})$, and the resulting organic extracts were dried and concentrated. The residue was purified by chromatography $\left(\mathrm{SiO}_{2}\right.$, hexane/ EtOAc 8:2) to give 5a as an oil ( $80 \mathrm{mg}, 83 \%$ ): $R_{\mathrm{f}}=0.27$ $\left(\mathrm{SiO}_{2}, 8: 2\right.$ hexane/EtOAc $) ;[\alpha]_{\mathrm{D}}^{20}+7.8\left(c 0.7\right.$ in $\left.\mathrm{CHCl}_{3}\right) ;{ }^{1} \mathrm{H}$ NMR ( $300 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) 1.08 (d, $J=6.6 \mathrm{~Hz}, 3 \mathrm{H}$ ), $1.75-$ $1.83(\mathrm{~m}, 1 \mathrm{H}), 1.83-1.93(\mathrm{~m}, 2 \mathrm{H}), 1.94(\mathrm{~s}, 3 \mathrm{H}), 2.16-2.28$ $(\mathrm{m}, 2 \mathrm{H}), 2.31-2.38(\mathrm{~m}, 2 \mathrm{H}), 2.75$ (quint, $J=6 \mathrm{~Hz}, 1 \mathrm{H}$ ), 3.18 (dm, $J=14 \mathrm{~Hz}, 1 \mathrm{H}), 3.41$ (d, $J=13.6 \mathrm{~Hz}, 2 \mathrm{H}), 3.78$ (d, $J=$ $13.6 \mathrm{~Hz}, 2 \mathrm{H}$ ), 5.17 (ddd, $J=9.6,7.6,3.4 \mathrm{~Hz}, 1 \mathrm{H}$ ), 6.53 (t, $J=4 \mathrm{~Hz}, 1 \mathrm{H}), 7.17-7.41(\mathrm{~m}, 10 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( 50 MHz , $\left.\mathrm{CDCl}_{3}\right) 8.8\left(\mathrm{CH}_{3}\right)$, $21.2\left(\mathrm{CH}_{3}\right), 22.9\left(\mathrm{CH}_{2}\right), 26.2\left(\mathrm{CH}_{2}\right), 33.2$ $\left(\mathrm{CH}_{2}\right), 38.2\left(\mathrm{CH}_{2}\right), 53.9\left(\mathrm{CH}_{2}\right), 55.1(\mathrm{CH}), 74.3(\mathrm{CH}), 126.7$ $(\mathrm{CH}), 128.1(\mathrm{CH}), 129.0(\mathrm{CH}), 136.4(\mathrm{C}), 139.9(\mathrm{C}), 146.4$ $(\mathrm{CH}), 170.4(\mathrm{C}), 198.7$ (C). Anal. Calcd for $\mathrm{C}_{26} \mathrm{H}_{31} \mathrm{NO}_{3}$ : C, 77.00; H, 7.70; N, 3.45. Found C, 76.75; H, 7.85; N, 3.39.
3.1.4. 2-[(2S,3S)-2-Acetoxy-3-(dibenzylamino)butyl]
cyclohex-2-enone ethylene acetal (5b). Operating as cyclohex-2-enone ethylene acetal (5b). Operating as
above, starting from 263 mg ( 0.64 mmol ) of alcohol $\mathbf{4 b}$, and after chromatography $\left(\mathrm{SiO}_{2}, 8: 2\right.$ hexane/EtOAc), acetate $\mathbf{5 b}(196 \mathrm{mg}, 81 \%)$ was isolated as an oil: $R_{\mathrm{f}}=0.25$ $\left(\mathrm{SiO}_{2}\right.$, hexane/EtOAc, 8:2); $[\alpha]_{\mathrm{D}}^{20}-32\left(c 1.8\right.$ in $\left.\mathrm{CHCl}_{3}\right) ;{ }^{1} \mathrm{H}$ NMR ( $300 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) 1.09 (d, $J=7.2 \mathrm{~Hz}, 3 \mathrm{H}$ ), $1.86-$ $1.94(\mathrm{~m}, 3 \mathrm{H}), 2.01(\mathrm{~s}, 3 \mathrm{H}), 2.16-2.30(\mathrm{~m}, 3 \mathrm{H}), 2.33-2.40$ $(\mathrm{m}, 1 \mathrm{H}), 2.60(\mathrm{dm}, J=14 \mathrm{~Hz}, 1 \mathrm{H}), 2.89$ (quint, $J=7 \mathrm{~Hz}$, $1 \mathrm{H}), 3.37$ (d, $J=13.8 \mathrm{~Hz}, 2 \mathrm{H}), 3.87(\mathrm{~d}, J=13.8 \mathrm{~Hz}, 2 \mathrm{H})$, 5.06 (ddd, $J=10.2,6.2,2.6 \mathrm{~Hz}, 1 \mathrm{H}), 6.60(\mathrm{t}, J=4.2 \mathrm{~Hz}$, $1 \mathrm{H}), 7.18-7.40(\mathrm{~m}, 10 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $50 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ): 9.7 $\left(\mathrm{CH}_{3}\right), 21.2\left(\mathrm{CH}_{3}\right), 22.9\left(\mathrm{CH}_{2}\right), 26.1\left(\mathrm{CH}_{2}\right), 33.0\left(\mathrm{CH}_{2}\right), 38.2$ $\left(\mathrm{CH}_{2}\right), 54.3\left(\mathrm{CH}_{2}\right), 55.4(\mathrm{CH}), 74.7(\mathrm{CH}), 126.6(\mathrm{CH}), 128.1$ $(\mathrm{CH}), 128.8(\mathrm{CH}), 136.3(\mathrm{C}), 140.3(\mathrm{C}), 146.3(\mathrm{CH}), 170.3$ (C), 198.8 (C). HRFABMS calcd for $\mathrm{C}_{26} \mathrm{H}_{32} \mathrm{NO}_{3}\left(\mathrm{M}^{+}+1\right)$ 406.2382, found 406.2339.
3.1.5. Aminocyclization of $\mathbf{5 a}$. A suspension of enone $\mathbf{5 a}$ $(50 \mathrm{mg}, 0.12 \mathrm{mmol})$ and activated ${ }^{30} \mathrm{Pd}(\mathrm{OH})_{2}$ in EtOH ( 2 mL ) was stirred overnight under hydrogen. The catalyst was removed by filtration through Celite, and the solvent was evaporated to give a residue, which was purified by chromatography $\left(\mathrm{Al}_{2} \mathrm{O}_{3}, 9: 1\right.$ hexane/EtOAc) to give $\mathbf{6 a}$ ( $13 \mathrm{mg}, 54 \%$ ) and $7 \mathrm{a}(9 \mathrm{mg}, 36 \%)$, both as oils.
( $2 S, 3 R, 4 \mathrm{a} R, 8 \mathrm{a} R$ )-3-Acetoxy-2-methyldecahydroquinoline (6a). $R_{\mathrm{f}}=0.51\left(\mathrm{Al}_{2} \mathrm{O}_{3}, 8: 2\right.$ hexane/EtOAc); $[\alpha]_{\mathrm{D}}^{20}-20.7(c$ 1.3 in $\mathrm{CHCl}_{3}$ ); ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}, \mathrm{COSY}$ ) 1.08 (d, $J=6.4 \mathrm{~Hz}, 3 \mathrm{H}, \mathrm{Me}$ ), 1.20 (m, H-5ax), 1.40 (m, 3H, H-6ax and H-7), 1.45 (m, H-4ax), 1.55 (m, H-8), 1.70 (m, H-8), 1.74 (m, 3H, H-4a, H-5eq, H-6eq), 1.87 (ddd, $J=11.0,3.6$, $1.2 \mathrm{~Hz}, \mathrm{H}-4 e q$ ), 2.05 (s, $3 \mathrm{H}, \mathrm{OAc}$ ), 2.70 (dq, $J=10.0$, $6.5 \mathrm{~Hz}, \mathrm{H}-2 a x$ ), 2.95 (br s, H-8a), $4.58(\mathrm{td}, J=10.5,4.8 \mathrm{~Hz}$, $\mathrm{H}-3 a x) ;{ }^{13} \mathrm{C}$ NMR see Table 1. HRFABMS calcd for $\mathrm{C}_{12} \mathrm{H}_{22} \mathrm{NO}_{2}\left(\mathrm{M}^{+}+1\right)$ 212.1651, found 212.1646.
(2S,3R,4aS,8aR)-3-Acetoxy-2-methyldecahydroquinoline (7a). ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$, COSY) 1.02 (qd, $J=$ $10.4,3.2 \mathrm{~Hz}, \mathrm{H}-5 a x$ ), 1.10 (masked, H-4ax), 1.12 (d, $J=$ $6.4 \mathrm{~Hz}, 3 \mathrm{H}, \mathrm{Me}), 1.20-1.30(\mathrm{~m}, 2 \mathrm{H}, \mathrm{H}-8 \mathrm{ax}$ and $\mathrm{H}-4 \mathrm{a}), 1.35$ (m, 2H, H-6ax and H-7ax), 1.65 (m, 2H, H-7eq and H-5eq), 1.8 (m, 2H, H-8eq and H-6eq), 2.04 (s, 3H, OAc), 2.05 (masked, H-4eq), 2.19 (td, $J=10.4,3.2 \mathrm{~Hz}, \mathrm{H}-8 \mathrm{a}), 2.76$ (dq, $J=10,6.4 \mathrm{~Hz}, \mathrm{H}-2 a x), 4.45$ (td, $J=10.4,4.4 \mathrm{~Hz}, \mathrm{H}-3 a x)$; ${ }^{13} \mathrm{C}$ NMR see Table 1 .
3.1.6. Aminocyclization of $\mathbf{5 b}$. Operating as above, starting from $49 \mathrm{mg}(0.12 \mathrm{mmol})$ of enone $\mathbf{5 b}$, and after chromatography $\left(\mathrm{Al}_{2} \mathrm{O}_{3}\right.$, from $9: 1$ to $7: 3$ hexane/EtOAc), 11 mg ( $43 \%$ ) of $\mathbf{6 b}$ and $11 \mathrm{mg}(43 \%)$ of $\mathbf{7 b}$, both as colorless oils, were isolated.
( $2 S, 3 S, 4 \mathrm{a} R, 8 \mathrm{a} R$ )-3-Acetoxy-2-methyldecahydroquinoline (6b). $R_{\mathrm{f}}=0.30\left(\mathrm{Al}_{2} \mathrm{O}_{3}, 8: 2\right.$ hexane/EtOAc); $[\alpha]_{\mathrm{D}}^{20}+10.8(c$ 0.8 in $\mathrm{CHCl}_{3}$ ); ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}, \mathrm{COSY}$ ) 1.08 (d, $J=6.6 \mathrm{~Hz}, 3 \mathrm{H}, \mathrm{Me}$ ), 1.20 (m, H-5ax), 1.40 (m, 3H, H-6ax, H-7), 1.50 (m, 2H, H-8ax, H-4ax), 1.72 (m, 4H, H-4a, $\mathrm{H}-5 e q, \mathrm{H}-6 e q, \mathrm{H}-8 e q$ ), 1.87 (ddd, $J=12,3.6,1.5 \mathrm{~Hz}$, H-4eq), 2.09 (s, 3H, OAc), 2.90 (qd, $J=6.8,2 \mathrm{~Hz}, \mathrm{H}-2 a x$ ), 2.92 (br, H-8a), 4.75 (ddd, $J=3.2,3.2,1.6 \mathrm{~Hz}, \mathrm{H}-3 e q)$; ${ }^{13} \mathrm{C}$ NMR see Table 1. HRFABMS calcd for $\mathrm{C}_{12} \mathrm{H}_{22} \mathrm{NO}_{2}\left(\mathrm{M}^{+}+\right.$ 1) 212.1651 , found 212.1648 .
( $2 S, 3 S, 4 \mathrm{a} S, 8 \mathrm{a} R$ )-3-Acetoxy-2-methyldecahydroquinoline
(7b). $R_{\mathrm{f}}=0.23\left(\mathrm{Al}_{2} \mathrm{O}_{3}, 8: 2\right.$ hexane/EtOAc $) ;[\alpha]_{\mathrm{D}}^{20}+28.6(c$ 0.8 in $\mathrm{CHCl}_{3}$ ); ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}, \mathrm{COSY}$ ) 0.94 (qd, $J=12,3 \mathrm{~Hz}, \mathrm{H}-5 a x$ ), 1.06 (dd, $J=6.8 \mathrm{~Hz}, 3 \mathrm{H}, \mathrm{Me}$ ), 1.21-1.34 (m, 5H, H-4ax, H-4a, H-6ax, H-7ax, H-8ax), 1.54 (dm, $J=12 \mathrm{~Hz}, \mathrm{H}-5 e q), 1.69(\mathrm{dm}, J=12 \mathrm{~Hz}, \mathrm{H}-6 e q), 1.77$ (dm, $2 \mathrm{H}, J=12 \mathrm{~Hz}, \mathrm{H}-7 e q, \mathrm{H}-8$ eq), 1.88 (dd, $J=10.8$, $3.2 \mathrm{~Hz}, \mathrm{H}-4 e q$ ), 2.12 (s, 3H, OAc), 2.22 (td, $J=10,3.2 \mathrm{~Hz}$, $\mathrm{H}-8 \mathrm{a}$ ), 2.92 (qd, $J=6.4,1.6 \mathrm{~Hz}, \mathrm{H}-2 a x$ ), 4.88 (ddd, $J=3.2$, $3.2,1.6 \mathrm{~Hz}, \mathrm{H}-3 e q) ;{ }^{13} \mathrm{C}$ NMR see Table 1. HRFABMS calcd for $\mathrm{C}_{12} \mathrm{H}_{22} \mathrm{NO}_{2}\left(\mathrm{M}^{+}+1\right)$ 212.1651, found 212.1648.
3.1.7. 2-[(2R,3S)-2-Acetoxy-3-(dibenzylamino)butyl]-1-butylcyclohex-2-en-1-ol (8a). To a cooled ( $-78^{\circ} \mathrm{C}$ ) solution of $5 \mathbf{5 a}(105 \mathrm{mg}, 0.258 \mathrm{mmol})$ in THF ( 3 mL ) was added $n-\mathrm{BuLi}(1.6 \mathrm{M}$ in hexanes, $0.8 \mathrm{~mL}, 1.29 \mathrm{mmol})$ and the reaction mixture was stirred for 4 h , the temperature slowly rising to rt. The reaction was quenched by addition of saturated aqueous $\mathrm{NH}_{4} \mathrm{Cl}(20 \mathrm{~mL})$ and extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}(3 \times 20 \mathrm{~mL})$. The dried organic extracts were concentrated and the residue was dissolved in pyridine $(1 \mathrm{~mL})$ and treated with $\mathrm{Ac}_{2} \mathrm{O}(0.25 \mathrm{~mL}, 2.58 \mathrm{mmol})$ and DMAP ( $5 \mathrm{mg}, 0.04 \mathrm{mmol}$ ). The reaction mixture was stirred overnight at rt, saturated aqueous $\mathrm{NaHCO}_{3}(10 \mathrm{~mL})$ was added and the mixture was extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}(3 \times$ 15 mL ). The dried organic extract was concentrated and purified by chromatography $\left(\mathrm{SiO}_{2}\right.$, hexane/EtOAc 8:2) to give the epimeric alcohols 8a and 1-epi-8a ( $81 \mathrm{mg}, 68 \%$ ), in a $1: 1$ ratio according to the NMR spectrum, which were used directly in the next step.Compound $\mathbf{8 a} . R_{\mathrm{f}}=0.82\left(\mathrm{SiO}_{2}\right.$, 8:2 hexane/EtOAc); ${ }^{1} \mathrm{H}$ NMR ( $200 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) 0.92 (t, $J=6.8 \mathrm{~Hz}, 3 \mathrm{H}), 1.11(\mathrm{~d}, J=7.0 \mathrm{~Hz}, 3 \mathrm{H}), 1.18-1.38(\mathrm{~m}, 4 \mathrm{H})$, $1.49-1.80(\mathrm{~m}, 7 \mathrm{H}), 1.82-1.93(\mathrm{~m}, 2 \mathrm{H}), 1.98(\mathrm{~s}, 3 \mathrm{H}), 2.75$ (quint, $J=7 \mathrm{~Hz}, 1 \mathrm{H}), 2.95(\mathrm{dm}, J=12 \mathrm{~Hz}, 1 \mathrm{H}), 3.44(\mathrm{~d}, J=$ $13.6 \mathrm{~Hz}, 2 \mathrm{H}), 3.75$ (d, $J=13.6 \mathrm{~Hz}, 2 \mathrm{H}), 5.29-5.40(\mathrm{~m}, 2 \mathrm{H})$, 7.18-7.40 (m, 10H). Compound 1-epi-8a. $R_{\mathrm{f}}=0.64\left(\mathrm{SiO}_{2}\right.$, 8:2 hexane/EtOAc); ${ }^{1} \mathrm{H}$ NMR ( $200 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) 0.89 (t, $J=6.6 \mathrm{~Hz}, 3 \mathrm{H}), 1.06(\mathrm{~d}, J=6.6 \mathrm{~Hz}, 3 \mathrm{H}), 1.18-1.38(\mathrm{~m}, 4 \mathrm{H})$, $1.40-1.70(\mathrm{~m}, 7 \mathrm{H}), 1.74-1.88(\mathrm{~m}, 2 \mathrm{H}), 2.00(\mathrm{~s}, 3 \mathrm{H}), 2.44-$ $2.54(\mathrm{~m}, 1 \mathrm{H}), 2.85$ (quint, $J=7 \mathrm{~Hz}, 1 \mathrm{H}), 3.48(\mathrm{~d}, J=$ $13.6 \mathrm{~Hz}, 2 \mathrm{H}), 3.73$ (d, $J=13.6 \mathrm{~Hz}, 2 \mathrm{H}), 5.30(\mathrm{~m}, 1 \mathrm{H}), 5.39$ (t, $J=3.9 \mathrm{~Hz}, 1 \mathrm{H}$ ), 7.18-7.40 (m, 10H).
3.1.8. 2-[(2S,3S)-2-Acetoxy-3-(dibenzylamino)butyl]-1-butylcyclohex-2-enol (8b). Operating as above, starting from $147 \mathrm{mg}(0.36 \mathrm{mmol})$ of cyclohexenone $\mathbf{5 b}$, and after chromatography $\left(\mathrm{SiO}_{2}\right.$, hexane/EtOAc 8:2), $85 \mathrm{mg}(51 \%)$ of 8b was obtained: $R_{\mathrm{f}}=0.58\left(\mathrm{SiO}_{2}, 8: 2\right.$ hexane/EtOAc $) ;{ }^{1} \mathrm{H}$ NMR ( $200 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $0.91(\mathrm{t}, J=6.8 \mathrm{~Hz}, 3 \mathrm{H}), 1.09(\mathrm{~d}$, $J=7 \mathrm{~Hz}, 3 \mathrm{H}), 1.20-1.40(\mathrm{~m}, 5 \mathrm{H}), 1.42-1.78(\mathrm{~m}, 6 \mathrm{H}), 1.84-$ $1.96(\mathrm{~m}, 2 \mathrm{H}), 2.05(\mathrm{~s}, 3 \mathrm{H}), 2.18-2.28(\mathrm{~m}, 1 \mathrm{H}), 2.85$ (quint, $J=7 \mathrm{~Hz}, 1 \mathrm{H}), 3.37(\mathrm{~d}, J=13.5 \mathrm{~Hz}, 2 \mathrm{H}), 3.90(\mathrm{~d}, J=$ $13.5 \mathrm{~Hz}, 2 \mathrm{H}), 5.13-5.22(\mathrm{~m}, 1 \mathrm{H}), 5.45(\mathrm{t}, J=3.7 \mathrm{~Hz}, 1 \mathrm{H})$, 7.18-7.40 (m, 10H).
3.1.9. 2-[(2R,3S)-2-Acetoxy-3-(dibenzylamino)butyl]-3-butylcyclohex-2-enone (9a). To a solution of epimeric alcohols $8 \mathbf{a}\left(81 \mathrm{mg}, 0.18 \mathrm{mmol}\right.$ ) in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(2 \mathrm{~mL})$ were added PCC ( $57 \mathrm{mg}, 0.26 \mathrm{mmol}$ ) and $\mathrm{SiO}_{2}(57 \mathrm{mg})$, and the mixture was stirred overnight at rt . The residue obtained after evaporation of the solvent was purified by chromatography $\left(\mathrm{SiO}_{2}\right.$, hexane/EtOAc 9:1) to give 9 a as a viscous oil ( $50 \mathrm{mg}, 62 \%$ ): $R_{\mathrm{f}}=0.36\left(\mathrm{SiO}_{2}, 8: 2\right.$ hexane/EtOAc); $[\alpha]_{\mathrm{D}}^{20}$ $-5.3\left(c 0.3\right.$ in $\left.\mathrm{CHCl}_{3}\right) ;{ }^{1} \mathrm{H}$ NMR $\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 0.91(\mathrm{t}$,
$J=6.8 \mathrm{~Hz}, 3 \mathrm{H}), 1.11(\mathrm{~d}, J=6.4 \mathrm{~Hz}, 3 \mathrm{H}), 1.20-1.50(\mathrm{~m}, 4 \mathrm{H})$, $1.70-1.8(\mathrm{~m}, 2 \mathrm{H}), 1.92(\mathrm{~s}, 3 \mathrm{H}), 1.98-2.33(\mathrm{~m}, 6 \mathrm{H}), 2.34-$ $2.42(\mathrm{~m}, 1 \mathrm{H}), 2.71$ (quint, $J=6.8 \mathrm{~Hz}, 1 \mathrm{H}), 3.08$ (dd, $J=$ $13.6,4.4 \mathrm{~Hz}, 1 \mathrm{H}$ ), 3.45 (d, $J=13.6 \mathrm{~Hz}, 2 \mathrm{H}$ ), 3.75 (d, $J=$ $13.6 \mathrm{~Hz}, 2 \mathrm{H}$ ), 5.20 (ddd, $J=8.8,6.8,5.2 \mathrm{~Hz}, 1 \mathrm{H}$ ), 7.18-7.40 (m, 10H); ${ }^{13} \mathrm{C}$ NMR ( $50 \mathrm{MHz}, \mathrm{CDCl}_{3}$, HSQC) $8.9\left(\mathrm{CH}_{3}\right)$, $14.1\left(\mathrm{CH}_{3}\right)$, $21.2\left(\mathrm{CH}_{3}\right), 22.4\left(\mathrm{CH}_{2}\right), 22.9\left(\mathrm{CH}_{2}\right), 28.3\left(\mathrm{CH}_{2}\right)$, $30.1\left(\mathrm{CH}_{2}\right), 30.8\left(\mathrm{CH}_{2}\right), 34.7\left(\mathrm{CH}_{2}\right), 37.7\left(\mathrm{CH}_{2}\right), 54.0\left(\mathrm{CH}_{2}\right)$, $55.2(\mathrm{CH}), 75.0(\mathrm{CH}), 126.7(\mathrm{CH}), 128.1(\mathrm{CH}), 128.9(\mathrm{CH})$, 131.3 (C), 140.1 (C), 160.9 (C), 170.4 (C), 198.7 (C). Anal. Calcd for $\mathrm{C}_{30} \mathrm{H}_{39} \mathrm{NO}_{3} \cdot \mathrm{H}_{2} \mathrm{O}: \mathrm{C}, 75.12 ; \mathrm{H}, 8.62 ; \mathrm{N}, 2.92$. Found C, 75.48; H, 9.02; N, 2.58.
3.1.10. 2-[(2S,3S)-2-Acetoxy-3-(dibenzylamino)butyl]-3-butylcyclohex-2-enone (9b). Operating as above, starting from $71 \mathrm{mg}(0.15 \mathrm{mmol})$ of alcohol $\mathbf{8 b}$ and after chromatography ( $\mathrm{SiO}_{2}, 9: 1$ hexane/EtOAc), enone 9b ( 41 mg , $61 \%$ ) was isolated as a viscous oil; ${ }^{1} \mathrm{H}$ NMR $(400 \mathrm{MHz}$, $\left.\mathrm{CDCl}_{3}\right) 0.89(\mathrm{t}, J=7.2 \mathrm{~Hz}, 3 \mathrm{H}), 1.10(\mathrm{~d}, J=6.8 \mathrm{~Hz}, 3 \mathrm{H})$, $1.20-1.32(\mathrm{~m}, 2 \mathrm{H}), 1.32-1.44(\mathrm{~m}, 2 \mathrm{H}), 1.81-1.88(\mathrm{~m}, 2 \mathrm{H})$, $1.96(\mathrm{~s}, 3 \mathrm{H}), 1.96-2.03(\mathrm{~m}, 2 \mathrm{H}), 2.16-2.42(\mathrm{~m}, 6 \mathrm{H}), 2.34-$ 2.42 (m, 1H), 2.68 (dd, $J=13.8,11.0 \mathrm{~Hz}$ ), 2.89-2.96 (m, 1 H ), 3.39 (d, $J=13.6 \mathrm{~Hz}$ ), 3.90 (d, $J=13.6 \mathrm{~Hz}$ ), 5.08 (ddd, $J=11.0,5.8,2.4 \mathrm{~Hz}, 1 \mathrm{H}), 7.18-7.40(\mathrm{~m}, 10 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR $\left(50 \mathrm{MHz}, \mathrm{CDCl}_{3}, \mathrm{HSQC}\right) 9.7\left(\mathrm{CH}_{3}\right), 14.1\left(\mathrm{CH}_{3}\right), 21.1$ $\left(\mathrm{CH}_{3}\right), 22.4\left(\mathrm{CH}_{2}\right), 23.0\left(\mathrm{CH}_{2}\right), 28.4\left(\mathrm{CH}_{2}\right), 30.1\left(\mathrm{CH}_{2}\right), 30.9$ $\left(\mathrm{CH}_{2}\right), 34.8\left(\mathrm{CH}_{2}\right), 37.8\left(\mathrm{CH}_{2}\right), 54.5\left(\mathrm{CH}_{2}\right), 55.8(\mathrm{CH}), 75.9$ $(\mathrm{CH}), 126.7(\mathrm{CH}), 128.1(\mathrm{CH}), 128.7(\mathrm{CH}), 131.7(\mathrm{C})$, 140.22 (C), 160.1 (C), 170.1 (C), 198.8 (C).
3.1.11. Aminocyclization of 9a. Following the above procedure for the aminocyclization of 5a using enone $\mathbf{9 a}$ ( $38 \mathrm{mg}, 0.08 \mathrm{mmol}$ ) and carrying out the hydrogenation process for 36 h , the crude product was purified by chromatography $\left(\mathrm{Al}_{2} \mathrm{O}_{3}\right.$, from $9: 1$ to $7: 3$ hexane/EtOAc) to give $7 \mathrm{mg}(33 \%)$ of $\mathbf{1 0 a}$ and $8 \mathrm{mg}(38 \%)$ of 11a, both as colorless oils.
( $2 S, 3 R, 4 \mathrm{a} S, 5 R, 8 \mathrm{a} R$ )-3-Acetoxy-5-butyl-2-methyldecahydroquinoline (10a). $R_{\mathrm{f}}=0.59\left(\mathrm{Al}_{2} \mathrm{O}_{3}, 8: 2\right.$ hexane/ EtOAc); $[\alpha]_{\mathrm{D}}^{20}-34.5$ (c 0.5 in $\mathrm{CHCl}_{3}$ ); ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$, COSY) $0.90\left(\mathrm{~m}, 1 \mathrm{H}, \mathrm{H}-1^{\prime}\right), 0.90$ (t, $\left.J=6.8 \mathrm{~Hz}, 3 \mathrm{H}, \mathrm{H}-4^{\prime}\right), 1.10(\mathrm{~d}, J=6.4 \mathrm{~Hz}, \mathrm{Me}), 1.12$ (masked, H-8ax), $1.20\left(\mathrm{~m}, 4 \mathrm{H}, \mathrm{H}-6\right.$ and $\left.\mathrm{H}-2^{\prime}\right), 1.25$ (m, $\left.2 \mathrm{H}, \mathrm{H}-3^{\prime}\right), 1.30$ (m, H-4ax), 1.40 (m, H-4a), 1.48 (m, 2H, H-7), 1.5 (m, H-8eq), 1.70 (m, 2H, H-5ax, H-1'), 2.04 (s, $3 \mathrm{H}, \mathrm{OAc}$ ), 2.27 (ddd, $J=12.4,3.6,2.8 \mathrm{~Hz}, \mathrm{H}-4 e q$ ), 2.76 (dq, $J=10,6.5 \mathrm{~Hz}, \mathrm{H}-2 a x$ ), 2.97 (br s, H-8a), 4.49 (td, $J=10.4$, $4.4 \mathrm{~Hz}, \mathrm{H}-3 \mathrm{ax})$; ${ }^{13} \mathrm{C}$ NMR see Table 1. HRFABMS calcd for $\mathrm{C}_{16} \mathrm{H}_{30} \mathrm{NO}_{2}\left(\mathrm{M}^{+}+1\right) 268.2198$, found 268.2202.
( $2 S, 3 R, 4 \mathrm{a} R, 5 S, 8 \mathrm{a} R$ )-3-Acetoxy-5-butyl-2-methyldecahydroquinoline (11a). $R_{\mathrm{f}}=0.28\left(\mathrm{Al}_{2} \mathrm{O}_{3}, 8: 2\right.$ hexane/ EtOAc); $[\alpha]_{\mathrm{D}}^{20}-4.3$ (c 0.3 in $\mathrm{CHCl}_{3}$ ); ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$, COSY) 0.88 (t, $J=6.8 \mathrm{~Hz}, 3 \mathrm{H}, \mathrm{H}-4^{\prime}$ ), 0.90 (masked, 1H, H-6ax), 0.94 (m, 2H, H-4ax, H-4a), 1.05 (masked, 2H, H-5 and H-1'), 1.07 (d, $J=6.4 \mathrm{~Hz}, 3 \mathrm{H}, \mathrm{Me}$ ), 1.15 (m, 1H, H-8ax), 1.25 (m, 4H, H-2', H-3'), 1.30 (m, 1H, $\mathrm{H}-7 a x), 1.45$ ( $\mathrm{m}, 1 \mathrm{H}, \mathrm{H}^{1}$ ), 1.75 (m, 3H, H-6, H-7, H-8), 2.05 (s, 3H, OAc), 2.20 (ddd, $J=11,9,3 \mathrm{~Hz}, \mathrm{H}-8 \mathrm{a}$ ), 2.29 (dm, $J=12 \mathrm{~Hz}, \mathrm{H}-4 e q), 2.70(\mathrm{dq}, J=10.4,6.4 \mathrm{~Hz}, \mathrm{H}-2 a x)$, $4.41(\operatorname{td}, J=10.4,4.8 \mathrm{~Hz}, \mathrm{H}-3 a x) ;{ }^{13} \mathrm{C}$ NMR see Table 1.

HRFABMS calcd for $\mathrm{C}_{16} \mathrm{H}_{30} \mathrm{NO}_{2}\left(\mathrm{M}^{+}+1\right)$ 268.2198, found 268.2203.
3.1.12. Aminocyclization of $9 \mathbf{b}$. Operating as in the cyclization of 9a, from enone $\mathbf{1 1}(22 \mathrm{mg}, 0.05 \mathrm{mmol})$ was obtained 11b as an oil ( $6 \mathrm{mg}, 52 \%$ ) after chromatography $\left(\mathrm{Al}_{2} \mathrm{O}_{3}\right.$, from 9:1 to 7:3 hexane/EtOAc). ${ }^{31}$
( $2 S, 3 S, 4 \mathrm{a} R, 5 S, 8 \mathrm{a} R$ )-3-Acetoxy-5-butyl-2-methyldecahydroquinoline (11b). $R_{\mathrm{f}}=0.13\left(\mathrm{Al}_{2} \mathrm{O}_{3}, 8: 2\right.$ hexane/ EtOAc); ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}$, COSY) 0.87 (t, $J=$ $\left.6.8 \mathrm{~Hz}, 3 \mathrm{H}, \mathrm{H}^{\prime}-4\right), 0.98$ (m, 4H, H-4a, H-5, H-6, H-1'), 1.06 (d, $J=6.8 \mathrm{~Hz}, 3 \mathrm{H}, \mathrm{Me}), 1.15$ (m, 2H, H-4ax, H-8ax), 1.25 ( $\mathrm{m}, 4 \mathrm{H}, \mathrm{H}-2^{\prime}$ and $\mathrm{H}-3^{\prime}$ ), 1.30 (m, H-7ax), 1.45 (m, 1H, H-1'), 1.77 (m, 3H, H-8eq, H-7eq, H-6eq), 2.11 (s, 3H, OAc), 2.20 (dt, $J=10,3.2 \mathrm{~Hz}, \mathrm{H}-4 e q), 2.27$ (td, $J=10,3 \mathrm{~Hz}, \mathrm{H}-8 \mathrm{a}$ ), 2.90 (qd, $J=6.4,1.6 \mathrm{~Hz}, \mathrm{H}-2 a x$ ), 4.91 (ddd, $J=3.2,3.2$, $1.6 \mathrm{~Hz}, \mathrm{H}-3 \mathrm{eq}) ;{ }^{13} \mathrm{C}$ NMR see Table 1. HRFABMS calcd for $\mathrm{C}_{16} \mathrm{H}_{30} \mathrm{NO}_{2}\left(\mathrm{M}^{+}+1\right) 268.2198$, found 268.2194 .

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Nom: H. Min ${ }^{3} / 05$
exp 3 gcos $\gamma$




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${ }^{6}$
116

Line broadening $0,3 \mathrm{~Hz}$
FT size 32768






## 6.2

Ring expansion of Functionalized Octahydroindoles to Enantiopure cisdecahydroquinolines

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# Ring Expansion of Functionalized 

## Octahydroindoles to Enantiopure cis-

## Decahydroquinolines ${ }^{\dagger}$

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## Table of C ontents Graphic



ABSTRACT A new synthetic entry to enantiopure cis-decahydroquinolines is reported Endo and exo derivatives of cis-1-benzyl-2-(hydroxymethyl)octahydroindol-6-one ethylene acetal undergo ring-enlargement upon treatment with TFAA and then $E t_{3} \mathrm{~N}$ (thermodynamic conditions) to give enantiopure 1-benzyl-3-hydroxydecahydroquinolin-7-one derivatives in 77\% and 82\% yield, respectively For 2-(1-hydroxyethyl) analogs, the best synthetic result is obtained from the $(2 S, 1$ 'R) endo isomer, which under kinetic reaction conditions ( $\mathrm{M} \mathrm{sCl}, \mathrm{THF},-20^{\circ} \mathrm{C}$, then AgOAc at rt) gives the expanded product in 54\% yield
cis-Decahydroquinoline constitutes the azabicyclic skeleton of natural products such as lepadins ${ }^{1}$ and several amphibian alkaloids, ${ }^{2}$ as well as some pharmacologically interesting synthetic compounds ${ }^{3} \mathrm{M}$ oreover, this heterocyclic motif occurs as a subunit of other azapolycyclic natural products (e.g. gephyrotoxins, ${ }^{4}$ cylindricines, ${ }^{5}$ and pseudoaspidopermidine and pandoline alkaloids ${ }^{6}$ ) (Figure 1) The extensive occurrence of this azabicyclic ring has stimulated the implementation of new procedures to gain access to functionalized enantiopure cis-decahydroquinolines that can be used as advanced intermediates in the synthesis of compounds embodying this skeleton-type ${ }^{7}$

lepadin F
$\mathrm{R}_{1}=\left(\mathrm{CH}_{2}\right)_{4} \mathrm{CHOH}-\mathrm{CH}_{2} \mathrm{CH}_{2} \mathrm{CH}_{3}$ $\mathrm{R}_{2}=\mathrm{COCH}=\mathrm{CH}\left(\mathrm{CH}_{2}\right)_{4} \mathrm{CH}_{3}$

gephyrotoxin

cylindricine C

pandoline

FIGURE 1. Natural and synthetic compounds embodying the cis-decahydroquinoline framework

In this paper we report the studies devoted to the ring-enlargement of cisoctahydroindole derivatives to cis-decahydroquinolines The diastereoselective ring expansion of monocyclic amines (azetidines, ${ }^{8}$ pyrrolidines, ${ }^{9,10}$ pyrrolines, ${ }^{11}$ piperidines ${ }^{12}$ ) and bicyclic amines (2-azabicyclo[3 30 0]octanes, ${ }^{13}$ 1azabicyclo[2 2 2]octanes, ${ }^{14}$ indolizidines, ${ }^{15}$ indolines, ${ }^{16}$ hexahydropyrrolo[3,4d]isoxazoles ${ }^{17}$ ) with a hydroxymethyl substituent adjacent to the nitrogen atom is a well known process, ${ }^{18,19}$ but it is unprecedented in octahydroindole compounds Considering the precedents, we envisaged that the transformation (Scheme 1) would occur via aziridinium intermediates, ${ }^{20}$ once the hydroxyl group is converted into a good leaving group A ring opening at the fused carbon atom would then lead to a new heterocyclic derivative with an expanded ring if the leaving group were a chloride or trifluoroacetate, in absence of another nucleophile in the reaction medium, the process would be reversible and the ratio of the expanded and non-expanded compounds would reflect their thermodynamic stability $O n$ the contrary, if $X^{-}$were an acetate and the
chloride ion were taken out of the reaction medium using silver acetate, the ring opening of the aziridinium ion would be irreversible and the ratio of the formed compounds would arise from a kinetic control

SCHEME 1. The synthetic approach to enantiopure cis-decahydroquinolines

reversible when $\mathrm{X}=\mathrm{Cl}$ or $\mathrm{OCOCF}_{3}$
irreversible when $\mathrm{X}=\mathrm{OAc}$

Our study of the stereospecific rearrangement of 2-hydroxymethyl- and 2-( $\alpha$ hydroxyethyl)octahydroindoles to 3-substituted and 2,3-disubstituted decahydroquinolines began with the preparation of the rearrangement precursors (Scheme 2) The starting materials were the azabicyclic esters endo 1 and exo 2 , which were available from 0 -methyltyrosine in three steps (Birch reduction, aminocyclization promoted by $\mathrm{MeOH}-\mathrm{HCl}$, and benzylation) ${ }^{21}$ B oth esters were protected to give acetals 3 and 4, which, in turn, were reduced with $\mathrm{LiBH}_{4}$ to the corresponding primary alcohols 5 and 6 in $80 \%$ overall yield in each series On the other hand, the preparation of the secondary alcohols was carried out as follows: esters 3 and 4 were transformed to methyl ketones 7 (endo) and 8 (exo), respectively, in a two-step sequence involving the formation of their corresponding Weinreb amides, ${ }^{22}$ followed by coupling with methylmagnesium bromide Then, ketones 7 and 8 were both reduced with $\mathrm{NaBH}_{4}$ to give a mixture of secondary alcohols 9a and 10a (55:45 ratio $)^{23}$ in the endo series and 11a and 12a (63:37 ratio) in the exo series At this point, improving the
diastereoselectivity of the reduction was not a priority since the availability of all the diastereomers would help evaluate the scope and limitations of the enlargement process

SCHEM E 2. Synthesis of cis-2-hydroxymethyl- and 2-( $\alpha-$
hydroxyethyl)octahydroindole derivatives


Two ${ }^{13} \mathrm{C}$ NMR features clearly differentiate the endo and exo series of cis-2-substituted octahydroindol-6-ones (Table 1, Supporting Information): (i) the chemical shift of the benzylic carbon resonates at a higher field in exo compounds ( $\delta 515-540$ ) than in the endo compounds ( $\delta 59-63$ ); (ii) the C-7 signal appears at lower values in the exo compounds ( $\delta 29-32$ ) than in the endo isomers ( $\delta 37-39$ ) It is worth noting that the stereochemistry at C-1' for alcohols 9a-12a was only unequivocally established after the stereochemical elucidation of the expanded decahydroquinolines (cf. vide infra) The conformationally mobile cis-octahydroindole system ${ }^{21}$ has two conformers ( $N$-outside
and N -inside) in which $\mathrm{H}-7 \mathrm{a}$ is axial or equatorial, respectively, with respect to the carbocyclic ring NMR studies allowed us to assign the conformational preference of the described cis-2-substituted octahydroindol-6-one derivatives, in which the coupling constants $\mathrm{H} 7-\mathrm{H} 7 \mathrm{a}$ (one of them of 11 Hz ) are consistent with antiperiplanar couplings, indicating that the $\mathrm{H}-7 \mathrm{a}$ proton ${ }^{24}$ is axially located with respect to the carbocyclic ring (Figure 2)




6
FIGURE 2. Preferred conformation of endo and exo compounds 5 and 6

Treatment of 2-(hydroxymethyl)octahydroindole derivative 5 with TFAA in THF followed by the addition of triethylamine ${ }^{9}$ led to the ring-expanded product, which after a hydrolytic work-up (aqueous NaOH ) allowed the isolation of decahydroquinoline 13 (Scheme 3) The NM R data (see below) of this compound proves that the configuration of $\mathrm{C}-3$ in (-)-13 is R , which supports the mechanism depicted in Scheme 1 (stereocontrolled process during the nucleophilic attack at C-2 of the aziridinium intermediate) The same protocol (TFAA/Et $\mathrm{N} / \mathrm{NaOH}$ ) was also applied to the exo isomer 6, decahydroquinoline (+)-14 being isolated as a single isomer in $82 \%$ yield Both decahydroquinolines show the same preferred conformation according to their NMR spectra Thus, the coupling constants of $\mathrm{H}-4 \mathrm{a}$ and $\mathrm{H}-8 \mathrm{a}$ in each diastereomer are in accordance with an axial and an equatorial relationship, respectively, with respect to the N -containing ring ( N -exo conformation) ${ }^{25}$ The ${ }^{13} \mathrm{C}$ NMR data corroborate the above stereochemical elucidation since both 13 and 14 show a relative upfield for $\mathrm{C}-2$ and C -

8, a characteristic feature of cis-decahydroquinolines in the N -exo conformation (see Figure 3)

SCHEME 3 Synthesis of 3-hydroxydecahydroquinolines


When we applied the same procedure (TFAA, THF, then $\mathrm{Et}_{3} \mathrm{~N}$ ) to the secondary alcohols 9a-12a, i e under thermodynamic reaction conditions, the results were disappointing since after long reaction times (4-5 days at reflux temperature) only the starting material and degradation products were obtained ${ }^{26,27}$ The next attempt to expand the ring involved converting alcohols 9a-12a to the secondary chlorides 9b-12b, which were then heated in refluxing THF in all cases, the unexpanded 2-( $\alpha$ chloroethyl)octahydroindoles 9b-12b were isolated (Scheme 4) ${ }^{28}$ Since the configuration at $\mathrm{C}-1$ ' was retained, these results show that the process occurred through an aziridinium salt intermediate and that chlorides 9b-12b were either directly formed by opening of an aziridinium by the chloride ion or by a reversion process from an initially expanded product Thus, under thermodynamic reaction conditions, we were unable to achieve 2,3-disubstituted decahydroquinolines from octahydroindoles 9a$12 a^{29}$

SCHEME 4. Attempted ring expansion under thermodynamic conditions

(see references 26 and 28)

In order to have an irreversible ring opening of the aziridinium intermediate, the chlorides $9 \mathrm{~b}-12 \mathrm{~b}$ were treated with AgOAc in a THF solution at reflux temperature (M ethod A) Under these kinetic conditions, the ring expanded decahydroquinolines 1719 were formed in variable yields (Scheme 5), and 20 was only detected in the GC-M S analysis The non-enlarged acetates 9c-12c formed by the acetate attack on the carbon linked to the methyl group in the aziridinium intermediate, were also isolated ${ }^{30}$ The best results from a synthetic point of view were obtained for the endo compound 9b, since it was only in this series that the decahydroquinoline derivative was isolated as the main product $W$ e then decided to carry out the process starting from alcohols 9a-12a, working at $-20^{\circ} \mathrm{C}$ to avoid the formation of the corresponding chlorides, ${ }^{10 \mathrm{c}}$ and promoting the ring opening of the aziridinium with AgOAc at room temperature (M ethod B) This decrease in temperature led to a slight increase in the ratio of the thermodynamically unfavoured decahydroquinolines (see Scheme 5) Not unexpectedly, in the endo series of 2-( $\alpha$-hydroxyethyl)octahydroindoles the best result was obtained from alcohol 9a, which was transformed into the decahydroquinoline 17 in 54-58\% yield ${ }^{31,32}$ The reason was that the aziridinium intermediate was formed and opened without generating steric repulsion due to the antiperiplanar relationship between the methyl group and the $\mathrm{C}(2)-\mathrm{C}(3)$ bond On the contrary, the reason for the poor yields
observed in the ring-expanded compounds in the exo series (11a and 12a) is unclear, even taking into consideration that the transition states between 12 a and decahydroquinoline 20 are likely to be the most sterically demanding of the four pathways leading to expanded compounds Figure 3 depicts the preferred conformation of all the synthesized 3-oxygenated cis-decahydroquinolines

SCHEME 5. Ring expansion under kinetic conditions

9c

$\qquad$

11c
19

12c
20

M ethod A: (i) M sCl, $\mathrm{Et}_{3} \mathrm{~N}$, THF reflux, 4h
(9b-12b formed); (ii) AgOAc, THF reflux, 4 h

M ethod B: (i) $\mathrm{MsCl}, \mathrm{Et}_{3} \mathrm{~N}, \mathrm{THF},-20^{\circ} \mathrm{C}$; (ii) $\mathrm{AgOAc}, \mathrm{rt}, \mathrm{lh}$

$13 R^{1}=R^{2}=H ; R^{3}=H$
$17 R^{1}=H ; R^{2}=M e ; R^{3}=A c$
$18 R^{1}=\mathrm{Me} ; \mathrm{R}^{2}=\mathrm{H} ; \mathrm{R}^{3}=\mathrm{Ac}$
$21 R^{1}=H ; R^{2}=M e ; R^{3}=H$

$14 \mathrm{R}^{1}=\mathrm{R}^{2}=\mathrm{H}$
$15 \mathrm{R}^{1}=\mathrm{Me} ; \mathrm{R}^{2}=\mathrm{H}$
$19 R^{1}=\mathrm{Me} ; \mathrm{R}^{2}=A c$

FIGURE 3 Stereochemistry of cis-decahydroquinolines

Finally, to obtain a better understanding of the ring enlargement process, decahydroquinoline 21 (obtained under kinetic conditions from 9 a$)^{31}$ was submitted to the thermodynamic conditions of the aziridinium ring formation and opening (TFAA, then $\mathrm{Et}_{3} \mathrm{~N}$ followed by heating at reflux for 8 h , and ending with an aqueous NaOH treatment) Under these conditions, octahydroindole 9a was formed, albeit as the only product (Scheme 6) This result confirms that 2-( $\alpha$-hydroxyethyl)octahydroindoles are more stable than 2-methyl-3-hydroxyquinolines in the series of compounds examined (9-12 vs 17-20)

## SCHEME 6


thermodynamic conditions

In summary, the study of the ring enlargement of cis-octahydroindole derivatives has given access to valuable functionalized enantiopure cis-decahydroquinolines (13 and 14, in excellent yields, and 17 in good yield), which could be used as building blocks in the synthesis of natural products $M$ oreover, it has been shown that subtle stereochemical differences in the octahydroindoles studied can have a significant impact on the ringexpansion pathway when the process is carried out under thermodynamic or kinetic conditions

## Experimental Section

(3R,4aS,8aS)-1-Benzyl-3-hydroxy-7-oxodecahydroquinoline ethylene acetal (13). To a solution of alcohol 5 ( $223 \mathrm{mg}, 074 \mathrm{mmol}$ ) in THF ( 2 mL ) cooled to $-78{ }^{\circ} \mathrm{C}$ was added TFAA ( $021 \mathrm{~mL}, 147 \mathrm{mmol}, 2$ equiv) and the reaction mixture was stirred for 3 h at this temperature $\mathrm{Et}_{3} \mathrm{~N}$ ( $05 \mathrm{~mL}, 368 \mathrm{mmol}$, 5 equiv) was added and after 15 min , the reaction mixture was heated at reflux for 20 h The mixture was cooled to $25^{\circ} \mathrm{C}$, and 25 N NaOH (15 mL, 50 equiv) was added A fter stirring for 3 h , the reaction mixture was extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}(3 \times 20 \mathrm{~mL})$ The organic extracts were dried and concentrated to give an oil, which was purified by chromatography $\left(\mathrm{SiO}_{2}, 1 \%\right.$ to $5 \% \mathrm{MeOH}$ in $\left.\mathrm{CH}_{2} \mathrm{Cl}_{2}\right)$ to give $173 \mathrm{mg}(77 \%)$ of 13 as a colorless oil: $\mathrm{R}_{\mathrm{f}}=035\left(\mathrm{SiO}_{2}, \mathrm{CH}_{2} \mathrm{Cl}_{2} / \mathrm{M} \mathrm{eOH} 95: 5\right)$; $[\alpha]_{D}{ }^{20}-44\left(\mathrm{c} 03, \mathrm{CHCl}_{3}\right) ;{ }^{1} \mathrm{H} \operatorname{NMR}\left(500 \mathrm{MHz}, \mathrm{gCOSY}, \mathrm{CDCl}_{3}\right) 145(\mathrm{dm}, \mathrm{J}=105 \mathrm{~Hz}$, 1H, H-5eq), 147 (m, 1H, H-6eq), 150 (q, J = 105 Hz, 1H, H-4ax), 155 (m, 1H, H6ax), 158 (m, 1H, H-4eq), 161 (ddd, J = 12 5, 45, $20 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H}-8 \mathrm{eq}$ ), 173 (tt, J = $135,50 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H}-5 a x), 182(\mathrm{t}, \mathrm{J}=125 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H}-8 \mathrm{ax}), 197(\mathrm{dm}, \mathrm{J}=105 \mathrm{~Hz}, 1 \mathrm{H}$, H-4a), 210 (t, J = $105 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H}-2 \mathrm{ax}), 262$ (ddd, J = $105,50,15 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H}-2 \mathrm{eq}$ ), 300 (dt, J = 125, $45 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H}-8 \mathrm{a}), 345$ and 366 (2d, J = $125 \mathrm{~Hz}, 1 \mathrm{H}$ each, $\mathrm{NCH}_{2} \mathrm{Ar}$ ), 368 (dddd, J = $\left.105,105,50,50,1 \mathrm{H}, \mathrm{H}-3 \mathrm{ax}\right), 380-390\left(\mathrm{~m}, 4 \mathrm{H}, \mathrm{OCH}_{2}\right)$,

7 20-7 30 (m, 5H, ArH); ${ }^{13} \mathrm{C}$ NMR (75 M Hz, gHSQC) 269 (C-5), 270 (C-8), 299 (C6), 328 (C-4a), 331 (C-4), 520 (C-2), 570 (C-8a), $583\left(\mathrm{NCH}_{2}\right), 641$ and 642 $\left(\mathrm{OCH}_{2}\right), 681$ (C-3), 1099 (C-7), 126 8, 1281,129 5, 1393 (Ar) Anal Calcd for $\mathrm{C}_{18} \mathrm{H}_{25} \mathrm{NO}_{3}: \mathrm{C} 7126, \mathrm{H} 831, \mathrm{~N} 462$ Found: C $7086, \mathrm{H} 803, \mathrm{~N} 438$
(3R,4aR,8aR)-1-Benzyl-3-hydroxy-7-oxodecahydroquinoline ethylene acetal (14). Operating as above, alcohol $6(464 \mathrm{mg}, 153 \mathrm{mmol})$ in THF ( 4 mL ) was treated with TFAA ( $043 \mathrm{~mL}, 304 \mathrm{mmol}, 2$ equiv), and then with $\mathrm{Et}_{3} \mathrm{~N}$ ( $107 \mathrm{~mL}, 765 \mathrm{mmol}, 5$ equiv) A fter work-up, the crude material was purified by chromatography ( $\mathrm{SiO}_{2}, 1 \%$ to $5 \% \mathrm{MeOH}$ in $\left.\mathrm{CH}_{2} \mathrm{Cl}_{2}\right)$ to give $14(380 \mathrm{mg}, 82 \%)$ as a white solid: $\mathrm{R}_{\mathrm{f}}=040\left(\mathrm{SiO}_{2}\right.$, $\left.\mathrm{CH}_{2} \mathrm{Cl}_{2} / \mathrm{M} \mathrm{eOH} 95: 5\right) ; \mathrm{mp} 101-103^{\circ} \mathrm{C} ;[\alpha]_{\mathrm{D}}{ }^{20}+39\left(\mathrm{c} 10, \mathrm{CHCl}_{3}\right) ;{ }^{1} \mathrm{H}$ NMR $(400 \mathrm{MHz}$, $\left.\mathrm{CDCl}_{3}, \mathrm{gCOSY}\right) 147$ (m, 1H, H-6eq), 152 (m, 1H, H-5eq), 155 (m, 1H, H-4), 160 (m, 1H, H-6ax), 168 (m, 2H, H-4, H-8eq), 180 (tt, J = 135, 50 Hz, 1H, H-5ax), 187 (t, J $=125 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H}-8 \mathrm{ax}), 231(\mathrm{dm}, \mathrm{J}=105 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H}-4 \mathrm{a}) ; 252$ and $257(2 \mathrm{~d}, \mathrm{~J}=120$ $\mathrm{Hz}, 1 \mathrm{H}$ each, $\mathrm{H}-2), 315(\mathrm{dt}, \mathrm{J}=125,45 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H}-8 \mathrm{a}), 348$ and $371(2 \mathrm{~d}, \mathrm{~J}=130 \mathrm{~Hz}$, 1H each, $\mathrm{NCH}_{2} \mathrm{Ar}$ ), 384 (br s, 1H, H-3eq), 390-3 $95\left(\mathrm{~m}, 4 \mathrm{H}, \mathrm{OCH}_{2}\right), 7$ 20-7 $35(\mathrm{~m}, 5 \mathrm{H}$, ArH); ${ }^{13} \mathrm{CNMR}\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}, \mathrm{gHSQC}\right), 256$ (C-8), 266 (C-5), 298 (C-6), 288 (C-4a), 30-5 (C-4), 506 (C-2), $577(\mathrm{C}-8 \mathrm{a}), 585\left(\mathrm{NCH}_{2}\right), 641$ and $643\left(\mathrm{OCH}_{2}\right), 654$ (C-3), 1098 (C-7), 127 2, $1284,1287,1390$ (Ar) Anal calcd for $\mathrm{C}_{18} \mathrm{H}_{25} \mathrm{NO}_{3}: \mathrm{C}$ 71 26, H 8 31, N, 462 Found: C 70 89, H 8 50, N 444

Ring expansion of alcohol 9a. A solution of alcohol 9a ( $50 \mathrm{mg}, 016 \mathrm{mmol}$ ) in THF ( 1 mL ) was treated with $\mathrm{MsCl}\left(16 \mu \mathrm{~L}, 019 \mathrm{mmol}, 12\right.$ equiv) and $\mathrm{Et}_{3} \mathrm{~N}(90 \mu \mathrm{~L}, 064$ mmol, 4 equiv) under an argon atmosphere at - $20{ }^{\circ} \mathrm{C}$ for 1 h AgOAc was added ( 80 $\mathrm{mg}, 048 \mathrm{mmol}, 3$ equiv) and the resulting mixture was warmed to rt over a period of 1 $h$ The reaction mixture was filtered through a bed of Celite and diluted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ The organic layer was washed with saturated $\mathrm{NaHCO}_{3}(10 \mathrm{~mL})$, dried and concentrated
to give a mixture of acetates 9 c and 17 Purification and separation of the compounds was performed by chromatography $\left(\mathrm{SiO}_{2}, \mathrm{CH}_{2} \mathrm{Cl}_{2} / \mathrm{EtOAC} 9: 1\right)$ to afford $14 \mathrm{mg}(24 \%)$ of 9c (for analytical data, see Supporting Information) and 31 mg (54\%) of 17 (2S,3aS,7aS)-1-Benzyl-2-[(1'R)-(1-acetoxyethyl)]octahydroindol-6-one ethylene acetal (9c): Colourless oil $R_{f}=038\left(\mathrm{SiO}_{2}, \mathrm{CH}_{2} \mathrm{Cl}_{2} / \mathrm{EtOAc} 8: 2\right) ;[\alpha]_{\mathrm{D}}{ }^{20}-107$ (c 04 , $\mathrm{CHCl}_{3}$ ); IR $1736 \mathrm{~cm}^{-1}$; ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}, \mathrm{gCOSY}$ ) $123(\mathrm{~d}, \mathrm{~J}=64 \mathrm{~Hz}, 3 \mathrm{H}$, $\left.\mathrm{CH}_{3}\right), 126(\mathrm{~m}, 2 \mathrm{H}, \mathrm{H}-5), 144(\mathrm{~d}, \mathrm{~J}=84 \mathrm{~Hz}, 2 \mathrm{H}, \mathrm{H}-7), 160(\mathrm{~m}, 2 \mathrm{H}, \mathrm{H}-4), 176(\mathrm{~m}, 2 \mathrm{H}$, $\mathrm{H}-3$ ), 207 (s, 3H, OAc), 220 (m, 1H, H-3a), 290 ( $\mathrm{q}, \mathrm{J}=84 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H}-7 \mathrm{a}$ ), 291 (ddd, $\mathrm{J}=84,80,40 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H}-2), 363$ and $386\left(2 \mathrm{~d}, \mathrm{~J}=140 \mathrm{~Hz}, 1 \mathrm{H}\right.$ each, $\left.\mathrm{NCH}_{2} \mathrm{Ar}\right)$, 3 66-3 $83\left(\mathrm{~m}, 4 \mathrm{H}, \mathrm{OCH}_{2}\right), 505\left(\mathrm{qd}, \mathrm{J}=64,40 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H}-1^{\prime}\right), 7$ 20-7 $35(\mathrm{~m}, 5 \mathrm{H}, \mathrm{ArH})$; ${ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{M} \mathrm{Hz}, \mathrm{CDCl}_{3}$ ), see Table 1 HRFABM S: calcd for $\mathrm{C}_{21} \mathrm{H}_{30} \mathrm{NO}_{4} 3602175$ $\left(\mathrm{MH}^{+}\right)$, found 3602170
(2S,3R,4aS,8aS)-3-A cetoxy-1-benzyl-2-methyl-7-oxodecahydroquinoline ethylene acetal (17): Colourless oil $\mathrm{R}_{\mathrm{f}}=084\left(\mathrm{SiO}_{2}, \mathrm{CH}_{2} \mathrm{Cl}_{2} / \mathrm{EtOAC} 8: 2\right)[\alpha]_{\mathrm{D}}{ }^{20}-37$ (c 10 , $\mathrm{CHCl}_{3}$ ); IR $1734 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H} \operatorname{NMR}\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}, \mathrm{gCOSY}\right) 103(\mathrm{~d}, \mathrm{~J}=64 \mathrm{~Hz}, 3 \mathrm{H}$, Me), 145-1 74 (m, 7H, H-4, H-5, H-6, and H-8eq), $191(\mathrm{t}, \mathrm{J}=124 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H}-8 \mathrm{ax})$, $206(\mathrm{~s}, 3 \mathrm{H}, \mathrm{OAc}), 210(\mathrm{dm}, \mathrm{J}=120 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H}-4 \mathrm{a}), 284(\mathrm{dq}, \mathrm{J}=100,60 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H}-$ 2ax), 293 (dt, J = 12 4, $44 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H}-8 \mathrm{a}), 363$ and $390(2 \mathrm{~d}, \mathrm{~J}=148 \mathrm{~Hz}, 1 \mathrm{H}$ each, $\left.\mathrm{NCH}_{2} \mathrm{Ar}\right), 3$ 81-3 $94\left(\mathrm{~m}, 4 \mathrm{H}, \mathrm{OCH}_{2}\right), 460(\mathrm{td}, \mathrm{J}=110,52 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H}-3 \mathrm{ax}), 7$ 18-7 35 ( $\mathrm{m}, 5 \mathrm{H}, \mathrm{ArH}$ ); ${ }^{13} \mathrm{C}$ NMR ( $\left.100 \mathrm{MHz}, \mathrm{CDCl}_{3}, \mathrm{DEPT}, \mathrm{gHSQC}\right) 168(\mathrm{Me}), 213$ (OAC), 266 (C-5), 281 (C-8), 296 (C-4), 297 (C-6), 314 (C-4a), $526\left(\mathrm{NCH}_{2}\right), 530(\mathrm{C}-2)$, 558 (C-8a), 640 and $641\left(\mathrm{OCH}_{2}\right), 755(\mathrm{C}-3), 1098(\mathrm{C}-7), 1265,1277,1282,1412$ (Ar), 1706 (CO) HRFABM S: calcd for $\mathrm{C}_{21} \mathrm{H}_{30} \mathrm{NO}_{4} 3602175\left(\mathrm{M} \mathrm{H}^{+}\right)$, found 3602171

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Supporting Information Available Experimental and NMR data for all compounds reported, including Tables of ${ }^{13} \mathrm{C}$ NMR chemical shifts of octahydroindoles and decahydroquinolines reported Copies of ${ }^{1} \mathrm{H}$ and ${ }^{13} \mathrm{C}$ NMR spectra of all new compounds as well as COSY and HSQC spectra when available This material is available free of charge via the Internet at http://pubs acs org

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26 Decahydroquinoline 15 was isolated in one run working from alcohol 11a, although only in $5 \%$ yield For ${ }^{13} \mathrm{C}$ NMR data see Table 2 (Supporting Information)


15

27 The use of microwave conditions did not give satisfactory results either

28 Together with chloride 9b, the expanded product 16 was formed according to the ${ }^{13} \mathrm{C}$ NMR spectrum of the reaction mixture For NMR data of $9 b-12 b$, see Supporting Information


29 There are scarcely any examples of ring enlargement via aziridinium ions from secondary alcohols and they are always of the benzylic type: see references 9 b and 10b-c

30 It was confirmed in two series that the configuration at $\mathrm{C}-1$ ' of the side chain was identical to that of the starting alcohol Treatment of 10a and 11a with acetic anhydride gave the same acetates 10c and 11c as those obtained through sequential treatment with MsCl and $\mathrm{Et}_{3} \mathrm{~N}$, then AgOAc

31 The use of silver trifluoroacetate instead of silver acetate slightly increased the yield of the expanded compound, which after a basic work-up gave alcohol 21 (58\%), see Scheme 6

32 The use of tetrabutylammonium acetate did not improve the course of the reaction From 9a, a mixture of acetates 17 (38\%) and 9c (34\%) were isolated, whereas from 10a-12a more complex reaction mixtures were formed, the decahydroquinolines 18, 19, and 20 being obtained in a yield lower than 10\%
Ring Expansion of Functionalized Octahydroindoles
to Enantiopure cis-Decahydroquinolines
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Table 1. ${ }^{13} \mathrm{C}$ NMR Chemical shifts of octahydroindoles 3-12a

|  | 3 | 4 | 5 | 6 | 7 | 8 | 9 a | 10a | 11a |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| C2 | 66.2 | 62.9 | 66.2 | 61.7 | 74.9 | 70.8 | 71.0 | 72.5 | 66.1 |
| C3 | 33.6 | 31.7 | 32.1 | 30.5 | 33.3 | 30.6 | 30.6 | 32.2 | 25.4 |
| С3a | 35.7 | 35.2 | 35.0 | 35.6 | 35.6 | 35.6 | 34.6 | 34.8 | 35.6 |
| C4 | 24.0 | 23.3 | 24.3 | 22.8 | 24.2 | 22.8 | 23.9 | 22.7 | 22.9 |
| C5 | 31.0 | 29.9 | 31.0 | 28.9 | 31.0 | 29.2 | 27.5 | 29.3 | 28.9 |
| C6 | 109.1 | 109.3 | 108.9 | 109.4 | 108.8 | 109.2 | 108.9 | 109.0 | 109.4 |
| C7 | 37.5 | 31.9 | 37.8 | 30.6 | 38.2 | 30.9 | 38.2 | 38.8 | 30.4 |
| C7a | 62.5 | 59.5 | 63.0 | 59.3 | 63.0 | 59.4 | 62.9 | 63.3 | 58.9 |
| NCH2 | 58.8 | 53.4 | 58.8 | 51.7 | 60.0 | 53.4 | 58.9 | 63.2 | 51.4 |
| ips-Ar | 139.1 | 139.0 | 139.2 | 139.3 | 139.2 | 138.5 | 140.0 | 140.0 | 139.4 |
| o-Ar | 129.1 | 128.8 | 128.9 | 128.4 | 129.4 | 128.7 | 129.1 | 128.3 | 128.4 |
| $m-\mathrm{Ar}$ | 127.9 | 128.1 | 128.3 | 128.3 | 128.1 | 128.2 | 128.4 | 128.3 | 128.2 |
| $\mathrm{p}-\mathrm{Ar}$ | 126.8 | 126.9 | 127.2 | 127.0 | 127.1 | 127.1 | 127.2 | 127.0 | 127.0 |
| C-1' | 175.0 | 174.2 | 61.2 | 62.4 | 213.2 | 216.4 | 63.5 | 71.9 | 64.7 |
| Me | 51.6 | 51.6 | --- | --- | 24.9 | 24.9 | 18.2 | 20.8 | 18.2 |
| $\mathrm{OCH}_{2}$ | 64.3 | 64.3 | 64.1 | 64.3 | 64.0 | 64.3 | 64.0 | 64.1 | 64.1 |
|  | 64.0 | 64.0 | 63.9 | 63.9 | 64.0 | 63.9 | 64.0 | 63.9 | 63.8 |

${ }^{a}$ V alues for compounds 3, 5, 9a, 10a and 11a were assigned on the basis of gHSQC spectra.


$4 \mathrm{R}=\mathrm{CO}_{2} \mathrm{Me}$
$6 \mathrm{R}=\mathrm{CH}_{2} \mathrm{OH}$
$8 \mathrm{R}=\mathrm{COCH}_{3}$
9a $\mathrm{R}=(R)-\mathrm{CHOHCH}_{3}$
10a $\mathrm{R}=(S)-\mathrm{CHOHCH}_{3}$


11a $\mathrm{R}=(\mathrm{S})-\mathrm{CHOHCH}_{3}$

Table 1 continued). ${ }^{13}$ C NM R Chemical shifts of octahydroindoles 3-12 ${ }^{\text {b }}$

|  | $12 a$ | $9 b$ | $10 b$ | $11 b$ | $12 b$ | $9 c^{c}$ | $10 c^{c}$ | $11 c^{c}$ | $12 c^{c}$ |
| ---: | ---: | ---: | ---: | ---: | ---: | ---: | ---: | ---: | ---: |
| C2 | 66.5 | 72.1 | 71.6 | 66.7 | 66.4 | 69.1 | 68.8 | 64.4 | 62.9 |
| C3 | 30.3 | 29.7 | 30.4 | 28.2 | 27.1 | 30.1 | 31.0 | 27.7 | 27.4 |
| C3a | 35.3 | 34.5 | 34.5 | 34.6 | 35.0 | 34.4 | 34.6 | 34.8 | 35.2 |
| C4 | 22.7 | 23.0 | 23.3 | 22.8 | 22.8 | 23.3 | 23.2 | 22.9 | 22.8 |
| C5 | 29.0 | 29.5 | 29.9 | 29.7 | 29.0 | 29.8 | 29.7 | 29.1 | 29.0 |
| C6 | 109.4 | 109.2 | 109.1 | 109.5 | 109.4 | 109.2 | 109.1 | 109.5 | 109.5 |
| C7 | 31.2 | 39.2 | 39.1 | 31.0 | 30.9 | 39.1 | 39.3 | 30.2 | 30.6 |
| C7a | 58.8 | 62.5 | 63.7 | 58.9 | 58.3 | 62.2 | 63.8 | 58.2 | 58.4 |
| NCH2 | 53.9 | 61.4 | 61.3 | 52.2 | 52.7 | 60.4 | 61.8 | 52.0 | 52.5 |
| ips-Ar | 139.5 | 140.6 | 140.7 | 139.9 | 139.8 | 139.2 | 141.2 | 139.8 | 140.0 |
| 0-Ar | 128.2 | 128.3 | 128.6 | 128.2 | 128.2 | 129.4 | 128.2 | 128.1 | 128.1 |
| m-Ar | 127.8 | 128.1 | 128.2 | 128.2 | 128.1 | 128.1 | 128.0 | 128.0 | 128.0 |
| p-Ar | 126.7 | 126.6 | 126.8 | 126.7 | 126.8 | 127.1 | 126.5 | 126.6 | 126.6 |
| C-1' | 71.1 | 63.4 | 60.6 | 60.9 | 59.8 | 71.1 | 75.2 | 70.7 | 72.3 |
| Me | 20.6 | 22.9 | 19.8 | 22.3 | 17.7 | 17.1 | 16.5 | 16.7 | 14.5 |
| OCH2 | 64.2 | 64.1 | 64.0 | 64.2 | 64.3 | 63.9 | 64.0 | 64.1 | 64.2 |
|  | 63.8 | 63.8 | 63.9 | 63.9 | 63.9 | 63.8 | 63.8 | 63.8 | 63.9 |

${ }^{\mathrm{b}} \mathrm{V}$ alues for compounds $12 \mathrm{a}, 10 \mathrm{~b}, 11 \mathrm{~b}, 9 \mathrm{c}, 11 \mathrm{c}$, and 12 c were assigned on the basis of gHSQC spectra.
${ }^{c} 0 A c: 170.6 / 170.7$ and $21.3 / 21.5$.


9b $\mathrm{R}=\mathrm{Cl}\left(1^{\prime} R\right)$
10b $\mathrm{R}=\mathrm{Cl}\left(1^{\prime} S\right)$
9c $\mathrm{R}=\mathrm{OAc}\left(1^{\prime} R\right)$
10c $\mathrm{R}=\mathrm{OAc}\left(1^{\prime} S\right)$

$\begin{array}{ll}\text { 12a } & \mathrm{R}=\mathrm{OH}\left(1^{\prime} R\right) \\ \text { 11b } & \mathrm{R}=\mathrm{Cl}\left(1^{\prime} S\right) \\ \text { 12b } & \mathrm{R}=\mathrm{Cl}\left(1^{\prime} R\right) \\ \text { 11c } & \mathrm{R}=\mathrm{OAc}\left(1^{\prime} S\right) \\ \text { 12c } & \mathrm{R}=\mathrm{OAc}\left(1^{\prime} R\right)\end{array}$

Table 2. ${ }^{13} \mathrm{C}$ NM R Chemical shifts of decahydroquinolines $13-21^{\text {a }}$

|  | 13 | 14 | 15 | $16^{\text {b }}$ | 17 | 18 | $19^{\text {c }}$ | 21 |
| ---: | ---: | ---: | ---: | ---: | ---: | ---: | ---: | ---: |
| C2 | 52.0 | 50.6 | 57.1 | 56.0 | 53.0 | 51.9 | 53.2 | 55.7 |
| C3 | 68.1 | 65.4 | 69.7 | 61.4 | 75.5 | 73.2 | 73.1 | 73.6 |
| C4 | 33.1 | 30.5 | 26.8 | 33.1 | 29.6 | 27.0 | 24.2 | 33.3 |
| C4a | 32.8 | 28.8 | 28.3 | 35.5 | 31.4 | 33.6 | 28.4 | 32.0 |
| C5 | 26.9 | 26.6 | 26.4 | 26.7 | 26.6 | 25.0 | 26.8 | 26.9 |
| C6 | 29.9 | 29.8 | 30.2 | 32.4 | 29.7 | 30.2 | 30.0 | 29.9 |
| C7 | 109.9 | 109.8 | 109.3 | 109.6 | 109.8 | 109.4 | 109.3 | 109.9 |
| C8 | 27.0 | 25.6 | 33.5 | 28.9 | 28.1 | 34.1 | 34.7 | 27.6 |
| C8a | 57.0 | 57.7 | 56.6 | 56.2 | 55.8 | 56.1 | 56.9 | 56.5 |
| Me | $\ldots-$ | $\ldots--$ | 14.9 | 18.4 | 16.8 | 11.5 | 15.7 | 16.9 |
| NCH2 | 58.3 | 58.5 | 55.3 | 52.7 | 52.6 | 55.7 | 55.9 | 52.8 |
| Ar | 126.8 | 127.2 | 127.1 | 126.5 | 126.5 | 126.9 | 140.2 | 126.5 |
|  | 128.1 | 128.4 | 128.3 | 127.6 | 127.7 | 128.2 | 128.3 | 127.9 |
|  | 129.5 | 128.7 | 128.5 | 128.1 | 128.2 | 128.2 | 127.9 | 128.1 |
|  | 139.3 | 139.0 | n.0. | 141.1 | 141.2 | 139.8 | 126.6 | 140.6 |
| OCH 24.2 | 64.3 | 63.9 | 64.0 | 64.1 | 64.2 | 64.2 | 64.1 |  |
| Other |  |  |  |  |  |  | 170.6 | 170.4 |
|  |  |  |  | 170.7 |  |  |  |  |

${ }^{a} V$ alues for compounds $13,17,18,19$ and 21 were assigned on the basis of $g H S Q C$ spectra ${ }^{b} V$ alues taken from an NMR spectrum of a mixture of 9 b and $16 .{ }^{\mathrm{C}}$ Values taken from an NMR spectrum of a mixture of 11c and 19 .


13


14

$15 \mathrm{R}=\mathrm{H}$
$19 R=A c$

$16 \mathrm{R}=\mathrm{Cl}$
$17 R=O A c$
$21 \mathrm{R}=\mathrm{OH}$


18

## Experimental Section

General: All reactions were carried out under an argon atmosphere with dry, freshly distilled solvents under anhydrous conditions. A nalytical thin-layer chromatography was performed on $\mathrm{SiO}_{2}$ (silica gel $60 \mathrm{~F}_{254}$ ) or $\mathrm{Al}_{2} \mathrm{O}_{3}\left(\mathrm{ALOX} \mathrm{N}^{2} / \mathrm{UV}_{254}\right)$, and the products were located with iodoplatinate spray. Chromatography refers to flash chromatography and was carried out on $\mathrm{SiO}_{2}$ (silica gel 60, 230-240 mesh ASTM ) or $\mathrm{Al}_{2} \mathrm{O}_{3}$ (aluminium oxide 90). Drying of organic extracts was performed over anhydrous $\mathrm{Na}_{2} \mathrm{SO}_{4}$. Evaporation of solvent was accomplished with a rotatory evaporator. Chemical shifts of ${ }^{1} \mathrm{H}$ and ${ }^{13} \mathrm{C}$ NMR spectra are reported in ppm downfield ( $\delta$ ) from M $\mathrm{e}_{4} \mathrm{Si}$. Only noteworthy IR absorptions ( $\mathrm{cm}^{-1}$ ) are listed.

M ethyl $2 S, 3 a S, 7 a S)-1-B e n z y l-6-o x o o c t a h y d r o i n d o l e-2-c a r b o x y l a t e ~ e t h y l e n e ~ a c e t a l ~ 3) . ~ T o ~ a ~$ solution of ketone $1(3.28 \mathrm{~g}, 11 \mathrm{mmol})$ in toluene ( 350 mL ) were added a catalytic amount of TsOH and ethyleneglycol ( $1.84 \mathrm{~mL}, 33 \mathrm{mmol}$ ), and the reaction mixture was heated at reflux temperature for 4 h in a flask incorporating a Dean-Stark apparatus. The cooled solution was diluted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ and washed with aqueous saturated $\mathrm{NaHCO}_{3}(100 \mathrm{~mL})$. The organic phase was dried and concentrated to give 3.64 g of 3 as a yellowish oil, which was used in the next step without further purification. An analytical sample was obtained by chromatography ( $\mathrm{SiO}_{2}, 1 \% \mathrm{MeOH}$ in $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ ). $\mathrm{R}_{\mathrm{f}}=0.37\left(\mathrm{SiO}_{2}\right.$, hexane/EtOAc 3:2); $[\alpha]_{\mathrm{D}}{ }^{20}-41\left(\mathrm{c}=0.9, \mathrm{CHCl}_{3}\right) ;{ }^{1} \mathrm{H} \mathrm{NMR}\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right.$, gCOSY ) 1.46-1.58 (m, 1H), 1.64-1.85 (m, 5H ), 1.87-1.95 (m, 1H), 2.02-2.11 (m, 1H), 2.13-2.26 (m, 1H) , $2.99(\mathrm{q}, \mathrm{J}=7.5 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H}-7 \mathrm{a}), 3.41(\mathrm{dd}, \mathrm{J}=9.1,7.7 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H}-2), 3.54\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{OCH}_{3}\right)$, 3.74-3.92 (m, 6H), 7.18-7.40 (m, 5H, ArH); ${ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}, \mathrm{DEPT}$, gHSQC), see Table 1. A nal. Calcd for $\mathrm{C}_{19} \mathrm{H}_{25} \mathrm{NO}_{4}$ : C 68.86, H 7.60, N 4.22. Found: C 68.48, H 7.55, N 4.20. M ethyl 2S,3aR,7aR)-1-Benzyl-6-oxooctahydroindole-2-carboxylate ethylene acetal 4). Operating as above, from ketone $2(1.03 \mathrm{~g}, 3.6 \mathrm{mmol})$, acetal 4 was obtained $(1.12 \mathrm{~g})$ as yellowish crystals and used in the next step without further purification. A $n$ analytical sample was obtained by chromatography $\left(\mathrm{SiO}_{2}, 1 \% \mathrm{MeOH}\right.$ in $\left.\mathrm{CH}_{2} \mathrm{Cl}_{2}\right): \mathrm{R}_{\mathrm{f}}=0.24\left(\mathrm{SiO}_{2}\right.$, hexane/EtOAc 3:2); mp $64-66^{\circ} \mathrm{C}$;
$[\alpha]^{20}-53\left(\mathrm{c} 0.4, \mathrm{CHCl}_{3}\right) ;{ }^{1} \mathrm{H} \operatorname{NMR}\left(300 \mathrm{M} \mathrm{Hz}, \mathrm{CDCl}_{3}\right)$ 1.44-1.72(m,4H), 1.79-1.94 (m,3H), 2.12 ( $\mathrm{dt}, \mathrm{J}=12.9,10.7 \mathrm{~Hz}, 1 \mathrm{H}$ ), 2.46-2.58(m,1H), 3.37(dt, J = 10.5, 5.3 Hz, $1 \mathrm{H}, \mathrm{H}-7 \mathrm{a}$ ), $3.50(\mathrm{dd}, \mathrm{J}=$ $10.2,3.6 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H}-2), 3.55\left(\mathrm{~s}, 3 \mathrm{H}, 0 \mathrm{OH}_{3}\right), 3.74(\mathrm{~d}, \mathrm{~J}=13.2 \mathrm{~Hz}, 1 \mathrm{H}), 3.81(\mathrm{~d}, \mathrm{~J}=13.2 \mathrm{~Hz}, 1 \mathrm{H})$, 3.83-3.96(m, 4H), 7.18-7.40 (m,5H, ArH). ${ }^{13} \mathrm{C}$ NMR ( $75 \mathrm{M} \mathrm{Hz}, \mathrm{CDCl}_{3}$, DEPT), see Table 1. A nal. calcd for $\mathrm{C}_{19} \mathrm{H}_{25} \mathrm{NO}_{4}$ : C 68.86, H 7.60, N 4.22 . Found: C 68.56, H 7.85, N 4.24.

2S,3aS,7aS)-1-Benzyl-2-hydroxymethyl-6-oxooctahydroindole ethylene acetal 5). Ester 3 ( $603 \mathrm{mg}, 1.82 \mathrm{mmol}$ ) was dissolved in THF ( 9 mL ) and then cooled to $0{ }^{\circ} \mathrm{C}$. $\mathrm{LiBH}_{4}(2 \mathrm{M}$ in THF , $2.8 \mathrm{~mL}, 5.46 \mathrm{mmol}, 3$ equiv) was slowly added, and the reaction mixture was stirred at rt for 24 h . The reaction was quenched by adding $\mathrm{H}_{2} \mathrm{O}(5 \mathrm{~mL})$ and the organic layer was dried and concentrated to give a residue, which was purified by chromatography ( $\mathrm{SiO}_{2}, 1 \% \mathrm{MeOH}$ in $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ ) to afford $441 \mathrm{mg}\left(80 \%\right.$ from 1) of 5 as a colourless oil: $\mathrm{R}_{\mathrm{f}}=0.31\left(\mathrm{SiO}_{2}, \mathrm{CH}_{2} \mathrm{Cl}_{2} / \mathrm{MeOH} 95: 5\right) ;[\alpha]_{\mathrm{D}}{ }^{20}-15$ (C $0.2, \mathrm{CHCl}_{3}$ ); ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}, \mathrm{gCOSY}$ ) 1.49 (dddd, J $=12.0,5.0,5.0,1.2 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H}-$ 5eq), $1.65(\mathrm{~m}, 3 \mathrm{H}, \mathrm{H}-5$ and $\mathrm{H}-7), 1.75(\mathrm{~m}, 3 \mathrm{H}, \mathrm{H}-3$ and $\mathrm{H}-4), 1.85(\mathrm{dd}, \mathrm{J}=12.4,7.2 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H}-3)$, 2.20 ( $\mathrm{m}, 1 \mathrm{H}, \mathrm{H}-3 \mathrm{a}$ ), 2.40 (brs, $1 \mathrm{H}, \mathrm{OH}$ ), 2.98 (dddd, J = $8.0,7.5,5.4,1.8 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H}-2$ ), $3.02(\mathrm{q}, \mathrm{J}=$ $7.2 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H}-7 \mathrm{a}$ ), 3.33 (dd, J = 11.0, $1.2 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{CH}_{2} \mathrm{OH}$ ), 3.42 (dd, J = 11.0, 3.6 Hz, 1 H , $\left.\mathrm{CH}_{2} \mathrm{OH}\right), 3.69-3.88\left(\mathrm{~m}, 6 \mathrm{H}, \mathrm{OCH}_{2}\right.$ and $\left.\mathrm{NCH}_{2}\right), 7.26-7.32(\mathrm{~m}, 5 \mathrm{H}, \mathrm{ArH}) ;{ }^{13} \mathrm{CNMR}(100 \mathrm{MHz}$, $\mathrm{CDCl}_{3}, \mathrm{gHSQC}$ ), see Table 1. HRFA BM S calcd for $\mathrm{C}_{18} \mathrm{H}_{26} \mathrm{NO}_{3} 304.1906\left(\mathrm{M} \mathrm{H}^{+}\right)$, found 304.1913.

2S,3aR,7aR)-1-Benzyl-2-hydroxymethyl-6-oxooctahydroindole ethylene acetal 6). Operating as above from ester 4 ( $627 \mathrm{mg}, 1.89 \mathrm{mmol}$ ), alcohol 6 ( $464 \mathrm{mg}, 81 \%$ from 2) was obtained after chromatography $\left(\mathrm{SiO}_{2}, 1 \% \mathrm{MeOH}\right.$ in $\left.\mathrm{CH}_{2} \mathrm{Cl}_{2}\right)$, as white crystals: $\mathrm{R}_{\mathrm{f}}=0.40\left(\mathrm{SiO}_{2}\right.$, $\mathrm{CH}_{2} \mathrm{Cl}_{2} / \mathrm{MeOH} 95: 5$ ); mp $72-74{ }^{\circ} \mathrm{C}$; $[\alpha]_{\mathrm{D}}{ }^{20}-72$ (c $0.25, \mathrm{CHCl}_{3}$ ); ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{COSY}$, $\left.\mathrm{CDCl}_{3}\right) 1.46(\mathrm{t}, \mathrm{J}=12.0 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H}-7 \mathrm{ax}), 1.52(\mathrm{dq}, \mathrm{J}=12.5,2.5 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H}-4 \mathrm{eq}), 1.63(\mathrm{dm}, \mathrm{J}=12.0$ Hz, 1H, H-5eq), 1.64 (td, J = 10.0, 3.2 Hz, 1H, H-5ax), 1.78 (m, 1H, H-4ax), 1.80 (m, 1H, H-3ß), 1.83 (dd, J = 12.0, 5.5 Hz, 1H, H-7eq), $2.05(q, J=12.0 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H}-3 \alpha), 2.30(\mathrm{~m}, 1 \mathrm{H}, \mathrm{H}-3 \mathrm{a}), 2.99$ ( $\mathrm{dt}, \mathrm{J}=10.0,3.2 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H}-2$ ), 3.24 (ddd, J = 12.0, $5.5,5.5 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H}-7 \mathrm{a}$ ), $3.38(\mathrm{dm} \mathrm{J}=10.8 \mathrm{~Hz}$, $1 \mathrm{H}, \mathrm{CH}_{2} \mathrm{OH}$ ), $3.55\left(\mathrm{dd}, \mathrm{J}=10.8,3.2 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{CH}_{2} \mathrm{OH}\right), 3.64$ and $3.71(2 \mathrm{~d}, \mathrm{~J}=13.6 \mathrm{~Hz}, 1 \mathrm{H}$ each,
$\left.\mathrm{NCH}_{2}\right), 3.77-3.93\left(\mathrm{~m}, 4 \mathrm{H}, \mathrm{OCH}_{2}\right), 7.26-7.32(\mathrm{~m}, 5 \mathrm{H}, \mathrm{ArH}) ;{ }^{13} \mathrm{CNMR}\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}, \mathrm{DEPT}\right)$, see Table 1. A nal. Calcd for $\mathrm{C}_{18} \mathrm{H}_{25} \mathrm{NO}_{3}: \mathrm{C} 71.26, \mathrm{H} 8.31, \mathrm{~N} 4.62$. Found: C 71.05, H 8.20, N 4.57. 2S,3aS,7aS)-2-A cetyl-1-benzyloctahydroindol-6-one ethylene acetal 7). To a solution of ester $3(2.45 \mathrm{~g}, 7.4 \mathrm{mmol})$ in THF ( 140 mL ) cooled to $-20^{\circ} \mathrm{C}$ was added $\mathrm{Me}(\mathrm{M} \mathrm{eO}) \mathrm{NH} . \mathrm{HCl}(1.82 \mathrm{~g}$, $18.5 \mathrm{mmol}, 2.5 \mathrm{eq}$ ) and then over 30 min was added a solution of $\mathrm{i}-\mathrm{PrMgCl}$ in THF ( $18.5 \mathrm{~mL}, 2.0$ M , 5 equiv) maintaining the temperature at $-10^{\circ} \mathrm{C}$. The mixture was stirred for 40 min and quenched with saturated aqueous $\mathrm{NH}_{4} \mathrm{Cl}$ solution ( 50 mL ). The organic layer was dried and concentrated to afford 2.65 g of the corresponding Weinreb amide as a yellow oil, which was used without purification: $\mathrm{R}_{\mathrm{f}}=0.21\left(\mathrm{SiO}_{2}, \mathrm{CH}_{2} \mathrm{Cl}_{2} / \mathrm{MeOH} 96: 4\right)$. To a solution of the aforementioned Weinreb amide ( $2.65 \mathrm{~g}, 7.4 \mathrm{mmol}$ ) in THF ( 95 mL ) cooled to $0{ }^{\circ} \mathrm{C}$ was added dropwise M eM gBr in $\mathrm{Et}_{2} \mathrm{O}(6.4 \mathrm{~mL}, 3 \mathrm{M}, 19.24 \mathrm{mmol}, 2.6 \mathrm{eq})$. The reaction mixture was stirred at $0{ }^{\circ} \mathrm{C}$ for 1 h and quenched with saturated aqueous $\mathrm{NH}_{4} \mathrm{Cl}$ solution ( 50 mL ). The organic layer was dried and concentrated to give ketone $7(2.32 \mathrm{~g})$ as an oil, which was used without further purification. $\mathrm{R}_{\mathrm{f}}=$ $0.52\left(\mathrm{SiO}_{2}, \mathrm{CH}_{2} \mathrm{Cl}_{2} / \mathrm{M} \mathrm{eOH} 95: 5\right) ;{ }^{1} \mathrm{H} \mathrm{NMR}\left(300 \mathrm{M} \mathrm{Hz}, \mathrm{CDCl}_{3}\right) 1.53(\mathrm{~m}, 1 \mathrm{H}), 1.61-1.85(\mathrm{~m}, 6 \mathrm{H})$, $2.02(\mathrm{dt}, \mathrm{J}=12.3,7.2 \mathrm{~Hz}, 1 \mathrm{H}), 2.04\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{CH}_{3}\right), 2.25(\mathrm{~m}, 1 \mathrm{H}, \mathrm{H}-3 \mathrm{a}), 3.02(\mathrm{q}, \mathrm{J}=6.9 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H}-$ 7a), 3.29 ( $\mathrm{dd}, \mathrm{J}=9.3,7.8 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H}-2$ ), 3.63 and $3.83\left(2 \mathrm{~d}, \mathrm{~J}=13.5 \mathrm{~Hz}, 1 \mathrm{H}\right.$ each, $\mathrm{NCH}_{2} \mathrm{Ar}$ ), 3.75$3.95\left(\mathrm{~m}, 4 \mathrm{H}, \mathrm{OCH}_{2}\right), 7.20-7.35(\mathrm{~m}, 5 \mathrm{H}, \mathrm{Ar}) ;{ }^{13} \mathrm{C} \operatorname{NMR}\left(50 \mathrm{M} \mathrm{Hz}, \mathrm{CDCl}_{3}\right.$, DEPT$)$, see Table 1 .

2S,3aR,7aR)-2-A cetyl-1-benzyloctahydroindol-6-one ethylene acetal 8). The above procedure was applied to ester $4(1.12 \mathrm{~g})$ to afford the corresponding W einreb amide ( 1.30 g ): $\mathrm{R}_{\mathrm{f}}=$ $0.16\left(\mathrm{SiO}_{2}, \mathrm{CH}_{2} \mathrm{Cl}_{2} / \mathrm{MeOH} 95: 5\right)$. The W einreb amide was treated with $\mathrm{M} \mathrm{eM} \mathrm{gBrin} \mathrm{Et}_{2} \mathrm{O}$ ( 3.12 mL , $3 \mathrm{M}, 9.36 \mathrm{mmol}, 2.6 \mathrm{eq}$ ) and operating as in the formation of ketone $7,1.21 \mathrm{~g}$ of ketone 8 was isolated, which was used without further purification. $\mathrm{R}_{\mathrm{f}}=0.49\left(\mathrm{SiO}_{2}, \mathrm{CH}_{2} \mathrm{Cl}_{2} / \mathrm{MeOH} 95: 5\right) ;{ }^{1} \mathrm{H}$ NMR ( $300 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $1.40(\mathrm{t}, \mathrm{J}=12.0 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H}-5 \mathrm{ax}), 1.50-1.70(\mathrm{~m}, 3 \mathrm{H}), 1.75-1.90(\mathrm{~m}, 3 \mathrm{H})$, $2.07\left(\mathrm{~s}, 3 \mathrm{H}, \mathrm{CH}_{3}\right), 2.16(\mathrm{dt}, \mathrm{J}=13.5,11.4 \mathrm{~Hz}, 1 \mathrm{H}), 2.50(\mathrm{~m}, 1 \mathrm{H}, \mathrm{H}-3 \mathrm{a}), 3.36(\mathrm{ddd}, \mathrm{J}=11.4,5.7,5.7$ $\mathrm{Hz}, 1 \mathrm{H}, \mathrm{H}-7 \mathrm{a}), 3.35$ (dd, J = 11.0, $1.5 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H}-2$ ), 3.57 and $3.71(2 \mathrm{~d}, \mathrm{~J}=13.2 \mathrm{~Hz}, 1 \mathrm{H}$ each, $\left.\mathrm{CH}_{2} \mathrm{Ar}\right), 3.75-3.94\left(\mathrm{~m}, 4 \mathrm{H}, \mathrm{OCH}_{2}\right), 7.20-7.35(\mathrm{~m}, 5 \mathrm{H}, \mathrm{Ar}) ;{ }^{13} \mathrm{C}$ NMR ( $50 \mathrm{MHz}, \mathrm{CDCl}_{3}$, DEPT $)$, see Table 1.

Reduction of ketone 7. To a solution of amino ketone $7(2.29 \mathrm{~g}, 7.25 \mathrm{mmol})$ in $\mathrm{MeOH}(85 \mathrm{~mL})$ at $-20{ }^{\circ} \mathrm{C}$ was added $\mathrm{NaBH}_{4}$ ( $571 \mathrm{mg}, 14.5 \mathrm{mmol}$ ) in small portions. The resulting mixture was maintained at this temperature for 6 h . Then, water ( 25 mL ) was added and the mixture was extracted with $\mathrm{Et}_{2} \mathrm{O}$ ( $3 \times 50 \mathrm{~mL}$ ). The organic extracts were washed with brine, dried, and concentrated. Purification of the residue by chromatography $\left(\mathrm{SiO}_{2}\right.$, hexane to hexane/EtOAc 1:1) provided $1.05 \mathrm{~g}(46 \%)$ of alcohol 9 a as a colourless oil and $862 \mathrm{mg}(37 \%)$ of alcohol 10a as a colorless oil, after two succesive purifications. Overall yield for three steps ( $3 \rightarrow 9 \mathrm{a}+10 \mathrm{a}$ ): 83\%; 1.2:1 ratio of alcohols 9a:10a.

2S,3aS,7aS)-1-Benzyl-2-[ 1'R)- 1-hydroxyethyl)]octahydroindol-6-one ethylene acetal 9a): $R_{f}=0.30\left(\mathrm{SiO}_{2}, \mathrm{CH}_{2} \mathrm{Cl}_{2} / \mathrm{MeOH} 95: 5\right) ;[\alpha]_{\mathrm{D}}{ }^{20}-41\left(\mathrm{c} 1.3, \mathrm{CHCl}_{3}\right) ;{ }^{1} \mathrm{H} \mathrm{NMR}\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right.$, gCOSY ) $1.09\left(\mathrm{~d}, \mathrm{~J}=6.6 \mathrm{~Hz}, 3 \mathrm{H}, \mathrm{CH}_{3}\right), 1.43-1.81(\mathrm{~m}, 8 \mathrm{H}, \mathrm{H}-3, \mathrm{H}-4, \mathrm{H}-5$, and $\mathrm{H}-7), 2.20(\mathrm{~m}, 1 \mathrm{H}, \mathrm{H}-$ 3a), 2.80 (ddd, J = 9.3, 6.9, 3.3 Hz, 1H, H-2), 3.02 ( $\mathrm{q}, \mathrm{J}=7.2 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H}-7 \mathrm{a}$ ), 3.66-3.85 (m, 7H, H$1^{\prime}, \mathrm{NCH}_{2} \mathrm{Ar}$, and $\mathrm{OCH}_{2}$ ), 7.26-7.32 (m, 5H, ArH); ${ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{M} \mathrm{Hz}, \mathrm{CDCl}_{3}, \mathrm{gHSQC}$ ), see Table 1. HRFABMS: calcd for $\mathrm{C}_{19} \mathrm{H}_{28} \mathrm{NO}_{3} 318.2069\left(\mathrm{M} \mathrm{H}^{+}\right)$, found 318.2070 .

2S,3aS,7aS)-1-Benzyl-2-[ 1'S)- 1-hydroxyethyl)]octahydroindol-6-one ethylene acetal 10a): $\mathrm{R}_{\mathrm{f}}=0.30\left(\mathrm{SiO}_{2}, \mathrm{CH}_{2} \mathrm{Cl}_{2} / \mathrm{MeOH} 95: 5\right) ;[\alpha]_{\mathrm{D}}{ }^{20}-32\left(\mathrm{c} \mathrm{0.8}, \mathrm{CHCl}_{3}\right) ;{ }^{1} \mathrm{H} \mathrm{NMR}(400 \mathrm{MHz}$, $\left.\mathrm{CDCl}_{3}, \mathrm{gCOSY}\right) 1.16\left(\mathrm{~d}, \mathrm{~J}=6.4 \mathrm{~Hz}, 3 \mathrm{H}, \mathrm{CH}_{3}\right), 1.50(\mathrm{~m}, 2 \mathrm{H}, \mathrm{H}-5$ and $\mathrm{H}-7), 1.60(\mathrm{~m}, 2 \mathrm{H}, \mathrm{H}-3$ and H-5), 1.63 (t, J = $12.0 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H}-7 \mathrm{ax}$ ), 1.77 (m, 2H, H-4), 1.88 (ddd, J = 12.0, 8.0, $7.0 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H}-$ 3), $2.34(\mathrm{~m}, 1 \mathrm{H}, \mathrm{H}-3 \mathrm{a}), 2.82(\mathrm{q}, \mathrm{J}=7.5 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H}-2), 2.97(\mathrm{dt}, \mathrm{J}=12.0,6.0 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H}-7 \mathrm{a}), 3.54$ (quint, J $\left.=6.2 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H}-\mathrm{l}^{\prime}\right), 3.74-3.86\left(\mathrm{~m}, 6 \mathrm{H}, \mathrm{OCH}_{2}, \mathrm{NCH}_{2} \mathrm{Ar}\right), 7.20-7.40(\mathrm{~m}, 5 \mathrm{H}, \mathrm{ArH}) ;{ }^{13} \mathrm{C}$ NM R ( $\left.100 \mathrm{M} \mathrm{Hz}, \mathrm{CDCl}_{3}, \mathrm{gHSQC}\right)$, see Table 1. HRFA BM S: calcd for $\mathrm{C}_{19} \mathrm{H}_{28} \mathrm{NO}_{3} 318.2069\left(\mathrm{M} \mathrm{H}^{+}\right)$, found 318.2074.

Reduction of ketone 8. The above procedure was followed using ketone 8 ( $1.06 \mathrm{~g}, 3.36 \mathrm{mmol}$ ). Purification by chromatography $\left(\mathrm{SiO}_{2}\right.$, hexane to hexane-EtOAc 1:1) afforded $622 \mathrm{mg}(54 \%)$ of alcohol 11a as a white solid and then 365 mg ( $32 \%$ ) of alcohol 12 a as a white solid. Overall yield for three steps $(4 \rightarrow 11 a+12 a)$ : 86\%; 1.7:1 ratio of alcohols 11a and 12a.

2S,3aR,7aR)-1-Benzyl-2-[ 1'S)- 1-hydroxyethyl)]octahydroindol-6-one ethylene acetal 11a): $\mathrm{R}_{\mathrm{f}}=0.22\left(\mathrm{SiO}_{2}, \mathrm{CH}_{2} \mathrm{Cl}_{2} / \mathrm{M} \mathrm{eOH} 98: 2\right) ; \mathrm{mp} 73-75^{\circ} \mathrm{C}$; $[\alpha]_{\mathrm{D}}{ }^{20}$-100 (c $\left.0.7, \mathrm{CHCl}_{3}\right) ;{ }^{1} \mathrm{H}$ NMR (400 M Hz, CDCl $3, ~ g C O S Y) 1.13\left(d, J=6.4 \mathrm{~Hz}, 3 \mathrm{H}, \mathrm{CH}_{3}\right), 1.41(\mathrm{t}, \mathrm{J}=12.0 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H}-7 \mathrm{ax}), 1.50$ $(\mathrm{dm}, \mathrm{J}=12.0 \mathrm{~Hz}, 1 \mathrm{H}), 1.60-1.68(\mathrm{~m}, 2 \mathrm{H}), 1.73-1.90(\mathrm{~m}, 4 \mathrm{H}), 2.21(\mathrm{~m}, 1 \mathrm{H}, \mathrm{H}-3 \mathrm{a}), 2.76(\mathrm{dq}, \mathrm{J}=$ $10.0,2.4 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H}-2), 3.21$ (ddd, J = 12.0, 5.5, 5.5 Hz, 1H, H-7a), 3.61 and 3.77 (2d, J = 14.0 Hz , 1H each, $\mathrm{NCH}_{2} \mathrm{Ar}$ ), 3.78-3.93 ( $\mathrm{m}, 5 \mathrm{H}, \mathrm{OCH}_{2}$ and $\mathrm{H}-1^{\prime}$ ), 7.20-7.40 (m,5H, ArH); ${ }^{13} \mathrm{C}$ NMR (100 $\mathrm{MHz}, \mathrm{CDCl}_{3}, \mathrm{gHSQC}$ ), see Table 1. HRFABMS: calcd for $\mathrm{C}_{19} \mathrm{H}_{28} \mathrm{NO}_{3} 318.2069\left(\mathrm{M} \mathrm{H}^{+}\right)$, found 318.2074.

2S,3aR ,7aR )-1-Benzyl-2-[ 1'R)- 1-hydroxyethyl)]octahydroindol-6-one ethylene acetal 12a): $\mathrm{R}_{\mathrm{f}}=0.11\left(\mathrm{SiO}_{2}, \mathrm{CH}_{2} \mathrm{Cl}_{2} / \mathrm{MeOH} 98: 2\right) ; \mathrm{mp} 91-93^{\circ} \mathrm{C} ;[\alpha]_{\mathrm{D}}{ }^{20}-36$ (c $\left.1.0, \mathrm{CHCl}_{3}\right) ;{ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}, ~ \mathrm{gCOSY}$ ) $1.22\left(\mathrm{~d}, \mathrm{~J}=6.6 \mathrm{~Hz}, 3 \mathrm{H}, \mathrm{CH}_{3}\right.$ ), $1.43(\mathrm{t}, \mathrm{J}=12.0 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H}-7 \mathrm{ax}), 1.50-$ 1.75 (m, 4H, H-3, H-4, H-5, H-7), $1.80(\mathrm{~m}, 2 \mathrm{H}, \mathrm{H}-4$ and $\mathrm{H}-5), 2.15(\mathrm{~m}, 1 \mathrm{H}, \mathrm{H}-3), 2.32(\mathrm{~m}, 1 \mathrm{H}, \mathrm{H}-$ 3a), 2.88 (ddd, J = 9.8, 4.8, 2.2 Hz, 1H, H-2), 3.22 (ddd, J = 11.0, 5.4, 5.4 Hz, 1H, H-7a), 3.69 (m, 1H, H-1'), 3.73-3.92 (m, 6H, NCH2 and $\mathrm{OCH}_{2}$ ), $3.96(\mathrm{brs}, 1 \mathrm{H}, \mathrm{OH}), 7.20-7.40(\mathrm{~m}, 5 \mathrm{H}, \mathrm{ArH}) ;{ }^{13} \mathrm{C}$ NM R ( $\left.100 \mathrm{M} \mathrm{Hz}, \mathrm{CDCl}_{3}, \mathrm{gHSQC}\right)$, see Table 1. HRFABM S: calcd for $\mathrm{C}_{19} \mathrm{H}_{28} \mathrm{NO}_{3} 318.2069\left(\mathrm{M} \mathrm{H}^{+}\right)$, found 318.2063.

Conversion of alcohols 9a-12a to their corresponding chlorides 9b-12b.
Compounds 9b-12b were prepared according to the following procedure: to a solution of alcohol (9a-12a, $50 \mathrm{mg}, 0.16 \mathrm{mmol})$ in THF ( 1 mL ) at $0{ }^{\circ} \mathrm{C}$ was added $\mathrm{M} \mathrm{sCl}(0.014 \mathrm{~mL}, 0.18 \mathrm{mmol}, 1.1$ equiv), followed by $E t_{3} N(0.09 \mathrm{~mL}, 0.64 \mathrm{mmol}, 4.0$ equiv). A fter 4 h at reflux, the reaction mixture was poured into an aqueous 2.5 M NaOH solution ( 1 mL ). A fter extraction with $\mathrm{CH}_{2} \mathrm{Cl}_{2}(3 \times 2 \mathrm{~mL})$, the organic phase was dried and concentrated to afford compounds 9b-12b, which were used without further purification.

2S,3aS,7aS)-1-Benzyl-2-[ 1'R)- 1-chloroethyl)]octahydroindol-6-one ethylene acetal 9b). This compound was obtained together with the expanded chloride $16 ;{ }^{1} \mathrm{H} \mathrm{NMR}\left(300 \mathrm{M} \mathrm{Hz}, \mathrm{CDCl}_{3}\right)$ $1.53\left(\mathrm{~d}, \mathrm{~J}=6.6 \mathrm{~Hz}, 3 \mathrm{H}, \mathrm{CH}_{3}\right), 1.40-2.10(\mathrm{~m}, 8 \mathrm{H}), 2.25(\mathrm{~m}, 1 \mathrm{H}), 2.85-3.00(\mathrm{~m}, 2 \mathrm{H}), 3.80-4.00(\mathrm{~m}$,

7H), 7.20-7.40 (m, 5H); ${ }^{13} \mathrm{C}$ NMR ( $75 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ), see Table 1. HRFABMS calcd for $\mathrm{C}_{19} \mathrm{H}_{27}{ }^{35} \mathrm{CINO}_{3} 336.1730\left(\mathrm{M} \mathrm{H}^{+}\right)$, found 336.1731.

2S,3aS,7aS)-1-Benzyl-2-[ 1'S)- 1-chloroethyl)]octahydroindol-6-one ethylene acetal 10b): yellow oil; ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}, \mathrm{gCOSY}$ ) $1.49\left(\mathrm{~d}, \mathrm{~J}=6.8 \mathrm{~Hz}, 3 \mathrm{H}, \mathrm{CH}_{3}\right), 1.55-1.65(\mathrm{~m}, 4 \mathrm{H}$, H-5 and H-7), 1.70-1.80 (m, 3H, H-3, H-4), 1.88 (ddd, J = 12.0, 8.0, 7.0 Hz, $1 \mathrm{H}, \mathrm{H}-3$ ), $2.25(\mathrm{~m}, 1 \mathrm{H}$, $\mathrm{H}-3 \mathrm{a}$ ), $2.94(\mathrm{dt}, \mathrm{J}=10.5,6.8 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H}-7 \mathrm{a}), 3.15(\mathrm{dt}, \mathrm{J}=9.2,6.4 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H}-2), 3.73$ and 3.87 ( $2 \mathrm{~d}, \mathrm{~J}$ $=13.5 \mathrm{~Hz}, 1 \mathrm{H}$ each, $\left.\mathrm{NCH}_{2} \mathrm{Ar}\right), 3.70-3.84\left(\mathrm{~m}, 4 \mathrm{H}, \mathrm{OCH}_{2}\right), 3.95\left(\mathrm{dq}, \mathrm{J}=11.5,6.4 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H}-1^{\prime}\right)$, 7.20-7.38 (m, 5H, ArH); ${ }^{13} \mathrm{C}$ NM R ( $100 \mathrm{M} \mathrm{Hz}, \mathrm{CDCl}_{3}, \mathrm{gHSQC}$ ), see Table 1). HRFABM S calcd for $\mathrm{C}_{19} \mathrm{H}_{27}{ }^{35} \mathrm{CINO}_{3} 336.1730\left(\mathrm{M} \mathrm{H}^{+}\right)$, found 336.1727.

2S,3aR ,7aR )-1-Benzyl-2-[ 1'S)- 1-chloroethyl)]octahydroindol-6-one ethylene acetal 11b): white solid; ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{M} \mathrm{Hz}, \mathrm{CDCl}_{3}, \mathrm{gCOSY}$ ) $1.39(\mathrm{t}, \mathrm{J}=12.0 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H}-7 \mathrm{ax}), 1.44(\mathrm{~d}, \mathrm{~J}=6.8$ $\mathrm{Hz}, 3 \mathrm{H}, \mathrm{CH}_{3}$ ) 1.50-1.65 (m, 4H, H-3 and H-4), 1.75-1.95 (m, 3H, H-7eq and H-5), $2.42(\mathrm{~m}, 1 \mathrm{H}, \mathrm{H}-$ 3a), 2.96 ( $\mathrm{dt}, \mathrm{J}=10.0,3.0 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H}-2$ ), 3.23 (ddd, J $=11.6,5.6,5.6 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H}-7 \mathrm{a}$ ), 3.72 and 3.84 ( $2 \mathrm{~d}, \mathrm{~J}=14 \mathrm{~Hz}, 1 \mathrm{H}$ each, $\mathrm{NCH}_{2}$ ) , 3.78-3.90 ( $\mathrm{m}, 4 \mathrm{H}, \mathrm{OCH}_{2}$ ), $4.05\left(\mathrm{qd}, \mathrm{J}=6.8,3.2 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H}-1^{\prime}\right)$, 7.20-7.40 (m, 5H, ArH); $\left.{ }^{13} \mathrm{C} \mathrm{NMR(100} \mathrm{M} \mathrm{Hz} \mathrm{CDCl} 3,, ~ g H S Q C\right)$, see Table 1. HRFABMS calcd for $\mathrm{C}_{19} \mathrm{H}_{27}{ }^{35} \mathrm{CINO}_{3} 336.1730\left(\mathrm{M} \mathrm{H}^{+}\right)$, found 336.1741.

2S,3aR ,7aR )-1-Benzyl-2-[ 1'R)- 1-chloroethyl)]octahydroindol-6-one ethylene acetal 12b): yellow oil; ${ }^{1} \mathrm{H} \mathrm{NMR}\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 1.48\left(\mathrm{~d}, \mathrm{~J}=6.6 \mathrm{~Hz}, \mathrm{CH}_{3}\right), 1.40-2.05(\mathrm{~m}, 8 \mathrm{H}), 2.25(\mathrm{~m}$, $1 \mathrm{H}), 3.20(\mathrm{~m}, 2 \mathrm{H}), 3.6-4.0(\mathrm{~m}, 7 \mathrm{H}), 7.20-7.35(\mathrm{~m}, 5 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR(75 M Hz, CDCl 3 ), see Table 1. Ring expansion of chlorides 9b-12b

M ethod A. A solution of the appropriate chloride derivative $9 \mathrm{~b}-12 \mathrm{~b}(0.16 \mathrm{mmol})$ in THF ( 1 mL ) was treated with AgOAc ( $0.48 \mathrm{mmol}, 3$ equiv) at reflux for 4 h . The reaction mixture was filtered through a bed of Celite and diluted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}$. The organic layer was washed with saturated $\mathrm{NaHCO}_{3}(10 \mathrm{~mL})$, dried and concentrated to afford the corresponding mixture of acetates $9 \mathrm{c}-12 \mathrm{c}$ and 17-20. (See Table 1 in the main paper for results in each series and below for the NM R data of formed acetates).

- From 9b, a mixture of acetates 9c and 17 was obtained ( $80 \%$ overall yield) in a 1:1.3 ratio according to the NM R spectrum. Purification and separation of the compounds was performed by chromatography $\left(\mathrm{SiO}_{2}, \mathrm{CH}_{2} \mathrm{Cl}_{2} / \mathrm{EtOAc} ; 9: 1\right)$.
- From 10b, a mixture of acetates 10 c and 18 was obtained ( $70 \%$ overall yield) in a 2.9:1 ratio according to the NMR spectrum. Purification and separation of the compounds was performed by chromatography ( $\mathrm{SiO}_{2}, \mathrm{CH}_{2} \mathrm{Cl}_{2} / \mathrm{EtOAc} ; 9: 1$ ).
- From 11b, a non-separable mixture of acetates 11 c and 19 was obtained ( $80 \%$ overall yield) in a 5.7:1 ratio according to the NMR spectrum and GC-MS analysis. Purification of the mixture of compounds was performed by chromatography $\left(\mathrm{SiO}_{2}, \mathrm{CH}_{2} \mathrm{Cl}_{2} / \mathrm{EtOA}\right.$ 9:1).
- From 12b, a non-separable mixture of acetates 12 c and 20 was formed ( $60 \%$ overall yield) in a $13: 1$ ratio according to the NMR spectrum. Purification of the mixture of compounds was performed by chromatography ( $\mathrm{SiO}_{2}, \mathrm{CH}_{2} \mathrm{Cl}_{2} / \mathrm{EtOAc} 9: 1$ ).

Ring expansion of alcohols 9a-12a
Method B. A solution of the appropriate alcohol derivative 9a-12a ( 0.16 mmol ) in THF ( 1 mL ) was treated with M sCl ( $0.19 \mathrm{mmol}, 1.2$ equiv) and $\mathrm{Et}_{3} \mathrm{~N}$ ( $0.64 \mathrm{mmol}, 4$ equiv) under an argon atmosphere at - $20{ }^{\circ} \mathrm{C}$ for 1 h . AgOAc ( $0.48 \mathrm{mmol}, 3$ equiv) was added and the resulting mixture was warmed to rt over a period of 1 h . The reaction mixture was filtered through a bed of Celite and diluted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}$. The organic layer was washed with a saturated aqueous $\mathrm{NaHCO}_{3}$ solution ( 10 mL ), dried and concentrated to afford the corresponding mixture of acetates.

From 9a. A mixture of acetates 9 c and 17 was obtained in a 1:2.2 ratio according the NMR spectrum in 78 \% yield. Purification and separation of the compounds was performed by chromatography $\left(\mathrm{SiO}_{2}, \mathrm{CH}_{2} \mathrm{Cl}_{2} / \mathrm{EtOAc} 9: 1\right)$. For analytical data of 9 c and 17 , see the main text and Tables 1 and 2.

From 10a. A mixture of acetates 10 c and 18 was obtained in a 1.1:1 ratio according the NMR spectrum in $66 \%$ yield. Purification and separation of the compounds was performed by chromatography $\left(\mathrm{SiO}_{2}, \mathrm{CH}_{2} \mathrm{Cl}_{2} / \mathrm{EtOAc} 9: 1\right)$.

2S,3aS,7aS)-1-Benzyl-2-[ 1'S)-1-acetoxyethyl)]octahydroindol-6-one ethylene acetal 10c): Colourless oil. $\mathrm{R}_{\mathrm{f}}=0.26\left(\mathrm{SiO}_{2}, \mathrm{CH}_{2} \mathrm{Cl}_{2} / \mathrm{EtOAc} 9: 1\right) .[\alpha]_{\mathrm{D}}{ }^{20}-57\left(\mathrm{c} 0.2, \mathrm{CHCl}_{3}\right) ;$ IR $1732 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}, \mathrm{gCOSY}$ ) $1.21\left(\mathrm{~d}, \mathrm{~J}=6.0 \mathrm{~Hz}, 3 \mathrm{H}, \mathrm{CH}_{3}\right), 1.50(\mathrm{t}, \mathrm{J}=11.0 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H}-5)$, 1.50-1.80 (m, 7H), $1.90(\mathrm{~s}, 3 \mathrm{H}, \mathrm{OAc}), 2.25(\mathrm{~m}, 1 \mathrm{H}, \mathrm{H}-3 \mathrm{a}), 2.85(\mathrm{dt}, \mathrm{J}=11.0,6.4 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H}-7 \mathrm{a})$, $3.00(\mathrm{dt}, \mathrm{J}=10.0,7.0 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H}-2), 3.70$ and $3.94\left(2 \mathrm{~d}, \mathrm{~J}=14.0 \mathrm{~Hz}, 1 \mathrm{H}\right.$ each, $\mathrm{NCH}_{2} \mathrm{Ar}$ ), 3.74-3.86 ( $\mathrm{m}, 4 \mathrm{H}, \mathrm{OCH}_{2}$ ) , 4.91 (quint, J $=6.2 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H}-1^{\prime}$ ), $7.20-7.35(\mathrm{~m}, 5 \mathrm{H}, \mathrm{ArH}) ;{ }^{13} \mathrm{CNMR}(100 \mathrm{MHz}$, $\mathrm{CDCl}_{3}$ ), see Table 1. HRFABM S: calcd for $\mathrm{C}_{21} \mathrm{H}_{30} \mathrm{NO}_{4} 360.2175\left(\mathrm{M} \mathrm{H}^{+}\right)$, found 360.2163 .

2R,3R,4aS,8aS)-3-A cetoxy-1-benzyl-2-methyl-7-oxodecahydroquinoline ethylene acetal 18): Colourless oil. $\mathrm{R}_{\mathrm{f}}=0.58\left(\mathrm{SiO}_{2}, \mathrm{CH}_{2} \mathrm{Cl}_{2} / \mathrm{EtOAc} 9: 1\right) .[\alpha]_{\mathrm{D}}{ }^{20}-24$ (c $1.0, \mathrm{CHCl}_{3}$ ); IR $1733 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{MHz}, \mathrm{CDCl}_{3}, \mathrm{gCOSY}$ ) $1.10\left(\mathrm{~d}, \mathrm{~J}=7.0 \mathrm{~Hz}, 3 \mathrm{H}, \mathrm{CH}_{3}\right), 1.48(\mathrm{~m}, 1 \mathrm{H}, \mathrm{H}-5), 1.52(\mathrm{~m}, 1 \mathrm{H}$, H-4), 1.59 ( $\mathrm{m}, 2 \mathrm{H}, \mathrm{H}-6$ ), $1.72(\mathrm{t}, \mathrm{J}=12.4 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H}-8 \mathrm{ax}), 1.80(\mathrm{~m}, 1 \mathrm{H}, \mathrm{H}-5), 1.87(\mathrm{brt}, \mathrm{J}=13.0$ $\mathrm{Hz}, 1 \mathrm{H}, \mathrm{H}-4 \mathrm{ax}), 1.94(\mathrm{dm}, \mathrm{J}=12.4 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H}-8 \mathrm{eq}), 2.00(\mathrm{~s}, 3 \mathrm{H}, \mathrm{OAc}), 2.08(\mathrm{~m}, 1 \mathrm{H}, \mathrm{H}-4 \mathrm{a}), 3.00$ ( $\mathrm{dt}, \mathrm{J}=12.8,4.4 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H}-8 \mathrm{a}$ ), $3.20(\mathrm{q}, \mathrm{J}=6.5 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H}-2) ; 3.78$ and $3.84(2 \mathrm{~d}, \mathrm{~J}=14.4 \mathrm{~Hz}, 1 \mathrm{H}$ each, $\mathrm{NCH}_{2} \mathrm{Ar}$ ), 3.86-3.95 ( $\mathrm{m}, 4 \mathrm{H}, \mathrm{OCH}_{2}$ ), $4.99(\mathrm{ddd}, \mathrm{J}=12.0,5.6,4.4 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H}-3), 7.20-7.35(\mathrm{~m}$, $5 \mathrm{H}, \mathrm{ArH}$ ); ${ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}, \mathrm{gHSQC}$ ), see Table 2 HRFABMS: calcd for $\mathrm{C}_{21} \mathrm{H}_{30} \mathrm{NO}_{4}$ $360.2175\left(\mathrm{M} \mathrm{H}^{+}\right)$, found 360.2171 .

From 11a. 11c and 19 were obtained in a 2.2:1 ratio according the NM R spectrum in $76 \%$ yield as a unseparable mixture. Purification of the mixture of compounds was performed by chromatography ( $\mathrm{SiO}_{2}, \mathrm{CH}_{2} \mathrm{Cl}_{2} / \mathrm{EtOAc} 9: 1$ ).

2S,3aR,7aR)-1-Benzyl-2-[ 1'S)- 1-acetoxyethyl)]octahydroindol-6-one ethylene acetal 11c): Colourless oil. $\mathrm{R}_{\mathrm{f}}=0.31\left(\mathrm{SiO}_{2}, \mathrm{CH}_{2} \mathrm{Cl}_{2}\right) ;{ }^{1} \mathrm{H} \mathrm{NMR}\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}, \mathrm{gCOSY}\right) 1.23(\mathrm{~d}, \mathrm{~J}=6.4 \mathrm{~Hz}$, $3 \mathrm{H}, \mathrm{CH}_{3}$ ), $1.35(\mathrm{t}, \mathrm{J}=12.4 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H}-7 \mathrm{ax}), 1.45-2.05(\mathrm{~m}, 7 \mathrm{H}), 2.07(\mathrm{~s}, 3 \mathrm{H}, \mathrm{OAc}), 2.25(\mathrm{~m}, 1 \mathrm{H}, \mathrm{H}-$ 3a), $2.89(\mathrm{dt}, \mathrm{J}=10.0,2.8 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H}-2), 3.08(\mathrm{dt}, \mathrm{J}=12.0,6.0 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H}-7 \mathrm{a}), 3.52$ and 3.88 (2d, J $=13.6 \mathrm{~Hz}, 1 \mathrm{H}$ each, $\mathrm{NCH}_{2}$ ), 3.80-4.00 ( $\mathrm{m}, 4 \mathrm{H}, \mathrm{OCH}_{2}$ ), $5.14\left(\mathrm{qd}, \mathrm{J}=6.4,2.4 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H}-\mathrm{I}^{\prime}\right), 7.15-$ 7.35 ( $\mathrm{m}, 5 \mathrm{H}, \mathrm{ArH}$ ); ${ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}, \mathrm{gHSQC}$ ), see Table 1. HRFABMS: calcd for $\mathrm{C}_{21} \mathrm{H}_{30} \mathrm{NO}_{4} 360.2175\left(\mathrm{M} \mathrm{H}^{+}\right)$, found 360.2158 .
$2 R, 3 R, 4 a R, 8 a R)-3-A$ cetoxy-1-benzyl-2-methyl-7-oxodecahydroquinoline ethylene acetal 19): Colourless oil. $\mathrm{R}_{\mathrm{f}}=0.31\left(\mathrm{SiO}_{2}, \mathrm{CH}_{2} \mathrm{Cl}_{2}\right) ;{ }^{1} \mathrm{H} \mathrm{NMR}\left(400 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 1.04(\mathrm{~d}, \mathrm{~J}=7.2 \mathrm{~Hz}$, $\left.3 \mathrm{H}, \mathrm{CH}_{3}\right), 1.45-2.05(\mathrm{~m}, 8 \mathrm{H}), 2.07(\mathrm{~s}, 3 \mathrm{H}, \mathrm{OAc}), 2.34(\mathrm{~m}, 1 \mathrm{H}, \mathrm{H}-4 \mathrm{a}), 2.89(\mathrm{qd}, \mathrm{J}=7.2,1.2 \mathrm{~Hz}, 1 \mathrm{H}$, $\mathrm{H}-2 \mathrm{ax}), 3.09(\mathrm{dt}, \mathrm{J}=11.5,4.4 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H}-8 \mathrm{a}), 3.68-3.80\left(\mathrm{~m}, 6 \mathrm{H}, \mathrm{NCH}_{2} \mathrm{Ar}\right.$ and $\left.\mathrm{OCH}_{2}\right), 4.82(\mathrm{q}, \mathrm{J}=$ $2.8 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H}-3 \mathrm{eq}), 7.15-7.30(\mathrm{~m}, 5 \mathrm{H}, \mathrm{ArH}) ;{ }^{13} \mathrm{C} \mathrm{NMR}\left(100 \mathrm{MHz}, \mathrm{CDCl}_{3}\right)$, see Table 2. HRFABM S: calcd for $\mathrm{C}_{21} \mathrm{H}_{30} \mathrm{NO}_{4} 360.2175\left(\mathrm{M} \mathrm{H}^{+}\right)$, found 360.2158 .

From 12a. Acetate 12c and traces of 20 were formed in a 13:1 ratio, according the NMR spectrum and GC-M S analysis, in $40 \%$ yield as a unseparable mixture. Purification of the mixture of compounds was performed by chromatography $\left(\mathrm{SiO}_{2}, \mathrm{CH}_{2} \mathrm{Cl}_{2} / \mathrm{EtOA} \mathrm{C} 9: 1\right)$.

2S,3aR,7aR)-1-Benzyl-2-[ 1'R)- 1-acetoxyethyl)]octahydroindol-6-one ethylene acetal 12c): White Solid. $\mathrm{R}_{\mathrm{f}}=0.31\left(\mathrm{SiO}_{2}, \mathrm{CH}_{2} \mathrm{Cl}_{2}\right) ;{ }^{1} \mathrm{H} \operatorname{NMR}\left(400 \mathrm{M} \mathrm{Hz}, \mathrm{CDCl}_{3}, \mathrm{gCOSY}\right) 1.24(\mathrm{~d}, \mathrm{~J}=6.4 \mathrm{~Hz}$, $3 \mathrm{H}, \mathrm{CH}_{3}$ ), $1.37(\mathrm{t}, \mathrm{J}=12.0 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H}-7 \mathrm{ax}), 1.49(\mathrm{dm}, \mathrm{J}=12.0 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H}-5 \mathrm{eq}), 1.60-1.70(\mathrm{~m}, 3 \mathrm{H}$, $\mathrm{H}-5$ and $\mathrm{H}-4), 1.75(\mathrm{~m}, 1 \mathrm{H}, \mathrm{H}-3), 1.80-1.90(\mathrm{~m}, 2 \mathrm{H}, \mathrm{H}-3$ and $\mathrm{H}-7 \mathrm{eq}), 1.96(\mathrm{~s}, 3 \mathrm{H}, \mathrm{OAc}), 2.28(\mathrm{~m}$, 1H, H-3a), 3.05 (ddd, J = 8.8, 6.4, $2.8 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H}-2$ ), 3.15 (ddd, J = 11.2, 5.6, $5.6 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H}-7 \mathrm{a}$ ), 3.69 and $3.88\left(2 \mathrm{~d}, \mathrm{~J}=14.4 \mathrm{~Hz}, 1 \mathrm{H}\right.$ each, $\left.\mathrm{NCH}_{2} \mathrm{Ar}\right), 3.73-3.90\left(\mathrm{~m}, 4 \mathrm{H}, \mathrm{OCH}_{2}\right), 4.95$ (quint, $\mathrm{J}=6.4$ $\left.\mathrm{Hz}, 1 \mathrm{H}, \mathrm{H}-1^{\prime}\right), 7.20-7.35(\mathrm{~m}, 5 \mathrm{H}, \mathrm{ArH}) ;{ }^{13} \mathrm{C}$ NMR ( $100 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ), see Table 1. HRFABMS: calcd for $\mathrm{C}_{21} \mathrm{H}_{30} \mathrm{NO}_{4} 360.2175\left(\mathrm{M} \mathrm{H}^{+}\right)$, found 360.2163 .

Ring expansion of octahydroindole 9a using silver trifluoroacetate. A solution of alcohol 9a $(62 \mathrm{mg}, 0.2 \mathrm{mmol})$ in THF ( 1.4 mL ) was treated with $\mathrm{M} \mathrm{sCl}(0.019 \mathrm{~mL}, 0.24 \mathrm{mmol}, 1.2$ equiv) and $\mathrm{Et}_{3} \mathrm{~N}$ ( $0.11 \mathrm{~mL}, 0.8 \mathrm{mmol}, 4$ equiv) under argon atmosphere at - $20{ }^{\circ} \mathrm{C}$ for $1 \mathrm{~h} . \mathrm{CF}_{3} \mathrm{CO}_{2} \mathrm{Ag}$ was added ( $221 \mathrm{mg}, 1 \mathrm{mmol}, 5$ equiv) and the resulting mixture was warmed to room temperature over a period of 1 h . The mixture was treated with $2.5 \mathrm{~N} \mathrm{NaOH}(1 \mathrm{~mL})$ and stirred for 3 h . The reaction mixture was filtered through a bed of Celite and diluted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}$. The organic layer was dried and concentrated to afford a mixture of 9 a and 21 , which was purified by chromatography $\left(\mathrm{Al}_{2} \mathrm{O}_{3}\right.$, hexane/EtOAc 9:1) to give $16 \mathrm{mg}(26 \%)$ of 9 a and 36 mg ( $58 \%$ ) of ( $2 \mathrm{~S}, 3 \mathrm{R}, 4 \mathrm{aS}, 8 \mathrm{aS}$ )-1-Benzyl-3-hydroxy-2-methyl-7-oxodecahydroquinoline ethylene acetal (21): Colourless oil. $\mathrm{R}_{\mathrm{f}}=0.10\left(\mathrm{Al}_{2} \mathrm{O}_{3}\right.$,

Hexane/EtOA c 8:2). $[\alpha]_{\mathrm{D}}{ }^{20}+1.5\left(\mathrm{c} 0.6, \mathrm{CHCl}_{3}\right) ;{ }^{1} \mathrm{H} N \mathrm{NR}\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}, \mathrm{COSY}\right) 1.17(\mathrm{~d}, \mathrm{~J}=$ $6.0 \mathrm{~Hz}, 3 \mathrm{H}, \mathrm{CH}_{3}$ ), 1.43-1.49 ( $\mathrm{m}, 2 \mathrm{H}, \mathrm{H}-5$ and $\mathrm{H}-6$ ), 1.55-1.74 ( $\mathrm{m}, 5 \mathrm{H}, \mathrm{H}-4, \mathrm{H}-5, \mathrm{H}-6$, and $\mathrm{H}-8 \mathrm{eq}$ ), $1.88(\mathrm{t}, \mathrm{J}=12.6 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H}-8 \mathrm{ax}), 2.04(\mathrm{~m}, 1 \mathrm{H}, \mathrm{H}-4 \mathrm{a}), 2.57(\mathrm{dq}, \mathrm{J}=9.0,6.0 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{H}-2 \mathrm{ax}), 2.93$ (dt, J = 12.6, 4.5 Hz, 1H, H-8a), 3.36 (td, J = 9.0, 7.2 Hz, 1H, H-3ax), 3.57 and $3.90(2 d, \mathrm{~J}=14.4$ Hz each, $\left.\mathrm{NCH}_{2} \mathrm{Ar}\right), 3.79-3.94\left(\mathrm{~m}, 4 \mathrm{H}, \mathrm{OCH}_{2}\right), 7.18-7.35(\mathrm{~m}, 5 \mathrm{H}, \mathrm{ArH}) ;{ }^{13} \mathrm{C} \mathrm{NMR}\left(75 \mathrm{M} \mathrm{Hz}, \mathrm{CDCl}_{3}\right.$, gHSQC), see Table 2. HRFA BM S: calcd for $\mathrm{C}_{19} \mathrm{H}_{28} \mathrm{NO}_{3} 318.2069\left(\mathrm{M} \mathrm{H}^{+}\right)$, found 318.2064.



































$204 x^{20}$









MTM 101




























## 6.3

Synthesis of enantiopure cis-decahydroquinolines from homotyramines by Birch reduction and aminocyclization

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## Graphical Abstract

Synthesis of enantiopure cis-decahydroquinolines from homotyramines by Birch reduction and aminocyclization M arisa M ena, N ativitat V alls, M ar B orregán and J osep B onjoch*


# Synthesis of enantiopure cis-decahydroquinolines from homotyramines by Birch reduction and aminocyclization 

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#### Abstract

Birch reduction of homotyramines with a syn- $\beta$-amino alcohol unit followed by acid treatment of formed dihydroanisole derivatives gives polysubstituted enantiopure cis-decahydroquinolines. The stereoselectivity of the process differs if the hydroxyl group is free or protected. The procedure allows the synthesis of 7 oxodecahydroquinolines embodying four stereogenic centers with the same relative configuration as that of lepadins F and G.


## 1. Introduction

The use of ( $\omega$-aminoalkyl)methoxybenzene derivatives (e.g. tyrosine and tyramine compounds) as starting materials to elaborate azabicyclic compounds through a Birch reduction followed by an intramolecular cyclization of the resulting amino-tethered cyclohexenone (Scheme 1) is well-precedented in the literature. Following this methodology, octahydroindoles, ${ }^{1,2}$ azaspiroundecanes, ${ }^{3} 6$-azabicyclo[3.2.1]octanes, ${ }^{4}$ 2azabicyclo[3.3.1]nonanes, ${ }^{5}$ and decahydroquinolines ${ }^{6}$ have been prepared, but, apart from of our work on the synthesis of enantiopure octahydroindoles, ${ }^{2}$ all described processes lead to racemic compounds.


Scheme 1. The Birch reduction-aminocyclization process leading to azabicyclic compounds

[^1]In this paper, we describe the synthesis of enantiopure polysubstituted decahydroquinolines from homotyramine precursors following the aforementioned Birch reduction/aminocyclization sequence. ${ }^{7}$ The interest of this work, aside from the studying the stereocontrol of the process, lies in the possible usefulness of the resulting compounds in the synthesis of lepadin alkaloids. These natural products are structurally characterized by the presence of a 2,3,5-trisubstituted cis-fused decahydroquinoline ring. The substitution pattern, which has a methyl group at $C(2)$, a hydroxyl group, free or protected, at $C(3)$, and a functionalized side chain at $C(5)$, shows a variety of stereochemical arrangements. ${ }^{8}$ Total enantioselective syntheses of lepadins $\mathrm{A},{ }^{9} \mathrm{~B},{ }^{9-11} \mathrm{C},{ }^{9}$ - , ${ }^{11}$ and $H,{ }^{11}$ as well as a formal route to rac-lepadin $B^{12}$ have been reported.


Lepadins F,G
(relative configuration)

* absolute configuration unknown

Scheme 2. Retrosynthetic approach to lepadin alkaloids

We focused our attention on the synthesis of cis-decahydroquinolines incorporating a methyl at $\mathrm{C}(2)$ and a hydroxyl at $\mathrm{C}(3)$, with an S configuration at both stereogenic centers, as occurs in lepadins A, B, and C. In lepadins F and G both substituents also have a cis relationship, although their absolute configuration is unknown (see Scheme 2). The strategies described for the construction of 3-hydroxy-2methyldecahydroquinolines involve the elaboration of a polyfunctionalized piperidine followed by carbocyclic ring closure through aldol processes ${ }^{9,10}$ or the construction of the piperidine ring from cyclohexanone derivatives either by an intramolecular enamine alkylation ${ }^{11}$ or using a xanthate-mediated radical cyclization. ${ }^{12}$ In our approach, we envisaged enantiopure anisole derivatives of type $\mathrm{I}(\mathrm{R}=\mathrm{H}$ or Me ) as potential intermediates for the aforementioned cis-decahydroquinolines, as they would bring about ring closure by forming the $\mathrm{N}-\mathrm{C}(8 \mathrm{a})$ bond. ${ }^{13}$

## RESULTS AND DISCUSSION

## Synthetic aspects

For the proposed studies of Birch reduction of homotyramines followed by an aminocyclization process to achieve cis-decahydroquinolines of interest in the lepadine field, $\alpha$-methyl- $\beta$-aminoalcohol I was required. The synthesis of syn $\alpha$-methyl- $\beta$-amino alcohols (II, Scheme 3) is well-precedented not only by the methodological studies of the reactivity of alanine derivatives but also by the presence of this structural motif in several natural products other than the aforementioned lepadins, such as various piperidine alkaloids ${ }^{14}$ (i.e. carpamic acid, azimic acid, julifloridine, and cassine inter alia). The most suitable procedures for syn amino alcohols of type II are the orgamometallic addition upon the Weinreb amide of $N, N$-dibenzylalanine ${ }^{15}$ followed by hydride reduction of the resulting $\alpha$-amino ketone ${ }^{16}$ or the organometallic addition upon the $N$-B oc-alaninal. ${ }^{17,18}$ To our knowledge, none of these versatile approaches have been used in reactions involving p-methoxybenzylmagnesium bromide, as was required in the present work. We decided to use the protocol involving the Weinreb amide of N,Ndibenzylalanine and, in addition, introduced a new approach based on the ring-opening of a suitable epoxide with a lithium reagent is introduced to achieve aminoalcohol I.


Scheme 3. Synthesis of enantiopure syn- $\alpha$-methyl- $\beta$-aminoalcohols

Coupling of either the $p$-methoxybenzylmagnesium bromide with Weinreb amide $1^{19,20}$ or the p-methoxyphenyllithium with the (R) isomer of [(S)-1'(dibenzylamino)ethyl Joxirane (3) ${ }^{21}$ in presence of $\mathrm{BF}_{3}$. $\mathrm{t}_{2} \mathrm{O}$ (Ganem's conditions) ${ }^{22,23}$ gave synthetic access to the required aminoalcohol 4a, a diastereoselective reduction of the initially formed $\beta$-amino ketone 2 being necessary in the former sequence (Scheme $4, *$ denotes that $10 \%$ of the epimer of $4 a^{24}$ was additionally isolated in this route, see experimental part). This sequence $(1 \rightarrow 4 \mathrm{a})$ seemed to result in some loss of enantiopurity ${ }^{25}$ as determined by optical rotations in comparison with the sample
obtained through enantiopure epoxide 3 . Since the goal was to study the course of the aminocyclization of the dihydroanisole derivatives, optimizing the described protocol to minimize any racemization was not pursued at this stage.


Scheme 4. Synthesis of cis-decahydroquinolines.
ebenzylation of 4a gave the primary amine 5 a , which was submitted to the Birch reduction conditions $\left(\mathrm{Li}_{\mathrm{L}} / \mathrm{NH}_{3}\right)$ to allow the formation of dihydroanisole 6a. This was treated with a 2 N HCl solution at $75{ }^{\circ} \mathrm{C}$, and the decahydroquinoline ring was formed after enol ether hydrolysis, double bond isomerization, and an intramolecular 1,4addition of the amino group across the cyclohexenone intermediate. The process is stereoselective, with the exlusive formation of cis isomers of the decahydroquinoline ring. olysubstituted decahydroquinolines $7 \mathrm{a}(43 \%)$ and 8 a ( $17 \%$ ) were isolated in a 2.5:1 ratio and a overall yield of $60 \%$ from the sequence $4 a \rightarrow 7 a+8 a$.

We then carried out the same sequence of reactions but starting from syn-amino ether 4b, which was obtained by 0 -methylation of aminoalcohol 4a (Scheme 5). In this series, the aminocyclization step starting from dihydroanisole 6 b gave a 1:2.3 mixture
of decahydroquinolines 7 b and 8b, which were only partially separated. However, when the reaction mixture was basified and treated with benzoyl chloride after aminocyclization, the corresponding amides 9b and 10b were isolated in 17\% and 39\% overall yield (four steps from 4b).


Scheme 5.

In the cyclization processes $(6 \rightarrow 7+8)$, both in series a $(3-\mathrm{OH})$ and series $b(3-\mathrm{OMe})$, the isolated decahydroquinolines showed a cis-fused relationship. The major compound of series a (i.e. 7a) showed the same pattern of absolute configuration in its four stereocentres as lepadins A-C, while that of series $b$ (i.e. 8b) matched the relative configuration of lepadins F and G, allowing them to be considered as advanced building blocks for elaborating the aforementioned alkaloids.

The stereoselective cis-perhydroquinoline formation through a 6-exo process agreed with the stereochemical outcome observed in related cyclizations, ${ }^{6}$ and with both the steric and electronic preference for a pseudo axial addition of the nucleophilic species to the cyclohexenone moiety. Interestingly, the configuration of the new methine carbons (i.e. $\mathrm{C}-4 \mathrm{a}$ and $\mathrm{C}-8 \mathrm{a}$ ) is controlled to some extent by the oxygenated function. Why does the decalin ring formation change diastereoselectivity if there is a free or protected
hydroxyl group?. Considering that the axial attack proceeds through a chair-like transition state, in the hydroxyl series perhaps a hydrogen bonding favours the formation of enone $A$ with respect to the epimeric enone $A$ ', which could be in equilibrium by means of a tautomeric process through their corresponding dienol ether. On the contrary in series $b$, in which the hydroxyl group is protected as methyl ether, the steric factors (a 1,3-diaxial relationship between the C3-OM e and C4a-C5 bonds) prevent to some extent the formation of epimer B, the formation of B' being favoured (Scheme 6). Thus, the ratio of cis decahydroquinoline with an S configuration at the two new stereogenic centers formed in the aminocylization to the diastereoisomers with an $R$ configuration was higher in compounds with a methoxy rather than hydroxyl substituent.




vs

B'

8b

Scheme 6.

NM R studies of decahydroquinolines 7-10 (series a and b)
The stereochemistry of the synthesized azabicyclic compounds was elucidated by 2 NM R spectra (COSY , HSQC). The N -inside (7a and 7b) and N -outside (8a and 8b) cisdecahydroquinoline isomers ${ }^{26}$ in the amino series are clearly differentiated by two NMR features: (i) the ${ }^{1} \mathrm{H}$ NMR chemical shift of $\mathrm{H}-2$, which appears more deshielded ( $\delta 3.1$ ) in the N -outside than in the N -inside derivatives ( $\delta 2.8$ ), due to the compression upon H 2 of the C8-C8a bond, which has a 1,3-cis relationship, on the N -outside derivatives; (ii) the ${ }^{13} \mathrm{C}$ chemical shift of $\mathrm{C}(2)$ is more upfielded ( $\sim 10 \mathrm{ppm}$ ) in compounds with the $N$ outside conformation than those with the N -inside conformation; moreover the signals given by the carbon atoms at $\mathrm{C}-4, \mathrm{C}-6$, and $\mathrm{C}-8$ also appear in a higher field in the N outside derivatives.

The key evidence for the conformational elucidation of 7a was found in the ${ }^{1} H N M R$ coupling pattern for the methylene protons at C-8, which appear as $\mathrm{dd}(\mathrm{J}=14.6$ and 5.4 Hz ). The relative configuration for methoxy derivative 7 b is the same as that observed in 7a and their NM R data follows the same pattern of chemical shifts. (Scheme 2). The absolute configuration of 7a was deduced by considering that: a) the coupling constants for $\mathrm{H}-2(\mathrm{qd}, \mathrm{J}=6.6,2 \mathrm{~Hz})$ and $\mathrm{H}-3(\mathrm{q}, \mathrm{J}=2.4 \mathrm{~Hz})$ determined their location and hence fixed the methyl at $\mathrm{C}(2)$ and the hydroxyl at $\mathrm{C}(3)$ to an equatorial and axial disposition, respectively; b) the multiplicity of $\mathrm{H}-8 \mathrm{a}(\mathrm{br} \mathrm{s}$ ) implied an equatorial relationship with respect to the cyclohexane ring, which discarded not only a trans junction of the decaline ring but also, taking into account the preferred conformation, implied an R configuration for $\mathrm{C}(8 \mathrm{a})$. For the major component in the methoxy series 8 b , the axial proton H8a is strongly coupled to one adjacent axial. Hence, its resonance signal appears as a deceptively simple doublet ( $\mathrm{J}=10.4 \mathrm{~Hz}$ ) of triplets ( $\mathrm{J}=4.8 \mathrm{~Hz}$ ) centered at $\delta 3.43$.

In summary, the twin chair conformation with the nitrogen axially substituting the carbocyclic ring is the lowest energy conformation for 7 a , whereas the twin chair conformation with the nitrogen equatorially substituting the carbocyclic ring is the lowest energy conformation for 8b.



7b
$N$-inside conformation



8b
$N$-outside conformation

Figure 2. referred conformation of decahydroquinolines 7 and 8.

Interestingly, the $N$-benzoyl derivatives 9a, prepared from amine 7a in quantitative yield, and 10b (Figure 3) showed a different preferred conformation to that of their precursors 7a and 8b, respectively, as has been observed in synthetic intermediates in lepadin synthesis ${ }^{9-11}$ when the amino group is converted to a carbamate or amide group.


9a


10b

Figure 3. referred conformation of decahydroquinolines 9 and 10 .

In summary, a new synthetic entry to enantiopure polysubstituted cisdecahydroquinolines has been reported. Since the observed stereoselectivity allows lepadin-type stereochemistries to be achieved, further studies using decahydroquinolines 9a and 10b as advanced synthetic intermediates are in progress with the aim of achieving lepadins A-C and F-G, respectively.

## 3. Experimental

3.1. General. All reactions were carried out under an argon atmosphere with dry, freshly distilled solvents under anhydrous conditions. A nalytical TLC was performed on $\mathrm{SiO}_{2}$ (silica gel $60 \mathrm{~F}_{254}, \mathrm{Merck}$ ) or $\mathrm{Al}_{2} \mathrm{O}_{3}$ ( $\mathrm{ALOX} \mathrm{N} / \mathrm{UV}_{254}$, olygram), and the spots were located with iodoplatinate reagent (compounds 1-8) or $1 \%$ aqueous $\mathrm{KMnO}_{4}$ (compounds 9 and 10). Chromatography refers to flash chromatography and was carried out on $\mathrm{SiO}_{2}$ (silica gel 60, S S, 230-240 mesh ASTM ) or $\mathrm{Al}_{2} \mathrm{O}_{3}$ (aluminium oxide 90, Merck). rying of organic extracts during workup of reactions was performed over anhydrous $\mathrm{Na}_{2} \mathrm{SO}_{4}$. Optical rotations were recorded with a erkin- Imer 241 polarimeter. ${ }^{1} \mathrm{H}$ and ${ }^{13} \mathrm{C}$ NM R spectra were recorded with a Varian Gemini 200 or 300, or a V arian M ercury 400 instrument. Chemical shifts are reported in ppm downfield ( $\delta$ ) from $\mathrm{Me}_{4} \mathrm{Si}$. All new compounds were determined to be $>95 \%$ pure by ${ }^{1} \mathrm{H}$ NMR spectroscopy.

### 3.1.1. (S)-(N,N-Dibenzyl)amino-N-methoxy-N-methylpropionamide (1). To a

 solution of benzyl (S)-2-(N,N-dibenzylamino)propionate ( $2.15 \mathrm{~g}, 6 \mathrm{mmol}$ ), which was prepared from L -alanine $\left(\mathrm{BnBr}, \mathrm{K}_{2} \mathrm{CO}_{3}, \mathrm{tOH}\right)$ by the previously reported procedure, ${ }^{28}$ and $\mathrm{HCl} . \mathrm{HN}(\mathrm{OM} \mathrm{e}) \mathrm{Me}(3.0 \mathrm{~g}, 30 \mathrm{mmol})$ in $\mathrm{THF}(90 \mathrm{~mL})$ at $-20{ }^{\circ} \mathrm{C}$, ${ }^{\mathrm{i}} \mathrm{rMgCl} 2 \mathrm{M}$ in THF ( $30 \mathrm{~mL}, 60 \mathrm{mmol}$ ) was added dropwise over a period of 30 min . The reaction mixture was stirred for 2 h at this temperature and then warmed to rt for $2.5 \mathrm{~h} . \mathrm{NH}_{4} \mathrm{Cl}$ ( 20 mL ) was added and the product was extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}(3 \times 20 \mathrm{~mL})$. The combined organic layer was dried and concentrated to an oil which contained 1 and BnOH . The latter was removed under vacuum to afford compound $1(1.94 \mathrm{~g})$, which was used without further purification. The ${ }^{1} H$ NMR data were identical to those previously reported. ${ }^{20} \mathrm{R}_{\mathrm{f}}=0.1\left(\mathrm{SiO}_{2}, 9: 1\right.$ hexane/ tOAc); $\left.{ }^{13} \mathrm{C} \mathrm{NMR(50MHz,C} \mathrm{\quad Cl}_{3}\right)$ $14.9\left(\mathrm{CH}_{3}\right), 54.4\left(\mathrm{CH}_{2}\right), 56.1(\mathrm{CH}), 60.1\left(\mathrm{CH}_{2}\right), 126.8(\mathrm{CH}), 127.4(\mathrm{CH}), 127.6(\mathrm{CH})$, $128.1(\mathrm{CH}), 128.3(\mathrm{CH}), 128.5(\mathrm{CH}), 139.8$ (C), 173.9 (C).3.1.2. (3S)-3-(N,N-Dibenzyl)amino-1-(4-methoxyphenyl)butan-2-one (2). To a solution of $1(1.81 \mathrm{~g}, 5.8 \mathrm{mmol})$ in THF ( 50 mL ) at $0{ }^{\circ} \mathrm{C}$, 2-methoxybenzylmagnesium chloride 0.25 M in THF ( $46 \mathrm{~mL}, 11.6 \mathrm{mmol}$ ) was added dropwise. The reaction mixture was stirred 1 h at $0{ }^{\circ} \mathrm{C}$ and then quenched with $\mathrm{NH}_{4} \mathrm{Cl}$. The organic layer was dried and concentrated to an oil, which was purified by chromatography $\left(\mathrm{SiO}_{2}, 9: 1\right.$ hexane/ tOAc) to give 2 as a colourless oil ( $2.17 \mathrm{~g}, 87 \%$ ): $\mathrm{R}_{\mathrm{f}}=0.5\left(\mathrm{SiO}_{2}, 9: 1\right.$ hexane/AcO t); $[\alpha]^{25} \quad-2.4$ (c $1.0, \mathrm{CHCl}_{3}$ ); IR (KBr) 1715, $1611 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H}$ NMR (300 $\mathrm{MHz}, \mathrm{C} \mathrm{Cl} 3) 1.15(\mathrm{~d}, \mathrm{~J}=6.6 \mathrm{~Hz}, 3 \mathrm{H}), 3.47(\mathrm{~d}, \mathrm{~J}=13.2 \mathrm{~Hz}, 2 \mathrm{H}), 3.52(\mathrm{q}, \mathrm{J}=6.6 \mathrm{~Hz}$, $1 \mathrm{H}), 3.72(\mathrm{~d}, \mathrm{~J}=15.0 \mathrm{~Hz}, 1 \mathrm{H}), 3.74(\mathrm{~d}, \mathrm{~J}=13.2 \mathrm{~Hz}, 2 \mathrm{H}), 3.76(\mathrm{~s}, 3 \mathrm{H}), 3.88(\mathrm{~d}, \mathrm{~J}=$ $15.3 \mathrm{~Hz}, 1 \mathrm{H}), 6.74(\mathrm{~d}, \mathrm{~J}=8.6 \mathrm{~Hz}, 2 \mathrm{H}), 6.84(\mathrm{~d}, \mathrm{~J}=8.6 \mathrm{~Hz}, 2 \mathrm{H}), 7.20-7.40(\mathrm{~m}, 10 \mathrm{H})$; ${ }^{13} \mathrm{C}$ NMR $\left(75 \mathrm{M} \mathrm{Hz}, \mathrm{C} \quad \mathrm{Cl}_{3}\right) 7.1\left(\mathrm{CH}_{3}\right), 45.3\left(\mathrm{CH}_{2}\right), 54.6\left(\mathrm{CH}_{2}\right), 55.2\left(\mathrm{CH}_{3}\right), 60.6(\mathrm{CH})$, 113.8 (CH), 126.5 (C), 127.2 (CH), 128.4 (CH), 128.9 (CH), 130.3 (CH), 139.2 (C), 158.3 (C). A nal. Calcd for $\mathrm{C}_{25} \mathrm{H}_{27} \mathrm{NO}_{2}$ : C, 80.40; H, 7.29; N, 3.75. Found: 80.00; H, 7.29; N, 3.67.
3.1.3. (2S,3S)-3-(N,N-Dibenzyl)amino-1-(4-methoxyphenyl)butan-2-ol (4a) Method A (from ketone 2). To a solution of $2(2.15 \mathrm{~g}, 5.75 \mathrm{mmol})$ in $\mathrm{MeOH}(68 \mathrm{~mL})$ at $-20{ }^{\circ} \mathrm{C} \mathrm{NaBH} 4$ ( $453 \mathrm{mg}, 11.5 \mathrm{mmol}$ ) was added. The reaction mixture was stirred for 1 h at this temperature, and then quenched with brine ( 30 mL ). The product was extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}(3 \times 30 \mathrm{~mL})$, dried and concentrated to give a mixture of
alcohols $4 a$ and epi- $4 a^{25}$ in a 9:1 ratio according to the NMR spectrum. urification by chromatography ( $\mathrm{SiO}_{2}, 9: 1$ hexane/ tOAc) gave $4 \mathrm{a}(1.94 \mathrm{~g}, 90 \%$ ) and epi-4a ( 216 mg , $10 \%)$.
4a: colourless oil. $\mathrm{R}_{\mathrm{f}}=0.24\left(\mathrm{SiO}_{2}, 9: 1\right.$ hexane/AcO t); $[\alpha]^{25}$-2.0 (c 0.4, $\left.\mathrm{CHCl}_{3}\right)$; IR (KBr) 3600-3100, $1611 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H} \operatorname{NMR}(300 \mathrm{MHz}, \mathrm{C} \mathrm{Cl} 3) 1.07(\mathrm{~d}, \mathrm{~J}=6.6 \mathrm{~Hz}, 3 \mathrm{H})$, $2.38(\mathrm{dd}, \mathrm{J}=14.0,7.8 \mathrm{~Hz}, 1 \mathrm{H}), 2.60(\mathrm{dq}, \mathrm{J}=9.3,6.6 \mathrm{~Hz}, 1 \mathrm{H}), 2.78\left(\mathrm{dd}, \mathrm{J}_{1}=14.3,3.0\right.$ $\mathrm{Hz}, 1 \mathrm{H}), 3.30(\mathrm{~d}, \mathrm{~J}=13.5 \mathrm{~Hz}, 2 \mathrm{H}), 3.68(\mathrm{~m}, 1 \mathrm{H}), 3.77(\mathrm{~s}, 3 \mathrm{H}), 3.82(\mathrm{~d}, \mathrm{~J}=13.5 \mathrm{~Hz}$, $2 \mathrm{H}), 6.77(\mathrm{~d}, \mathrm{~J}=9 \mathrm{~Hz}, 2 \mathrm{H}), 7.11(\mathrm{~d}, \mathrm{~J}=9 \mathrm{~Hz}, 2 \mathrm{H}), 7.20-7.40(\mathrm{~m}, 10 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR(75 $\mathrm{MHz}, \mathrm{C} \mathrm{Cl} 3) 8.3\left(\mathrm{CH}_{3}\right), 39.1\left(\mathrm{CH}_{2}\right), 53.2\left(\mathrm{CH}_{2}\right), 55.2\left(\mathrm{CH}_{3}\right), 57.7(\mathrm{CH}), 71.9(\mathrm{CH})$, 113.5 (CH), 127.1 (CH), 128.4 (CH), 128.9 (CH), 130.1 (CH), 131.0 (C), 138.7 (C), 157.8 (C). A nal. Calcd for $\mathrm{C}_{25} \mathrm{H}_{29} \mathrm{NO}_{2} .1 / 2 \mathrm{H}_{2} \mathrm{O}: \mathrm{C}, 78.12 ; \mathrm{H}, 7.81 ; \mathrm{N}, 3.64$. Found : C, 77.84; H, 8.16; N, 3.34.
epi-4a: colourless oil. $\mathrm{R}_{\mathrm{f}}=0.14\left(\mathrm{SiO}_{2}\right.$, 9:1 hexane/AcO t); IR (KBr) 3600-3100, 1611 $\mathrm{cm}^{-1}{ }^{1}{ }^{1} \mathrm{H} N \mathrm{NR}\left(300 \mathrm{MHz}, \mathrm{C} \quad \mathrm{Cl}_{3}\right) 1.17(\mathrm{~d}, \mathrm{~J}=6.6 \mathrm{~Hz}, 3 \mathrm{H}), 2.30(\mathrm{dd}, \mathrm{J}=13.8,9.6 \mathrm{~Hz}$, 1 H ), 2.75 (quint, J $=6.9 \mathrm{~Hz}, 1 \mathrm{H}$ ), 3.21 (dd, J $=13.8,3.0 \mathrm{~Hz}, 1 \mathrm{H}$ ), $3.50(\mathrm{~J}=13.8 \mathrm{~Hz}$, $2 H$ ), 3.73-3.81 (m, 1H), 3.80 (d, J = $14.1 \mathrm{~Hz}, 2 \mathrm{H}), 6.80(\mathrm{~d}, \mathrm{~J}=9.0 \mathrm{~Hz}, 2 \mathrm{H}), 7.02(\mathrm{~d}, \mathrm{~J}=$ $9.0 \mathrm{~Hz}, 2 \mathrm{H}), 7.20-7.40(\mathrm{~m}, 10 \mathrm{H}) ;{ }^{13} \mathrm{C} \operatorname{NMR}\left(75 \mathrm{M} \mathrm{Hz}, \mathrm{C} \mathrm{Cl}_{3}\right) 8.6\left(\mathrm{CH}_{3}\right), 40.6\left(\mathrm{CH}_{2}\right)$, $54.7\left(\mathrm{CH}_{2}\right), 55.3\left(\mathrm{CH}_{3}\right), 57.2(\mathrm{CH}), 74.7(\mathrm{CH}), 113.9(\mathrm{CH}), 126.8(\mathrm{CH}), 128.2(\mathrm{CH})$, 128.8 (CH), 130.2 (CH), 131.0 (C), 140.0 (C), 158.1 (C).

Method $B$ (from epoxide 3). To a solution of $n-B u L i(1.6 ~ M ~ i n ~ h e x a n e s, ~ 1.05 ~ m L, ~ 1.68 ~$ mmol ) in THF ( 3.5 mL ) at $-78{ }^{\circ} \mathrm{C}$ was added 4-bromoanisole ( $\left.0.2 \mathrm{~mL}, 1.56 \mathrm{mmol}\right)$. The reaction mixture was stirred for 90 min , treated with a solution of $(2 R)-\left[1^{\prime}(S)\right.$ (dibenzylamino)ethyl Joxirane ${ }^{21}(162 \mathrm{mg}, 0.6 \mathrm{mmol})$ in THF ( 2 mL ) and $\mathrm{BF}_{3}$. $\mathrm{t}_{2} \mathrm{O}(0.21$ $\mathrm{mL}, 1.68 \mathrm{mmol})$, and continuously stirred at $-78{ }^{\circ} \mathrm{C}$ for 2 h prior to being quenched with saturated $\mathrm{NH}_{4} \mathrm{Cl}(4 \mathrm{~mL})$ and warmed to rt . The product was extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ (3 X 10 mL ), and the organic layer was dried and concentrated to give an oil, which was purified by chromatography ( $\mathrm{SiO}_{2}, 9: 1$ hexane/AcO t) to give 4a as a colourless oil ( $153 \mathrm{mg}, 69 \%$ ). The spectroscopic data were identical with the product obtained by method A, but its rotatory power was higher: $[\alpha]^{25}=-5.6\left(c 1.4, \mathrm{CHCl}_{3}\right)$.
3.1.4. (2S,3S)-N,N-Dibenzyl-3-methoxy-4-(4-methoxyphenyl)-2-butanamine (4b). To a suspension of $\mathrm{NaH}(195 \mathrm{mg}, 4.89 \mathrm{mmol})$ in dry THF ( 2 mL ) at $0{ }^{\circ} \mathrm{C}$ under argon atmosphere a solution of $4 \mathrm{a}(1.22 \mathrm{~g}, 3.26 \mathrm{mmol})$ in dry THF was transferred ( $1 \mathrm{~mL}+1$
$\mathrm{mL})$. The reaction mixture was warmed over 20 min to rt and then $\mathrm{Mel}(2 \mathrm{~mL}, 32.6$ mmol ) was added. The reaction was sealed and stirred for $48 \mathrm{~h} . \mathrm{NH}_{4} \mathrm{Cl}$ was added (10 $\mathrm{mL})$ and the product was extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}(3 \times 20 \mathrm{~mL})$. The resulting organic layer was washed with $\mathrm{H}_{2} \mathrm{O}(15 \mathrm{~mL})$, brine ( 15 mL ), dried, and concentrated to give an oil, which was purified by chromatography ( $\mathrm{SiO}_{2}, 9: 1$ hexane/ AcO t) to give 4 b as a colourless oil. ( $1.14 \mathrm{~g}, 90 \%$ ): $\mathrm{R}_{\mathrm{f}}=0.42\left(\mathrm{SiO}_{2}, 9: 1\right.$ hexane/AcO t); $[\alpha]^{25}-4.6$ (c 1.0, $\mathrm{CHCl}_{3}$ ); IR (KBr) $1611 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H} \operatorname{NMR}\left(300 \mathrm{MHz}, \mathrm{C} \quad \mathrm{Cl}_{3}\right) 1.13(\mathrm{~d}, \mathrm{~J}=6.9 \mathrm{~Hz}, 3 \mathrm{H})$, 2.74-2.88 (m, 3H), 3.13 (s, 3H), 3.22 (dt, J = 7.2, 4.8 Hz, 1H), 3.43 (d, J = 13.5 Hz , $2 \mathrm{H}), 3.77(\mathrm{~s}, 3 \mathrm{H}), 4.01(\mathrm{~d}, \mathrm{~J}=13.5 \mathrm{~Hz}, 2 \mathrm{H}), 6.73(\mathrm{~d}, \mathrm{~J}=8.7 \mathrm{~Hz}, 2 \mathrm{H}), 6.94(\mathrm{~d}, \mathrm{~J}=8.7$ $\mathrm{Hz}, 2 \mathrm{H}), 7.20-7.32(\mathrm{~m}, 6 \mathrm{H}), 7.39-7.41(\mathrm{~m}, 4 \mathrm{H}) ;{ }^{13} \mathrm{C} \mathrm{NMR}\left(75 \mathrm{MHz}, \mathrm{C} \quad \mathrm{Cl}_{3}\right) 10.1$ $\left(\mathrm{CH}_{3}\right), 37.4\left(\mathrm{CH}_{2}\right), 55.1(\mathrm{CH}), 55.2\left(\mathrm{CH}_{2}\right), 55.2\left(\mathrm{CH}_{3}\right), 59.2\left(\mathrm{CH}_{3}\right), 88.1(\mathrm{CH}), 113.5$ (CH), 126.6 (CH), 128.1 (CH), 128.9 (CH), 130.2 (CH), 132.3 (C), 140.9 (C), 157.7 (C). A nal. Calcd for $\mathrm{C}_{26} \mathrm{H}_{31} \mathrm{NO}_{2}$ : C, 80.17; H, 8.02; N, 3.60. Found: C, 80.07; H, 8.31; N, 3.40.
3.1.5. (2S,3S)-3-A mino-1-(4-methoxyphenyl)butan-2-ol (5a). A suspension of 4 a $(2.16 \mathrm{~g}, 5.75 \mathrm{mmol})$ and $\mathrm{d}(\mathrm{OH})_{2} / \mathrm{C}(20 \%, 210 \mathrm{mg})$ in $\mathrm{tOH}(110 \mathrm{~mL})$ was stirred at rt under hydrogen atmosphere overnight. The catalyst was removed by filtration through Celite and the filtrate was concentrated to give 5 a as an oil ( 1.12 g ), which was used directly in the next step. A $n$ analytical sample was obtained by chromatography $\left(\mathrm{Al}_{2} \mathrm{O}_{3}\right.$, $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ saturated with $\mathrm{NH}_{3}$ ); $[\alpha]^{25}$-15.5 (c 0.7, $\mathrm{CHCl}_{3}$ ); ${ }^{1} \mathrm{H}$ NMR ( 300 MHz , $\left.\mathrm{C} \mathrm{Cl}_{3}\right) 1.12(\mathrm{~d}, \mathrm{~J}=6.6 \mathrm{~Hz}, 3 \mathrm{H}), 2.56(\mathrm{dd}, \mathrm{J}=14.0,8.6 \mathrm{~Hz}, 1 \mathrm{H}), 2.74-2.86(\mathrm{~m}, 2 \mathrm{H})$, $3.40-3.46(\mathrm{~m}, 1 \mathrm{H}), 3.78(\mathrm{~s}, 3 \mathrm{H}), 6.84(\mathrm{~d}, \mathrm{~J}=8.7 \mathrm{~Hz}, 2 \mathrm{H}), 7.14(\mathrm{~d}, \mathrm{~J}=8.7 \mathrm{~Hz}, 2 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR $(75 \mathrm{MHz}, \mathrm{C} \mathrm{Cl} 3) 20.5\left(\mathrm{CH}_{3}\right), 39.6\left(\mathrm{CH}_{2}\right), 50.3(\mathrm{CH}), 54.7\left(\mathrm{CH}_{2}\right), 55.2\left(\mathrm{CH}_{3}\right)$, 76.4 (CH), 113.7 (CH), 130.1 (C), 130.2 (CH), 158.0 (C).
3.1.6. (2S,3S)-3-M ethoxy-4-(4-methoxyphenyl)-2-butanamine (5b). Operating as above, starting from $4 \mathrm{~b}(1.14 \mathrm{~g}, 2.92 \mathrm{mmol}), 5 \mathrm{~b}$ was obtained ( 615 mg ) as an oil which was used directly in the next step; ${ }^{1} \mathrm{H}$ NMR ( $200 \mathrm{MHz}, \mathrm{C} \quad \mathrm{Cl}_{3}$ ) 1.11 ( $\mathrm{d}, \mathrm{J}=6.3 \mathrm{~Hz}$, $3 \mathrm{H}), 2.56(\mathrm{dd}, \mathrm{J}=14.3,6.5 \mathrm{~Hz}, 1 \mathrm{H}), 2.90-2.81(\mathrm{~m}, 2 \mathrm{H}), 3.07, \mathrm{dt}, \mathrm{J}=6.6,5.4 \mathrm{~Hz}, 1 \mathrm{H})$, $3.30(\mathrm{~s}, 3 \mathrm{H}), 3.79(\mathrm{~s}, 3 \mathrm{H}), 6.84(\mathrm{~d}, \mathrm{~J}=8.4 \mathrm{~Hz}, 2 \mathrm{H}), 7.15(\mathrm{~d}, \mathrm{~J}=8.4 \mathrm{~Hz}, 2 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( $50 \mathrm{MHz}, \mathrm{C} \mathrm{Cl} 3) 20.1\left(\mathrm{CH}_{3}\right), 35.8\left(\mathrm{CH}_{2}\right), 49.2(\mathrm{CH}), 55.2\left(\mathrm{CH}_{3}\right), 58.8\left(\mathrm{CH}_{3}\right), 87.6$ (CH), 113.7 (CH), 130.4 (CH), 130.8 (C), 158.0 (C).
3.1.7. (2S,3S)-3-A mino-1-(4-methoxy-2,5-dihydrophenyl)butan-2-ol (6a). To a solution of $5 \mathrm{a}(1.12 \mathrm{~g}, 5.75 \mathrm{mmol})$ in $\mathrm{tOH}(6 \mathrm{~mL})$ at $-78{ }^{\circ} \mathrm{C}$, ammonia ( 46 mL ) was added. Small chips of lithium ( $280 \mathrm{mg}, 40 \mathrm{mmol}$ ) were added until the solution was a persistent deep blue for 1.5 h . The cooling bath was removed, the ammonia was allowed to evaporate overnight, and the reaction mixture was evaporated. The dried extract was dissolved in brine ( 15 mL ) and the product was extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ ( 3 X 15 mL ), dried with $\mathrm{Na}_{2} \mathrm{SO}_{4}$ and concentrated to give $6 \mathrm{a}(1.112 \mathrm{~g})$ as an oil which was used directly in the next step; ${ }^{1} \mathrm{H} \operatorname{NMR}\left(200 \mathrm{MHz}, \mathrm{C} \quad \mathrm{Cl}_{3}\right) 1.11(\mathrm{~d}, \mathrm{~J}=6.3 \mathrm{~Hz}, 3 \mathrm{H})$, 2.08 ( dd, J = 14.1, 9.0 Hz, 1H), 2.21 (dd, J = 13.5, $3.3 \mathrm{~Hz}, 1 \mathrm{H}$ ), 2.69-2.84 (m, 6 H ), 3.33-3.40(m, 1H), $3.55(\mathrm{~s}, 3 \mathrm{H}), 4.63(\mathrm{~m}, 1 \mathrm{H}), 5.51(\mathrm{~m}, 1 \mathrm{H}) ;{ }^{13} \mathrm{C}$ NMR ( 75 MHz , $\left.\mathrm{C} \mathrm{Cl}_{3}\right) 20.2\left(\mathrm{CH}_{3}\right), 29.1\left(\mathrm{CH}_{2}\right), 29.4\left(\mathrm{CH}_{2}\right), 41.6\left(\mathrm{CH}_{2}\right), 50.8(\mathrm{CH}), 53.7\left(\mathrm{CH}_{3}\right), 73.1$ (CH), 90.2 (CH), 120.2 (CH), 132.4 (C), 152.6 (C).
3.1.8. (2S,3S)-3-M ethoxy-4-(4-methoxy-2,5-dihydrophenyl)-2-butanamine (6b). Operating as above, starting from 5b ( $611 \mathrm{mg}, 2.92 \mathrm{mmol}$ ), 6b was obtained ( 620 mg ) as an oil which was used directly in the next step; ${ }^{1} \mathrm{H} N \mathrm{NR}\left(200 \mathrm{MHz}, \mathrm{C} \quad \mathrm{Cl}_{3}\right) 1.09$ ( d , $J=6.6 \mathrm{~Hz}, 3 \mathrm{H}), 2.14-2.29(\mathrm{~m}, 2 \mathrm{H}), 2.74-2.84(\mathrm{~m}, 3 \mathrm{H}), 2.87-2.96(\mathrm{~m}, 1 \mathrm{H}), 3.02-3.07$ $(\mathrm{m}, 1 \mathrm{H}), 3.40(\mathrm{~s}, 3 \mathrm{H}), 3.55(\mathrm{~s}, 3 \mathrm{H}), 4.62(\mathrm{~m}, 1 \mathrm{H}), 5.49(\mathrm{~m}, 1 \mathrm{H}) ;{ }^{13} \mathrm{CNMR}(50 \mathrm{MHz}$, $\left.\mathrm{C} \mathrm{Cl}_{3}\right) 20.1\left(\mathrm{CH}_{3}\right), 29.2\left(\mathrm{CH}_{2}\right), 30.0\left(\mathrm{CH}_{2}\right), 37.9\left(\mathrm{CH}_{2}\right), 49.3(\mathrm{CH}), 53.9\left(\mathrm{CH}_{3}\right), 58.4$ $\left(\mathrm{CH}_{3}\right), 84.7(\mathrm{CH}), 90.4(\mathrm{CH}), 120.1(\mathrm{CH}), 132.6(\mathrm{C}), 152.9(\mathrm{C})$.
3.1.9. Aminocyclization of 6 a . A solution of $6 \mathrm{a}(95 \mathrm{mg}, 0.48 \mathrm{mmol})$ in $2 \mathrm{~N} \mathrm{HCl}(1.6$ mL ) was stirred for 3.5 h at $70{ }^{\circ} \mathrm{C}$. The mixture was basified with $\mathrm{NaOH}(1 \mathrm{~N}, 10 \mathrm{~mL})$ and the solution was extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}(4 \times 10 \mathrm{~mL})$ and $\mathrm{CHCl}_{3} / \mathrm{MeOH}(4 \times 10$ mL ), dried, and concentrated to give a brown oil. urification by chromatography $\left(\mathrm{Al}_{2} \mathrm{O}_{3}, \mathrm{CH}_{2} \mathrm{Cl}_{2}\right.$ saturated with $\mathrm{NH}_{3}$ ) gave a partially separated 2.5:1 mixture of 7 a ( 37 $\mathrm{mg}, 43 \%$ ) and 8 a ( $15 \mathrm{mg}, 17 \%$ ).
(2S,3S,4aR,8aR)-3-H ydroxy-2-methyloctahydroquinolin-7-one (7a): white solid; mp 112-114 ${ }^{\circ} \mathrm{C} . \mathrm{R}_{\mathrm{f}}=0.17\left(\mathrm{Al}_{2} \mathrm{O}_{3}, 99: 1 \mathrm{CH}_{2} \mathrm{Cl}_{2}\right.$ saturated with $\left.\mathrm{NH}_{3} / \mathrm{M} \mathrm{eOH}\right) ;{ }^{1} \mathrm{H}$ NMR (400 $\mathrm{MHz}, \mathrm{C} \mathrm{Cl} 3, \mathrm{gCOSY}$ ) 1.11 ( $\mathrm{d}, \mathrm{J}=6.8 \mathrm{~Hz}, 3 \mathrm{H}, \mathrm{Me}$ ), 1.81 (ddd, J $=14.8,5.6,3.6 \mathrm{~Hz}$, $\mathrm{H}-4 \mathrm{eq}), 1.87(\mathrm{dm}, \mathrm{J}=14 \mathrm{~Hz}, \mathrm{H}-5 \mathrm{eq}), 1.95(\mathrm{dt}, \mathrm{J}=14.4,2 \mathrm{~Hz}, \mathrm{H}-4 \mathrm{ax}), 2.05(\mathrm{~m}, \mathrm{H}-4 \mathrm{a})$, 2.24 (dt, J = 14.4, $2 \mathrm{~Hz}, \mathrm{H}-6 \mathrm{eq}$ ), 2.29 ( $\mathrm{dd}, \mathrm{J}=14.4,5.6 \mathrm{~Hz}, \mathrm{H}-8 \mathrm{ax}$ ), 2.32 (m, H-6ax), 2.50 (qd, J = 13.6, 4.8 Hz, H-5ax), 2.65 (ddd, J = 14.8, 4.8, $0.8 \mathrm{~Hz}, \mathrm{H}-8 \mathrm{eq}$ ), 2.80 (qd, J
$=6.6,2 \mathrm{~Hz}, \mathrm{H}-2 \mathrm{ax}), 3.35$ (brs, H-8a), 3.58 ( $\mathrm{q}, \mathrm{J}=2.4 \mathrm{~Hz}, \mathrm{H}-3 \mathrm{eq}$ ); ${ }^{13} \mathrm{C}$ NMR ( 100 MHz , C Cl ${ }_{3}$, gHSQC) $18.1(\mathrm{Me}$ ), 28.8 (C-5), 33.5 (C-4a), 36.4 (C-4), 41.6 (C-6), 47.5 (C-8), 56.8 (C-8a), 59.3 (C-2), 68.2 (C-3), 210.8 (C-7). HRMS ( SI-TOF) calcd for $\mathrm{C}_{10} \mathrm{H}_{18} \mathrm{NO}_{2}\left(\mathrm{M}^{+}+1\right)$ 184.1332, found 184.1337.
(2S,3S,4aS,8aS)-3-H ydroxy-2-methyloctahydroquinolin-7-one (8a): Colourless oil. $R_{f}=0.14\left(\mathrm{Al}_{2} \mathrm{O}_{3}, 99: 1 \mathrm{CH}_{2} \mathrm{Cl}_{2}\right.$ saturated with $\left.\mathrm{NH}_{3} / \mathrm{MeOH}\right) ;{ }^{1} \mathrm{H} \mathrm{NMR}(300 \mathrm{MHz}$, $\mathrm{C}_{\mathrm{Cl}}^{3}$ ) $1.10(\mathrm{~d}, \mathrm{~J}=6.6 \mathrm{~Hz}, 3 \mathrm{H}, \mathrm{Me}), 1.72-1.98(\mathrm{~m}, 4 \mathrm{H}), 2.15-2.44(\mathrm{~m}, 4 \mathrm{H}), 2.93(\mathrm{t}, \mathrm{J}=$ 12.6, H-8ax), 3.06 (qd, J $=6.5,1,8 \mathrm{~Hz}, \mathrm{H}-2 \mathrm{ax}$ ), 3.39 ( $\mathrm{dt}, \mathrm{J}=11.7,4.8 \mathrm{~Hz}, \mathrm{H}-8 \mathrm{a}$ ), 3.76 (brs, H-3eq; ${ }^{13} \mathrm{C}$ NMR ( $\left.75 \mathrm{MHz}, \mathrm{C} \mathrm{Cl}_{3}, \quad \mathrm{~T}\right) 17.7$ ( Me ), 28.0 (C-5), 28.1 (C-4a), 31.9 (C-4), 36.6 (C-6), 42.6 (C-8), 47.5 (C-2), 55.9 (C-8a), 68.2 (C-3), 210.9 (C-7). HRMS ( $\mathrm{SI}-\mathrm{TOF}$ ) calcd for $\mathrm{C}_{10} \mathrm{H}_{18} \mathrm{NO}_{2}\left(\mathrm{M}^{+}+1\right)$ 184.1332, found 184.1331.
3.1.10. Aminocyclization of 6 b . Following the above procedure for the aminocyclization of 6a using methoxy derivative 6 b ( $225 \mathrm{mg}, 1.07 \mathrm{mmol}$ ), heating at $70{ }^{\circ} \mathrm{C}$ for 3 h , and purifying by chromatography $\left(\mathrm{Al}_{2} \mathrm{O}_{3}, 99: 1 \mathrm{CH}_{2} \mathrm{Cl}_{2}\right.$ saturated with $\mathrm{NH}_{3} / \mathrm{MeOH}$ ), a partially separated mixture of 7b ( $36 \mathrm{mg}, 17 \%$ ) and 8b (50 mg, 22\%) was obtained.
(2S,3S,4aR,8aR)-3-M ethoxy-2-methyldecahydroquinolin-7-one (7b): white solid, mp 45-47 ${ }^{\circ} \mathrm{C} ; \mathrm{R}_{\mathrm{f}}=0.25\left(\mathrm{Al}_{2} \mathrm{O}_{3}, 99: 1 \mathrm{CH}_{2} \mathrm{Cl}_{2}\right.$ saturated with $\left.\mathrm{NH}_{3} / \mathrm{M} \mathrm{eOH}\right) ;[\alpha]^{25}+14.6$ (c 0.7, $\mathrm{CHCl}_{3}$ ); ${ }^{1} \mathrm{H} \operatorname{NMR}\left(400 \mathrm{MHz}, \mathrm{C} \quad \mathrm{Cl}_{3}, \mathrm{gCOSY}\right) 1.12(\mathrm{~d}, \mathrm{~J}=6.8 \mathrm{~Hz}, 3 \mathrm{H}, \mathrm{Me})$, 1.62 (ddd, J = 14.8, 5.6, 3.2 Hz, H-4eq), 1.74 (m, H-5eq), 2.00 (dm, J = $12 \mathrm{~Hz}, \mathrm{H}-4 \mathrm{a}$ ), 2.13 ( dt, J = 14.8, 2.2 Hz, H-4ax), 2.23 (td, J = 14, 6 Hz, H-6ax), 2.26 (dm, J = 14.8 Hz, H-8), 2.32 (dddd, J = 14, 4.8, 2.4, 2.4 Hz, H-6eq), 2.61 (dd, J = 14.8, 5.6 Hz, H-8), 2.63 (qd, J = 14, 4.2 Hz, H-5ax), 2.78 (qd, J $=6.5,2.4 \mathrm{~Hz}, \mathrm{H}-2 a x$ ), $3.05(q, J=2.7 \mathrm{~Hz}$, $\mathrm{H}-3 \mathrm{eq}$ ), 3.30 (masked, $\mathrm{H}-8 \mathrm{a}$ ), 3.31 ( $\mathrm{s}, 3 \mathrm{H}, \mathrm{OMe}$ ); ${ }^{13} \mathrm{C} \mathrm{NMR}\left(100 \mathrm{MHz}, \mathrm{C} \mathrm{Cl}_{3}\right.$, gHSQC) 18.1 ( Me ), 26.9 (C-5), 30.9 (C-4), 33.4 (C-4a), 41.4 (C-6), 47.6 (C-8), 56.3 (C-2), 56.9 (OM e), 58.7 (C-8a), 76.9 (C-3), 210.6 (C-7). HRMS ( SI-TOF) calcd for $\mathrm{C}_{11} \mathrm{H}_{20} \mathrm{NO}_{2}\left(\mathrm{M}^{+}+1\right)$ 198.1489, found 198.1487.
(2S,3S,4aS,8aS)-3-M ethoxy-2-methyldecahydroquinolin-7-one (8b): colourless oil, $R_{f}=0.19\left(\mathrm{Al}_{2} \mathrm{O}_{3}, 99: 1 \mathrm{CH}_{2} \mathrm{Cl}_{2}\right.$ saturated with $\left.\mathrm{NH}_{3} / \mathrm{MeOH}\right) ;{ }^{1} \mathrm{H} \mathrm{NMR}(400 \mathrm{M} \mathrm{Hz}, \mathrm{C} \mathrm{Cl} 3$, gCOSY ) 1.11 ( $\mathrm{d}, \mathrm{J}=6.8 \mathrm{~Hz}, 3 \mathrm{H}, \mathrm{Me}$ ), 1.75-2.00 ( $\mathrm{m}, 4 \mathrm{H}, \mathrm{H}-4$ and $\mathrm{H}-5$ ), 2.20-2.30 (m, $3 \mathrm{H}, \mathrm{H}-4 \mathrm{a}, \mathrm{H}-6$ ), 2.39 (ddd, J = 14.4, 4.4, 1.5 Hz, H-8eq), 2.62 (dd, J = $14.4,10 \mathrm{~Hz}, \mathrm{H}-$ 8ax), 3.13 (qd, J = 6.5, 3.2 Hz, H-2ax), 3.34 (masked, H-3eq), 3.36 (s, 3H, OM e), 3.43 (ddd, J $=10,4.8,4.8 \mathrm{~Hz}, \mathrm{H}-8 \mathrm{a}) ;{ }^{13} \mathrm{CNMR}(100 \mathrm{MHz}, \mathrm{C} \mathrm{Cl} 3, \mathrm{gHSQC}) 16.0(\mathrm{Me}$ ), 27.6
(C-5), 27.8 (C-4), 29.6 (C-4a), 37.7 (C-6), 43.7 (C-8), 48.1 (C-2), 53.3 (C-8a), 56.7 (OM e), 76.4 (C-3), 210.9 (C-7). HRMS ( SI-TOF) calcd for $\mathrm{C}_{11} \mathrm{H}_{20} \mathrm{NO}_{2}\left(\mathrm{M}^{+}+1\right)$ 198.1489, found 198.1487.
3.1.11. (2S,3S,4aR,8aR)-1-Benzoyl-3-hydroxy-2-methyloctahydroquinolin-7-one (9a). A solution of $7 \mathrm{a}(12 \mathrm{mg}, 0.07 \mathrm{mmol})$ was dissolved in THF ( 0.2 mL ) and $\mathrm{H}_{2} \mathrm{O}$ $(0.2 \mathrm{~mL})$ was added. Then, $\mathrm{K}_{2} \mathrm{CO}_{3}(39 \mathrm{mg}, 0.28 \mathrm{mmol})$ and $\mathrm{BzCl}(8.4 \mu \mathrm{~L}, 0.074 \mathrm{mmol})$ were added. The reaction mixture was stirred for 2 h at rt , extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}(4 \mathrm{x}$ 15 mL ), dried, and concentrated to give a brown oil. urification by column chromatography $\left(\mathrm{Al}_{2} \mathrm{O}_{3}\right.$, from $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ saturated with $\mathrm{NH}_{3}$ to $98: 2 \mathrm{CH}_{2} \mathrm{Cl}_{2}$ saturated with $\mathrm{NH}_{3} / \mathrm{MeOH}$ ) gave 9a ( $19 \mathrm{mg}, 99 \%$ ): $\mathrm{R}_{\mathrm{f}}=0.44\left(\mathrm{Al}_{2} \mathrm{O}_{3}, 98: 2 \mathrm{CH}_{2} \mathrm{Cl}_{2}\right.$ saturated with $\mathrm{NH}_{3}$ ) ; ${ }^{1} \mathrm{H}$ NMR ( $300 \mathrm{MHz}, \mathrm{C} \quad \mathrm{Cl}_{3}$, mixture of rotamers) 1.20 and 1.30 ( 2 brd, $\mathrm{CH}_{3}$ ), 1.70-2.20 (m, 6H), $2.34(b r, 1 H), 2.75(\mathrm{~m}, 1 \mathrm{H}), 3.85-4.15(\mathrm{br}, 2 \mathrm{H}), 5.07$ (br, 1H), 7.25$7.45(\mathrm{~m}, 5 \mathrm{H}, \mathrm{ArH}) ;{ }^{13} \mathrm{C} \operatorname{NMR}\left(75 \mathrm{MHz}, \mathrm{C} \quad \mathrm{Cl}_{3}\right) 14.1$ and $15.7\left(\mathrm{CH}_{3}\right), 27.5$ and 28.6 (C.4), 29.7 (C-5), 31.9 (C-4a), 36.2 (C-6), 49.1 (C-8), 53.4 (C-2), 55.3 (C-8a), 74.6 (C3), 125.9, 128.8, 129.5, 136.5 (Ar), 171.6 and 172.2 (NCO), 208.0 (C-7). HRM S ( SITOF) calcd for $\mathrm{C}_{17} \mathrm{H}_{22} \mathrm{NO}_{3}\left(\mathrm{M}^{+}+1\right) 288.1594$, found 288.1585.
3.1.12. (2S,3S,4aR,8aR)-1-Benzoyl-3-methoxy-2-methyloctahydroquinolin-7-one (9b). Operating as above, starting from 7 b ( $16 \mathrm{mg}, 0.08 \mathrm{mmol}$ ) and after purification by chromatography $\left(\mathrm{Al}_{2} \mathrm{O}_{3}, \mathrm{CH}_{2} \mathrm{Cl}_{2}\right.$ saturated with $\left.\mathrm{NH}_{3}\right)$, amide 9b ( 24 mg , 99\%) was obtained as a white solid: mp 100-102 ${ }^{\circ} \mathrm{C}$; $\mathrm{R}_{\mathrm{f}}=0.52\left(\mathrm{Al}_{2} \mathrm{O}_{3}, \mathrm{CH}_{2} \mathrm{Cl}_{2}\right.$ saturated with $\mathrm{NH}_{3}$ ); ${ }^{1} \mathrm{H}$ NMR ( $300 \mathrm{MHz}, \mathrm{C} \quad \mathrm{Cl}_{3}$, mixture of rotamers) 1.05 and 1.25 ( $2 \mathrm{brd}, \mathrm{CH}_{3}$ ), $1.70-2.20(\mathrm{~m}, 6 \mathrm{H}), 2.35(\mathrm{br}, 1 \mathrm{H}), 2.75(\mathrm{~m}, 1 \mathrm{H}), 3.20$ and $3.40\left(2 \mathrm{~s}, 3 \mathrm{H}, \mathrm{OCH}_{3}\right), 3.25-$ 3.45 (masked, 2 H ), $3.95(\mathrm{br}, 0.5 \mathrm{H}), 4.15(\mathrm{br}, 0.5 \mathrm{H}), 5.0(\mathrm{br}, 0.5 \mathrm{H}), 5.20(\mathrm{br}, 0.5 \mathrm{H})$, 7.20-7.65 (m, 4H, ArH), $8.20(d, J=7.5 \mathrm{~Hz}, 1 \mathrm{H}, \mathrm{ArH}) ;{ }^{13} \mathrm{CNMR}(75 \mathrm{MHz}, \mathrm{C} \mathrm{Cl} 3)$ 15.0 and $16.0\left(\mathrm{CH}_{3}\right), 25.3$ and $25.7(\mathrm{C}-4), 27.6(\mathrm{C}-5), 32.6$ and $33.5(\mathrm{C}-4 \mathrm{a}), 36.0$ and 36.4 (C-6), 43.6 and 45.2 (C-8), 45.4, 49.4, 51.2, and 56.2 (C-2 and C-8a), 55.6 and $56.6\left(\mathrm{OCH}_{3}\right), 77.9$ and $78.4(\mathrm{C}-3), 125.7,128.7,129.4,136.6$ (Ar), 171.6 (NCO), 207.4 and 207.9 (C-7). HRMS ( $\mathrm{SI}-\mathrm{TOF}$ ) calcd for $\mathrm{C}_{18} \mathrm{H}_{24} \mathrm{NO}_{3}\left(\mathrm{M}^{+}+1\right) 302.1751$, found 302.1752.
3.1.13. $(2 \mathrm{~S}, 3 \mathrm{~S}, 4 \mathrm{aR}, 8 \mathrm{aR})$ - and $(2 \mathrm{~S}, 3 \mathrm{~S}, 4 \mathrm{aS}, 8 \mathrm{aS})-1$-B enzoyl-3-methoxy-2-methyloctahydro-quinolin-7-one (9b and 10b). A solution of 6 b ( $90 \mathrm{mg}, 0.42 \mathrm{mmol}$ )
in $\mathrm{HCl} 2 \mathrm{~N}(2 \mathrm{~mL})$ was stirred for 3 h at $75{ }^{\circ} \mathrm{C}$. The mixture was basified with $\mathrm{K}_{2} \mathrm{CO}_{3}$ ( $464 \mathrm{mg}, 3.36 \mathrm{mmol}$ ) and $\mathrm{BzCl}(0.06 \mathrm{ml}, 0.5 \mathrm{mmol})$ in THF ( 2 mL ) was added. The reaction mixture was stirred for 2 h at rt, concentrated and extracted with $\mathrm{CH}_{2} \mathrm{Cl}_{2}(4 \mathrm{x}$ 20 mL ). The dried organic layers were concentrated to give a brown oil, which was purified by chromatography $\left(\mathrm{SiO}_{2}\right.$, from Hexane/AcO t 7:3 to AcO t) to give a 1:2.3 mixture of $9 \mathrm{~b}(22 \mathrm{mg}, 17 \%$ from 4b) and 10b ( $50 \mathrm{mg}, 39 \%$ from 4b). For data of 9a, see above.

Compound 10b: Colourless oil. $\mathrm{R}_{\mathrm{f}}=0.14\left(\mathrm{SiO}_{2}, 1: 1 \mathrm{Hexane} / \mathrm{AcO} \mathrm{t}\right) ;[\alpha]^{25}+16(\mathrm{c} 0.9$, $\mathrm{CHCl}_{3}$ ); ${ }^{1} \mathrm{H}$ NMR ( $400 \mathrm{M} \mathrm{Hz}, \mathrm{C} \quad \mathrm{Cl}_{3}, \mathrm{gCOSY}$ ) $1.25(\mathrm{~d}, \mathrm{~J}=6.8 \mathrm{~Hz}, 3 \mathrm{H}, \mathrm{Me}), 1.80(\mathrm{~m}$, H-5), 1.87 (m, H-4), 1.99 (dt, J = 14, 5.2 Hz, H-4eq), 2.11 (dddd, J = 12, 11, 9.4, 4.4 Hz, H-5ax), 2.25 (ddd, J = 15.4, 10, 5.4 Hz, H-6ax), 2.36 (m, H-4a), 2.64 (masked, 1 H , $\mathrm{H}-6), 2.59$ and 2.67 ( $2 \mathrm{dd}, \mathrm{J}=16.8,5.6 \mathrm{~Hz}, 1 \mathrm{H}$ each, $\mathrm{H}-8$ ), 3.20 (s, $3 \mathrm{H}, \mathrm{OM}$ e), 3.51 (ddd, J = 10.8, 5.4, 5.4 Hz, H-3ax), 4.12 (ddd, J = 5.6, 5.6, $2.8 \mathrm{~Hz}, \mathrm{H}-8 \mathrm{a}$ ), 4.24 (quint, J $=6.4 \mathrm{~Hz}, \mathrm{H}-2 \mathrm{eq}), 7.40(\mathrm{~s}, 5 \mathrm{H}, \mathrm{ArH}) ;{ }^{13} \mathrm{C}$ NMR ( $\left.100 \mathrm{M} \mathrm{Hz}, \mathrm{C} \mathrm{Cl} 3, \mathrm{gHSQC}\right) 12.5(\mathrm{Me})$, 26.5 (C-5), 28.5 (C-4), 33.3 (C-4a), 38.5 (C-6), 43.0 (C-8), 51.9 (C-8a), 52.8 (C-2), 56.2 (OM e), 75.0 (C-3), 126.8, 128.6, 130.0, 136.6 (Ar), 173.2 (NCO), 205.4 (C-7). HRMS ( SI-TOF) calcd for $\mathrm{C}_{18} \mathrm{H}_{24} \mathrm{NO}_{3}\left(\mathrm{M}^{+}+1\right) 302.1751$, found 302.1750.

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 Hom: Marisa mian Cirvicon puta:23/11/05/sist sequence: e2pu1




















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[^0]:    Keywords: Lepadin alkaloids; Decahydroquinolines; Epoxides; Organolithiums; Nitrogen heterocycles.

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